نسنى دراسة لغسع

Ministry Of Higher Education

University Of Technology Chemical Engineering Department



B.Sc PROGRAMME IN CHEMICAL ENGINEERING

OUTLINE OF SYLLABUSES ALLOCATION OF SUBJECTS AND WEEKLY LOAD

2003 - 2004

INTRODUCTION

Chemical Engineering is distinguished from other branches of engineering by its strong dependence on chemistry. This enables the chemical engineer to understand properties involving changes in physical state, chemical composition, or energy content for systems ranging in scale from molecules to full sized manufacturing plants.

The chemical engineer may be employed in an established industry producing chemicals, petroleum products, petrochemicals, pharmaceuticals, synthetic fibers, foods, plastics or metals. These products are steadily needed in an increasing amounts due to the expanding growth of population. His function may be involved in making innovations in operations, doing research, development and analysis of existing plants, technical services, or scale up.

Due to his board knowledge, the chemical engineer often occupies a dominant position in the above mentioned industries.

He also apply his knowledge in such diverse areas as air and water pollution or biochemical research.

The formal course work for B.Sc. programme involves mathematics, including both analysis and computer application, chemistry, mechanics, electricity chemical engineering principles and practice subjects including fluid dynamics, heat, unit operation, chemical reactor design, thermodynamics, and chemical processes. In addition the programme involves English Language which is essential for any engineering student to enable him/her to benefit from thousands of textbooks and periodicals.

A considerable emphasis is laid, in this department, on practical training and laboratory work, which helps students in getting better understanding of the theoretical part of their curriculum.

Moreover students are required to spent twelve weeks, training in industry during summer vacations.

This booklet contains the academic programme and the syllabus for subjects which are taken by the chemical engineering students during their four years study.

Prof. M. A. Zablouk Chairman of Chemical Eng. Dept.

Contents

B.Sc. Programme

·•	Page								
• First Year Syllabus									
- English	10								
 Mathematics Principle of Chemical Engineering Chemistry 									
								- Engineering Drawing	16
								- Mechanics and Strength of Materials	17
- Electrical Technology	19								
- Computer Science	20								
- Workshop	22								
Second Year Syllabus	22								
- English	23								
- Mathematics	25 26								
Fluid FlowProperties of Materials									
							- Computer Programming	29	
- Fuels Technology	31								
- Thermodynamics	32								
- Physical Chemistry	34								
• Third Year Syllabus									
- Applied Mathematics	35								
- Mass Transfer	36								
- Statistics and Instruments Measurement	37								
- Reactor Design	38								
- Heat Transfer	39								
- Equipment Design by Computer	40								
- Biochemical Engineering	42								
- Environmental Engineering	43								
- Storage and Transport of Crude and Petroleum Products	44								
- Combustion Engineering									

	Page
• Fourth Year Syllabus	
- Project	46
- Transport Phenomena	47
- Process Control	49
- Chemical Industries	50
- Optimization	52
- Engineering of Catalysts	53
- Corrosion Engineering	55
- Technology of Oil and Gas	56
- Management and Engineering Economic	57
- Polymer Technology	58
- Petroleum Refining	59
- Gas Technology	60
- Petrochemical Industries	61
- Environmental - Pollution	62

Subject	Subject	Hours / Week						
Symbol		F	irst Term		Se	econd Term		Units
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 111	English Language	2	-	-	2	-	-	4
CE. 121	Mathematics	3	-	1	3	-	1	6
CE. 141	Principle of Chemical Engineering	3	-	1	3	-	1	6
CE. 221	Chemistry	3	2	1	3	2	1	8
CE. 131	Engineering Drawing	1	2	-	1	2	-	4
CE. 231	Mechanics and Strength of Materials	2	-	1	2	-	1	4
CE. 331	Electrical Technology	1		1	1	-	1	2
CE. 321	Computer Science	2	2	-	2	2	-	6
CE. 431	Workshop	-	6	-	-	6	-	4
	Total	17	12	5	17	12	5	44

Riest Term	- Second Term
34	34

Subject	Subject	Hours / Week						
Symbol		F	irst Term		Sec		Units	
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 112	English Language	2	-	-	2	_	-	4
CE. 122	Mathematics	2	-	1	2	-	1	4
CE. 142	Fluid Flow	3	2	1	3	2	1	8
CE. 132	Properties of Materials	1	-	1	1	2	1	3
CE. 222	Computer Programing	2	2	-	2	2	-	6
CE. 342	Fuels Technology	1	2	1	1	-	1	3
CE. 242	Thermo dynamics	2	_	1	2	-	-1	4
CE. 232	Physical Chemistry	2	2	1	2	2	1	6
	Total	15	8	6	15	- 8	6	38

Pirst Term	Second Term
29	29

A-Unit Operation OF Chemical Industry Branch

Subject Symbol	Subject	Hours / Week						
·		F	irst Term		Se	cond Term		Units
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 133	Applied Mathematics	3	-	1	3	-	1	6
CE. 143	Mass Transfer	3	2	1	3	-	1	7
CE. 233	Statistics and Instruments Measurement	1	1	1	1	-	1	3
CE. 243	Reactor Design	3	_	1	3	-	1	6
CE. 343	Heat Transfer	3	-	1	3	2	1	7
CE. 443	Equipment Design by Computer	2	1	-	2	1	-	6
CE. 543	Biochemical Engineering	2	~	1	-	-	_	2
CE. 643	Environmental Engineering	-	-	-	2	-	1	2
	Total	17	4	6	17	3	6	39

Cirst Ferm	Second Term
27	26

B-Refinery Engineering of Oil and Gas Branch

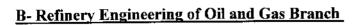
Subject Symbol	Subject	Hours / Week						
Symbol		F	irst Term		Se	cond Term		Units
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 133	Applied Mathematics	3	_	1	3	-	1	6
CE. 143	Mass Transfer	3	2	1	3	-	1	7
CE. 233	Statistics and Instruments	1	1	1	1	-	1	3
CE. 243	Reactor Design	3	-	1	3	-	1	6
CE. 343	Heat Transfer	3	-	1	3	2	1	7
CE. 443	Equipment Design by Computer	2	1	-	2	1	-	6
CE. 543	Storage and Transport of Crude and Petroleum Products	2	-	1	-	-	. =	2
CE. 643	Combustion Engineering	-	-	-	2	-	1	2
	Total	17	4	6	17	3	6	39

kirst Term	Second Parm
27	26

A-Unit Operation of Chemical Industry Branch

Subject Symbol	Subject	Hours / Week						
Бушоог			irst Term		Se	Units		
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 144	Project	1	2	-	1	2	-	4
CE. 244	Transport Phenomena	3	3	1	3	-	1	7
CE. 344	Process Control	2	-	1	2	3	1	5
CE. 444	Chemical Industries	2	3	1	2	3	1	6
CE. 234	Optimization	-	-	-	2	1	-	3
CE. 644	Engineering of Catalyst	2	-	1	-	-	· _	2
CE. 844	Corrosion Engineering	-	-	-	2	-	1	2
CE. 544	Technology of Oil and Gas	-	-	-	2	-	1	2
CE. 134	Management and Engineering Economics	-		.=	2	-	1	2
CE. 744	Polymers Technology	2	-	1	-	 _		2
	Total	14	8	6	14	9	5	35

First Term	Second Term
28	28



Subject Symbol	Subject	Hours / Week						
		F	irst Term		Se	cond Term		Units
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 144	Project	1	2	-	1	2	-	4
CE. 244	Transport Phenomena	3	3	1	3	-	1	7
CE. 344	Process Control	2	-	1	2	3	1	5
CE. 444	Petroleum Refining	2	3	1	2	3	1	6
CE. 544	Gas Technology	-	_	-	2	-	1	2
CE. 134	Management and Eingineering Economics	-	-	-	2	-	1	2
CE. 234	Optimization	-	-	-	2	1	-	3
CE. 644	Catalyst Engineering	2	-	1	_	-	-	2
CE. 744	Petrochemical Industries	2	-	1	-	-	-	2
CE. 844	Environmental Pollution	2	-	1	-	- '	_	2
	Total	14	8	6	14	9	5	35

Kirst Lerm	Second Term
28	28

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 111	<u>English</u>	Units 4 Theoretical 2 hr/week Tutorial - hr/week Practical - hr/week
First Semester		
1- Comprehension		
Reading Scientific Pa	ssages	
A- Solids, liquids	and Gases	
B- The Control of	Electric Currents	
C-Sir Isaac Newto	n	
		(18hrs)
2- Grammatical Points		
A-The Sentence		
1- Affirmative		
2- Interrogative		
3- Negative		
_		(4hrs)
B-Parts of Speech		
1- Nouns		
A-Function		
B-Countibility		
C-non-Countibi	lity	
	•	(1hrs)
2-Pronouns:		
Types and function	n	
V 1		(lhrs)
3-Adjectives		•
A-Degree/Compar	ison	
B- Order		
	*	(1hrs)
4-Verbs		
A-Finite		
B-Non-Finite		
C-Transitive		
D-Intransitive		
		(lhrs)
5-Adverbs		
A-Classification		
B-Formation		
C-Order		
		(1hrs)
		` ,

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 111	<u>English</u>	Units 4 Theoretical 2 hr/week Tutorial - hr/week Practical - hr/week
6-Prepositions: A general idea		/1L\
7-Conjunctions A-Main Conjunction B-Subsidiary	ıs	(1hrs)
8-Interjections:		(1hrs)
Common ones		(1hrs)
Second Semester		
1- Comprehension Reading Scientific Pas 1-Electricity in Earl 2-Comets 3-Exploring the mo	ly Days	(16hea)
2- Grammatical Points		(16hrs)
1-Articles A-definite B-indefinite		(2h)
2-The Sentence A-Structure B-Types C-Main Parts:		(2hrs)
	ibject edicate	
D-Expansion		(Qhua)
3-Written and Oral Tra	nslation	(8hrs)
		(4hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
		Units 6
CE 121	Mathematics	Theoretical 3 hr/week
02 121		Tutorial 1 hr/week
		Practical - hr/week

1- Revision:

Equation of the straight line, Trigonometric functions and their sketches, Domain, Range, Inverse of functions, Absolute value, limits, Limits, applications, Polar coordinates (general definition) Conic sections (general definition).

(12 hrs)

2- Differentiation and Integration:

Algebraic functions and trigonometric functions, Application (limits, Hopitals rule)
(6 hrs)

3- Transcendental Functions:

Inverse of Trigonometric functions, Natural logarithmic Exponential and Power functions (Definitions, Properties, Sketches, Derivatives and Integrals.

(12 hrs)

4- Hyperbolic Functions:

Definitions, Properties, Sketches, inverse and Derivatives and Integrals.

(8 hrs)

5- Determinants:

Definitions, Properties, Solution of systems of equations (Cramers Rule)

(6 hrs)

6- Application of Definite Integrals:

Area between two curves, Volumes, Length of curves, Surface area.

(12 hrs)

7- Moods of integration:

Substitution, Integration by parts, The Substitution $U = \tan \frac{X}{2}$, Power of trigonometric functions, Integrals involve $\sqrt{a^2 - x^2}$, $\sqrt{a^2 + x^2}$, $\sqrt{x^2 - a^2}$, integrals of $ax^2 + bx + c$, Partial functions, Improper Integrals (conv and dity)

(16 hrs)

8- Complex Numbers:

Definitions, Argands diagram, Addition and subtraction and multiplication and dividing numbers, Trigonometric and exponential formula of complex number, Demoivors theorem, Roots of equation, Power of complex numbers.

(9 hrs)

9- Vector Analysis:

Definition, Addition and subtraction of vectors, Multiplication by scalars, Direction, products of vector (scalar product and vector product).

(9 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. S	<u>Syllabus</u>
CE 141	Principle of Chemical Engineering	Units Theoretical Tutorial Practical	6 3 hr/week 1 hr/week - hr/week

1- Concept of Chemical Engineering:

Units, Dimensions, Conversion factors, Conversion equation, Temperature, Pressure, Composition, Chemical Analysis, Chemical equations.

(24 hrs)

2- Material Balance Calculation:

Without and with chemical reaction, Recycle, By pass and purge calculations for steady state, Combustion Calculation.

(30 hrs)

3- Ideal Gas laws:

Definitions of Ideal gas, Real gas, Ideal gases mixture, Real gas mixture.

(9 hrs)

4- Energy Balance:

General energy balance, Enthalpy, Heat capacities their predications and variation with temperature, Heat effects.

(18 hrs)

5- Simultaneous, Mass and Energy Balance:

(9 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
OE 221	Chemistry	Units 8 Theoretical 3 hr/week
CE 221	Chemistry	Tutorial 1 hr/week Practical 2 hr/week

Part -1- Analytical Chemistry

1- Introduction:

Atomic weight, Molecular formula, Chemical equations, Mole concept, and Chemical equilibrium.

(4 hrs)

2- Solution:

Definition, Preparation and properties, Molarity, Normality, Formality, PH, POH, Solubility

(8 hrs)

3- Analytical Methods of Analysis:

a-Qualitative Analysis

b-Quantitative Analysis.

- Volumetric (titrimetric) & analysis,
 Acid- base, Redox, Precipitation, Complex titration, Methods of calculation,
 Titration curves
- ii. Gravimetric AnalysisPrecipitation reactions, Direct and indirect methods of analysis, Ksp.
- iii. instrumental Methods of Analysis
 Photometric,, Colourimetic, Potentiometric titration,, PH-measurements
 Chromatography, Atomic absorption.

(33 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
OF 221		Units 8 Theoretical 3 hr/week
CE 221	<u>Chemistry</u>	Tutorial 1 hr/week Practical 2 hr/week

Part -2- Organic Chemistry

1- Introduction:

Definition and classification of organic compounds.

(3 hrs)

2- Aliphatic Compounds:

Nomenclature, Preparation, Properties, Reactions.

- a) Alkanes
- b) Alkenes
- c) Alkynes
- d) Aliphatic Derivatives: Alkyl halides Alcohol's, Ethers, Aldehydes, Ketones, Esters Carboxylic acids, Amines, Mercaptans, Carboxylic acids, Amines, Mercaptans, Thiophenes, Disulphide.

(20 hrs)

3- Aromatic Compounds:

Structural formula of benzene ring, Nomenclature, Preparations, Properties, Chemical reactions; Nitration, Sulphonation, Halogenation, Alkylation, Benzene-homologues; Preparation, properties, chemical reactions of — Toluene, Xylene, Ethyl benzene, Styrene, Phenols, Anilines, Naphthalene.

(10 hrs)

4- Heterocylic Compounds:

Pyridine, Furan, Pyrrole.

(3 hrs)

5- Introduction to Polymer and Carbohydrate:

(5 hrs)

6- Organo-Metallic Compounds:

(4 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 131	Engineering Drawing	Units 4 Theoretical 1 hr/week Tutorial - hr/week Practical 2 hr/week
1- Introduction:		
	bout engineering drawing and used in drawings, Hidden lines.	its importance, Drawing
• •		(2 hrs)
2- Geometrical construction	on:	(2 hrs)
3- <u>Pictorial Drawings</u> : (Isometric and Oblique) 4- <u>Projection Drawings:</u> First and Third Angle Pr	ojection	(2 hrs)
First and Third Angle Pro	ojection.	(4 hrs)
5- The Finding of 3 rd vie	<u>w:</u>	(6 hrs)
6- Sectioning:		
7 Fastanaus		(6 hrs)
7- Fasteners:		(2 hrs)
8- Assembly and detail dr	awing:	(=)

(4 hrs)

Universit	ty of Technology	Chemical Engineering Department	First Year B.Sc. S	<u>yllabus</u>
CE 231	<u>M</u>	echanics and Strength of Materials	Theoretical Tutorial	4 2 hr/week 1 hr/week - hr/week
A-Mech	nanies			
1- Princip	les of static's:			(2 hrs)
2- Resulta	nts of Force sys	tems:		(2 hrs)
3- Fauilib	rium of force sy	rstams•		(5 hrs)
		stems.		(5 hrs)
4- Friction	<u>1:</u>			(6 hrs)
5- Centuri	ions and Center	s of Gravity:		(5 hrs)
6- Momen	t of Inertia:			(5 hrs)
7- Princip	le of Movement	•		(4 hrs)
· 111111				(3 hrs)
D C				
B-Stren	gth of Materials	<u>s</u>		
1- Interna	l Forces in Rigio	d Bodies:		
2 Dofiniti	an of stress and	straine		(1 hrs)
	on of stress and f stresses and stra			
	Strain Diagrams			(1 hrs)
20,622- 2	aram diagrams ic	or ductile and brittle materials		(2 hrs)
Elastic 1		asticity, plasticity, Toughness, Resili		•
Working	g stress.			(2 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 231 <u>M</u>	lechanics and Strength of Materials	Units 4 Theoretical 2 hr/week Tutorial 1 hr/week Practical - hr/week
5- Hook law:		(2 hrs)
6- Poisson's ratio, Compo		(4 hrs)
8- Thin walled Cylinders:	<u>.</u>	(4 hrs) (4 hrs)
9- Thermal stress: 10- shear and Moments in	Rooms	(4 hrs)
Deflection		(4 hrs)
11- Wind Effect on the To	wer high Tower:	(2 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 331	Electrical Technology	Units 2 Theoretical 1 hr/week Tutorial 1 hr/week Practical - hr/week
1- Semiconductors Equips 2. Destifiers and Detector		(3 hrs)
2- Rectifiers and Detector	<u>s:</u>	(2 hrs)
3- Electronic Amplifiers:		(4 hrs)
4- D.C. Circuits:		(2 hrs)
5- D.C. Generators and M	loto <u>rs:</u>	(3 hrs)
6- A.C Circuits:		(3 hrs)
7- Polyphases Circuits:		(2 hrs)
8- Transformers and Indu	ection Motors:	(3 hrs)
9- <u>Starters</u> :		(2 hrs)
10- Integrated Circuits, Mo	easuring Instruments, Tranducer	s, Transmitter:
11- Electrical Heating App	liance:	(4 hrs)
	•	(2 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Synabus
CE 321	Computer Science	Units 6 Theoretical 2 hr/week Tutorial - hr/week Practical 2 hr/week

1- Introduction:

Component (hard ware), computer work, computer, composition, Application

(2 hrs)

2- Files:

Definition, Types, Nomenclature, MSDos, External and internal commends.

(4 hrs)

3- Intoduction to window operating system:

Desktop, Mouse, my computer-icons, close window, stand by.

(2 hrs)

4- Folders:

Size and cascade, windows folder construction, construction choose file or folder find file or folder copy file or folder.

(2 hrs)

5- Start up:

Start up, Print, shutdown windows, format & Test Floppy disk, Arrange Icons, Run, Help.

(2 hrs)

6- Word Program:

File commands (New, Open, Close, Save, Save as, Page setup, Edit, Undo typing, Repeat Typing, Cut, Copy, Paste, Clear, Select All, Find, Replace, Goto).

(2 hrs)

7- Menu Table:

View (zoom, Header and footnote, Tool bar Insert (Picture, Footnote, Symbol, Page numbers, Textbox, Object).

Format (Paragraph, Font, Borders, Bullets & numbering Spelling and grammar Table (shading, Columns, Insert, Table Insert Rows, Language, Delete, Cells, Split Cells, Select Row, Select column) (Table Autoformat Sort, Formula).

(6 hrs)

8- Program Excel:

Open, Taskbar, Toolbar, Format, Sheet, Cell, Construction of sheet, Open file, Clear (Delete), Save (Save as), Close, cascade, Copy and Paste, Add, Delete, Find, Replace, Auto summation, Insert, equation, Format sheet, Auto format, Print, Print preview, Draw.

(12 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 321	Computer Science	Units 6 Theoretical 2 hr/week Tutorial - hr/week Practical 2 hr/week

9- Power Point:

Definition

FILE: New, Open, Open, Close, Save, Save as, Page set up, Print, Exit)

EDIT: Undo, Repeat, Cut, Copy, Paste, Select All, clear Duplicate, Delete slide,

find, Replace

FORMAT: Font, Alignment, Text Direction slide, Layout, Background, Apply Slid Show, View show, Rehearse timing design set up show, present Animation, Animation Preview, Slide Transition).

(11 hrs)

10- Auto Cad:

Introduction, define screen draw, select point in the screen Limits, Status, Circle, Erase, Line, Color, Point, zoom, Redraw, Arc, Ellipse, Polygon, Donut, Solid, Fill Break, Copy, More, Pan, Mirror Trim, Extend, Charge, Line type Ltscale, Dis Area, Trace, Hatch text, Snap, Grid, Rotate, Cops, Ortho, Fillet, Dim Layer.

(14 hrs)

11-Internet:

(6 hrs)

 	 			 <u>-</u>	 <u> </u>	

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 431	<u>Workshop</u>	Units 6 Theoretical - hr/week Tutorial - hr/week Practical 6 hr/week

Refer to Centralize Curriculum

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 122	<u>English</u>	Units 4 Theoretical 2 hr/week Tutorial - hr/week Practical - hr/week
First Semester		
1- Comprehension Reading Scientific Pa	•	
1-Metals and Ores		
2-Heat and its Eff		
3- Mending Faces		(101)
2- Grammatical Points	5	(18hrs)
1-Verb Tenses (Simp	le and Continuous)	
A-Present	and commutation	
B-Past		
C-Future		
		(5hrs)
2-Active and Passive	Voice	` ,
A-importance	in scientific studies	
B-the passive	voice sentence	
		(5hrs)
3-Scientific Report w	riting	
		(3hrs)

Second Semester

1-Comprehension

Reading Scientific Passages

- 1-Electromagnetic
- 2-Light and Heat on the Moon
- 3-Time and the Stars

(16hrs)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
	<u> </u>	Units 4
CE 122	<u>English</u>	Theoretical 2 hr/week Tutorial - hr/week Practical - hr/week
2- Grammatical Po	i <u>nts</u>	
	fect and perfect Continuous)	
A-Present	•	
B-Past		
C-Future		
		(5hrs)
2-Active and Passiv	ve Voice	
A- Construction	1	
B- fields of app	lication	
	•	(5hrs)
3-Relative Clauses		

(3hrs)

	Uni	ts	4
CE 122	<u>Mathematics</u> The	oretical	2 hr/week
	Tut	orial	1 hr/week

Chemical Engineering Department

1. Partial Differentiation:

University of Technology

Function of two or more variables, limits and continuity, partial derivatives, chain rule, Gradients, Directional derivatives, Higher order derivatives, Maxima, minimum and saddle points, Lagrange multipliers.

(10hrs)

Practical - hr/week

Second Year B.Sc. Syllabus

2. Multiple integrals:

Double integrals, Area, Triple integrals in rectangular coordinates, Physical application of double and triple integration.

(9hrs)

3. Ordinary Differential Equations:

Solution of first order ordinary differential equations, Solution of second order ordinary differential equations, and higher order differential equation.

(9hrs)

4. Infinite Series:

Power series of functions, Taylor's theorem, Integration, Differentiation, Fouries series, Even and odd functions, Half range expansion, Periodic functions.

(8hrs)

5. Function and Definite Integrals:

The error function, The gamma function, The beta function, factorial function.

(6hrs)

6. Vector Analysis:

Products of three vectors, Equation of lines and planes.

(5hrs)

7. Complex Algebra:

Continuity, Derivation of Complex Variables, Analytic function, Integration of functions of complex variables and cauchy's theorem.

(5hrs)

8. Matrices:

The Matrix, Matrix algebra, The transpose of matrix, The inverse of matrix, Eigen Values and Eigen vectors, solution of linear equation by matrix.

(8hrs)

University of Technology	Chemical Engineering Department	Second Y	Year B.Sc. S	Syllabus
			Units	8
CE 142	<u>Fluid Flow</u>		Theoretical	3 hr/week
			Tutorial	1 hr/week

1. Introduction:

Physical properties of fluid, Definition of type fluid: Newtonian's, non-Newtonian, Incompressible, compressible fluid, static, Dynamic fluid, shear stresses in fluid.

(4hrs)

Practical __2 hr/week

2. Dimensional- Analysis:

Rayleigh's method (Power series), Buckingham's П-theorem.

(6hrs)

3. Fluid- Dynamic:

Fluid dynamic system, Boundary layer, Continuity equation, Momentum equation, Bernoulli's equation, Euler's equation, Application of Bernoulli's equation and continuity equation.

(10hrs)

4. Newtonian's Fluid-(Incompressible Fluid):

Laminar and Turbulent flow, Reynolds's number, Friction losses and Pressure drop in pipes and Fitting, Determine of pipe diameter, Flow in Branch pipes, Friction losses in non-circular pipes, velocity Distribution.

(10hrs)

5. Non- Newtonian Fluid:

Definition, Type of fluid depend on time, Calculation of friction and pressure drop for general time independent in laminar and Turbulent flow.

(6hrs)

6. Compressible Fluid in pipes:

Pressure wave, Mach number, General equation, General equation for isothermal and adiabatic condition, work for isothermal and adiabatic of Compressor.

(10hrs)

7. Flow-Measurement:

Pilot-tube, Orifice meter, Venturi meter, Rotameters, Notch air Weirs, Special Flowerets (Hot wire anemometer, Gas meter).

(12hrs)

8. Pumping of Liquids:

Calculation of Total head, NPSH, Performance characteristics curve, Calculation of horsepower, equipment, and cost.

(8hrs)

9. Flow of Fluid Through Granular Beds:

Beds & Packed Column, Fixed beds, Fluidized Bed.

(8hrs)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 142	Fluid Flow	Units 8 Theoretical 3 hr/week Tutorial 1 hr/week Practical 2 hr/week
10. Mixing of liquids in Mixing, Mixers groups, Power consumption	type, Small blade area, large	
11 Two Dhose Flow as	nd Drossuvo duon in ninos s	(6hrs)
11. Iwo- rnase riow a	nd Pressure drop in pipes:	(4hrs)
12. Introduction in uns	<u>teady Flow:</u>	

(6hrs)

	University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
1			Units 3
	CE 132	Properties of Materials	Theoretical 1 hr/week
		•	Tutorial 1 hr/week
			Practical _ 2 hr/week

1. Engineering requirement for materials:

Mechanical Properties, stress-strain curve, elasticity, plasticity, ductility, young modulus, tensile stress, yield stress, bricking stress, true and engineering stress-strain diagram.

(4hrs)

2. Atoms and atomic coordination:

Ionic bond, inter-atomic distance attraction forces between atoms, coordination number, covalent bond, Metallic bond.

(4hrs)

3. Crystal structure:

Crystal structure, unit cell, types of unit cells simple cubic, Face centered cubic, body centered cubic, atomic packing factor, Previous lattice, Miller index, crystalline Planes, X-ray diffraction.

(4hrs)

4. Atomic disorder in solid:

impurities in solids, solid solutions in metals, Imperfection incrystals, point defect, line dislocation, surfaces + grain boundaries.

(4hrs)

5. Atomic movement in solid state:

Atomic diffusion, self-diffusion, diffusivity Fick's first law, diffusion constant, diffusivities versus temperature, Factors effect diffusivity.

(4hrs)

6. Phase diagrams:

Definition of alloys, binary alloys, phase diagrams (equilibrium thermal diagrams), eutectic; solid solution and combination type diagram, Iron-carbon face diagram, allotrope Trans formation (polymorphism).

(4hrs)

7. Corrosion:

Definition, why it happen, Type of corrosion, Form of corrosion, corrosion reactions (anode + cathode) corrosion rate measurement, methods of prevention.

(4hrs)

Note:

• 2nd term

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 222	Computer Programming	Units 6 Theoretical 2 hr/week Tutorial - hr/week Practical 2 hr/week
1. Introduction in visu	al Basic:	
2. Forms and Toll box	controls:	(2hr)
· · · · · · · · · · · · · · · · · · ·	setting a control's name.	
3. Project Explorer:	·	(2hr)
	dow, Properties, Events.	(2hr)
4. Project:	Basic application (project) saving	, ,
Designing the visual	basic application (project) saving	(2hr)
5. Files and projects: Files and projects, E.	xamples, exercise.	
6. Menus:		(2hr)
	rogramming menu commands.	(2hr)
7. <u>Boxes</u> :		(2111)
Dialogue boxes, mes	ssage dialogues Boxes, Font dialog	ue box, color dialogue box,
print dialogue box.		(2hr)
8. General visual Basic		C
declaration statement,	nd string data, variables, Assignated declaration of constants, mathm strings, comment statement, using a	etcd and logic operations,
		(6hr)
	ents: ent, multiple IFTHEN statem EISE, Select case statement, Sw	
choose function.		(10hr)

10. FOR....NEXT Loop, Do WHILE....Loop, Do.... Loop WHILE, Nested Loops.

(6hr)

11. Array:

Arrays definition, Declaring Arrays, using FOR....NEXT Loop with Arrays, Arrays and producers, operations on Arrays.

(6hr)

University of Technology	Chemical Engineering Department	Second Year B.Sc.	Syllabus
·		Units	6
CE 222	Computer Programming	Theoretical	2 hr/week
		Tutorial	- hr/week
		Practical	2 hr/week
Working with stri	ngs, strings management		(01)
. 13. Files:			(2hr)
. 13. <u>Files:</u> Files sequential fi	iles statements User Defined Tyne	Random files	(2hr)
	iles statements, User, Defined Type,	Random files.	(2hr)
	iles statements, User, Defined Type,	Random files.	, ,

		Units	3
CE 342	Fuels Technology	Theoretical Tutorial	l hr/week l hr/week

Chemical Engineering Department

1. Energy and Fuel:

University of Technology

Energy and his classification in Technology section, Physical and chemical classification of fuel, solid Fuels (coal), the industrial processes on coal to Derived different Product.

(3hr)

Second Year B.Sc. Syllabus

Practical 2 hr/week

2. Properties of oil stocks and prepare fractions oil:

Oil stocks, industrial processes which applied from the well to the different fractions in tower distillation and the processes on fraction.

(4hr)

3. Petroleum Fractions properties:

Benzene, Kerosene, Jet Fuel, Diesel, lubricants oil Asphalt, Fuel oil.

(9hr)

4. Gaseous Fuels and Liquefied gases:

Energy source types gases fuel types and liquefied gases which source of energy in industrial and houses.

(4hr)

5. Combustion Calculation:

Different type of equations about the subject and from it found the component which burner and the combustion equation and guess the flue gases and study behavior of combustion.

(6hr)

6. Burner of gases fuel:

The burner types of gases fuel and liquid fuel properties arrange it in Furnace.

(2hr)

7. Combustion in Furnaces:

Furnace types in oil industrial, Boilers and steam production that used in different industrial.

(2hr)

Note:

• 1st term

	**************************************	Units 4
CE 242	Thermodynamics	Theoretical 2 hr/week
		Tutorial 1 hr/week
		Practical - hr/week

Chemical Engineering Department

1. Introduction:

University of Technology

Fundamental quantities Time, Length, Mass, Force, Temperature.

(2hrs)

Second Year B.Sc. Syllabus

2. First Law and Other Basic Concepts:

Joules experiments, Internal Energy Formulation of the first law, Thermodynamics state and state functions Enthalpy, The steady state flow processes, The reversible processes, Heat Capacity and specific Heat, Equilibrium, the phase rule.

(4hrs)

3. <u>Applications of the Equations of Ideal Gases, Generalized Correlation and the Acentric Factor</u>.

(8hrs)

4. Heat Effects:

Heat capacities of gases as a function of temperature, solids and liquids, Heat change accompanying phase change, Heat of industrial reactions.

(6hrs)

5. The Second Law of Thermodynamics:

Heat engine, Entropy, Second law limitation and real process Entropy change and irreversibility and probability (statistical thermodynamics), Third law of thermodynamics.

(6hrs)

6. Thermodynamics Properties of Fluids:

Relationships among the thermodynamics properties (including Helomtz and Gibes free energies and chemical potential), Steam formation and two phase system, Saturated temperature and pressure, Triple point, Wet vapor and dryness fraction, Types of thermodynamic tables and diagrams, Steam power plant cycle and analysis, Barometric condenser, Metering and throttling processes, Steam and as turbines.

(10hrs)

7. Refrigeration and Liquefaction:

Refrigeration cycles (Carnot, Air, Vapor – Compression) and Comparisons, Choice of refrigerant, Absorption Refrigeration, The heat pump, Liquefaction process.

(10hrs)

University of Technology	Chemical Engineering Department	Second Year B.Sc. S	<u>yllabus</u>
		Units	4
CE 222	Thermodynamics	Theoretical Tutorial Practical	2 hr/week 1 hr/week - hr/week

8. Phase Equilibrium:

The nature and criteria of equilibrium, Binary system, Vapor pressure of an ideal solution and non ideal solutions, Henrys law, Activity and activity coefficients, Flash separation calculations.

(8hrs)

9. Chemical Reaction Equilibrium:

Thermodynamics of ideal gases and mixtures, Derivation of the general equilibrium expression, Chemical equilibrium of ideal and non-ideal gases, Reaction equilibrium in solution, effect of temperature on chemical equilibrium.

(6hrs)

University of Technology	Chemical Engineering Department	Second Tear B.Sc. Synabus
	- <u>, , , , , , , , , , , , , , , , , , ,</u>	Units 6
CE 232	Physical Chemistry	Theoretical 2 hr/week
	-	Tutorial 1 hr/week
		Practical 2 hr/week

1. Change of State:

One component systems: Water, carbon dioxide and helium, Vapor pressure and external pressure, Liquid-vapor composition diagram, Liquid-Liquid binary solution mixture, Roulette's law for ideal mixtures, Non-ideal Liquid mixtures, The principle of distillation, Henry's law, solubility of gases in Liquids.

(10hrs)

2. Surface Chemistry:

Pressure difference across curved surfaces, Surface tension, and capillary rise, Langmuir adsorption isotherm and langmuir theory of adsorption, Calculations of surface area of the adsorbent, BET equation for surface area calculation.

(12hrs)

3. Chemical Kinetics:

The rate of chemical reactions, order of reaction and rate constant, zero, first, second and third order rate equations, Reversible reactions, Consecutive reactions, parallel reactions, Determination of the order, Reactions in flow systems, Effect of temperature on reaction rate, The transition-state theory.

(16hrs)

4. Catalysis:

Reaction of catalyst in homogeneity system, Enzyme reactions and Kinetic of reactions

(6hrs)

5. Electrochemistry:

Conductivity measurements, Diffusion and ionic mobilities, Activity and ionic strength, Determination of activity coefficient from solubility, The Debye-Hackle theory, Acid-base catalysis and their dissociation constant.

(6hrs)

6. Electrochemical Cells:

Electromotive force (EMF) of a cell, Measurements of EMF- the potentiometer, The polarity of electrodes, The cell reactions and reversible cells, Free energy and reversible cells, Types of half cells and classification EMF, Standard electrode potentials, Standard free energy and entropy of aqueous ions, Calculation of EMF of a cell, Oxidation-reduction reactions, Concentrations cells, Electrolysis, Corrosion. Kinetic of reactions.

(8hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
CE 133	Applied Mathematics	Units 6 Theoretical 3 hr/week Tutorial 1 hr/week Practical - hr/week

1. Revision ordinary differential equations:

First and second order, simultaneous differential equations, Application for chemical engineering.

(18hrs)

2. Solution of differential equations by series:

Simple series, method of Frobeins, Bassel's equation. Application for chemical engineering.

(18hrs)

3. The Laplace transformation:

The Laplace transformer, Properties of the Laplace transformation. The inverse transformation. Inversionby partial fraction. Convolution. Solution of differential equations. The transforms of special functions. Step function, staircase function, periodic function, Trigonometric function.

(18hrs)

4. Partial Differential Equations:

Formulating of partial differential equations. Solution of partial differential equations. Method of separating variables and Laplace transformation method.

(18hrs)

5. Numerical Analysis:

The difference operators. Interpolation. Finite difference equations. Solution of first and second order differential equations.

(18hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
		Units 7
CE 143	Mass Transfer	Theoretical 3 hr/week Tutorial 1 hr/week
		Practical 2 hr/week

1. Introduction:

(4hrs)

2. Mass transfer fundamentals:

Steady state mass transfer. Diffusion in binary and multicompouent gases. Diffusion in binary and multicompouent solids. Connective mass transfer. Mass transfer coefficients. Two-film theory. Penetration theory. Unsteady state mass transfer methods

(25hrs)

3. Methods of gas-liquid separation (absorption and stripping):

Determination of interfacial mole fractions. Individual and overall mass transfer coefficients. Effect of gas solubility on equilibrium curve. Single stage absorption, multistage absorption. Finding number of stages graphically and through use of correlations for absorption and stripping. Packed columns limits of operation. Design of packed beds for dilute and concentrated solutions. Finding number of transfer units. Absorption with chemical reaction.

(16hrs)

4. Binary distillation:

Introduction, (T-x,y) and (x,y) equilibrium curves. Flash or equilibrium distillation for binary and multicomponent mixtures. Distillation with reflux using Mckabe-Thiele. Sorel method to find number of stages. Fenske equation to find minimum reflux ratio and minimum theoretical stages. Column efficiency. Thermal method for finding number of ideal stages (Ponchoue-Savarit Method). Multicomponent differential distillation.

(25hrs)

5. Multicompouent distillation:

Dew and bubble points calculation methods. Multicomponent equilibrium relationships. Minimum reflux and minimum number of stages. Short-cut method. Checking validity of reflux ratio used. Operating limits of distillation columns.

(14hrs)

6. Mechanical separation process and size reduction:

Classification of equipment. Particle shape factor. Power requirements in crushers. Particle size analysis.

(6hrs)

Note:

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University of Technolog	y Chemical Engineering Department	Third Year B.Sc. Syllabus
CE 233	Statistics and Instrument Measureme	Units 3 Theoretical 1 hr/week Tutorial 1 hr/week Practical 1 hr/week
Part A: Statistics		
	onclusive statistics, population sample	es, graphical representation
of data.		(2hrs)
2. <u>Table and graphic</u> Primary data, repe	cal representation: eatability, normal distribution.	(2hrs)
3. Media determinat	ion:	(2hrs)
4. Scatter determina Standard and relat		
5. Standard distribu	tions:	(2hrs) (4hrs)
6. Chi distribution: Confidence lim	nits, Chi test, degree of freedom.	(2hrs)
-	method, linear relations, exponentia	l equations, gradient, and
correlation coefficien	·	(2hrs)
Part B: Measure	<u>ments</u>	
1. Measurements: Absolute and reproperties.	elative. Environmental conditions.	Application on physical (2hrs)

(2hrs)

2. Errors:

Types of errors, determination of errors.

(4hrs)

3. Instruments:

For temperature, Pressure, level, composition, velocity, flow rate.

(8hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
		Units 6
CE 243	Reactor Design	Theoretical 3 hr/week
		Tutorial I hr/week
		Practical - <u>hr/week</u>

1.Introduction:

Stoichiometry, reaction rate constant, order of reaction, reversible and non-reversible reactions, conversion, yield, selectivity.

(9hrs)

2. Types of reactors:

Batch reactors, continuous flow reactors i.e. CSTR, PFR. Industrial reactors.

(9hrs)

3. Bach reactors:

Design equation, reaction time, maximum production rate, non-isothermal operation, adiabatic operation

(18hrs)

4. Tubular reactors:

Design equation, isothermal and non-isothermal operation, residence time, length of reactor, number of tubes, pressure drop, space velocity.

(18hrs)

5. Continuous stirred tank reactions:

Design equation for isothermal operation, scale-up of liquid-phase batch data to the design of CSTR. CSTR's in parallel and in series.

(15hrs)

6. Unsteady state operation of reactors:

Start up of CSTR, semi-batch reactor.

(9hrs)

7. Multiple reactions:

Operation in parallel and in series for single and multiple reactions to obtain maximum amount of desired product.

(6hrs)

8. Comparison between reactors:

Comparison between batch, tubular, and CSTR for single and multiple reactions

(6hrs)

University of Technol	ogy Chemical Engineering Department	Third Year B.Sc. Syllabus
	And the second s	Units 7
CE 343	Heat Transfer	Theoretical 3 hr/week
		Tutorial l hr/week
		 Practical 2 hr/week

1. Modes of heat transfer:

Material properties of importance in heat transfer.

(2hrs)

2. Steady state heat conduction in one dimension:

Plane wall, radial systems, heat source systems. Boundary surrounded by fluids. Overall heat transfer coefficient. Extended surface, conduction-convection systems, fins.

(24hrs)

3. Principles of convection:

Transport equations. Fluid mechanism aspect of convection. Laminar boundary layer. Thermal boundary layer. Empirical and practical relations for pipe and tube flow. Flow normal to single and tube banks.

(18hrs)

4. Heat exchangers:

Various types and their general characteristics. Fouling factor. Heat exchanger mean temperature differences. Co-current and counter-current flow.

(12hrs)

5. Shell and Tube Exchangers:

Types and various specifications. Design calculations by conventional and by effectiveness (NTU) methods. Optimum design calculation.

(6hrs)

6. Condensation and Boiling Heat Transfer:

Condensation of single vapors. Design calculations for condenser, condensersubcooler and superheated condenser.

(9hrs)

7. Radiation and furnace design:

Radiation properties, shape factor, heat exchange for non black bodies, parallel planes, shields, gas tradition, boiler.

(9hrs)

8. Unsteady state heat transfer:

Temperature as a function of time, lumped capacity system, quenching of small bodies, heating of tank reactor.

(10hrs)

Note:

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University of Techno	ology Chemical Engineering Departmen	Third Year B.Sc. Syllabus
<u> </u>		Units 6
CE 443	Equipment Design using Computer	Theoretical 2 hr/week Tutorial - hr/week
		Practical 1 hr/week

Part (A)

1. Process Planning:

Scheduling and flow sheet design. Flow sheet types and designation. Block diagram. Process flow sheet. Piping and instruments diagram. Utility flow sheet. Equipment layout and plot plan. Project evaluation. Feasibility studies. Site selection. Project cost estimation.

(6hrs)

2. Piping networks and pumps:

Pipe fittings, valves. Steam traps. Piping design standards. Piping material and selection. Pump specifications and data sheets. Optimum diameter. Types of pumps. Hydraulic characteristics for centrifugal pumps.

(6hrs)

3. Vessels and Tanks:

Types of vessels. Flash drums. LPG tanks. Criteria in vessel design. Stress considerations. Design of tall vertical vessel. Design of pressure vessels. Supports and foundations

(6hrs)

4. Heat transfer equipment:

Types of exchangers and applications. Exchanger ratings. Exchanger design. Exchanger specification sheets. Furnaces. Convection and radiation zones. Types of fuels. Burners and arrangements. Steam boilers, types of boilers.

(6hrs)

5. Mass Transfer Equipment:

Types of columns. Plate and packed Types of plates and packing. Design features. Pressure drops in columns.

(6hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc.	Syllab <u>us</u>
		Units	6
CE 443	Equipment Design using Computers	Theoretical	
		Tutorial Practical	- hr/week 1 hr/week

Part (B): Complete equipment design of each of the following units:

1. Pressure vessels, pumps and compressors:

Flash drum. Gas-liquid separator, LPG storage tanks. Cyclone separators. Gas movers and compressors. Pumps. Piping network. etc.

(8hrs)

2. Heat Equipment:

Shell and tube heat exchangers. Plate heat exchanger. Boilers. Furnaces. Vaporizers. Air coolers. Coil type heaters. Condensers. etc.

(10hrs)

3. Mass transfer equipment and reactors:

Distillation column. Absorber. Extractor. Dryer. Absorption towers. Leaching equipment. Scrubbers. Stripping towers. Reactors.

(12hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
CE 543	Biochemical Engineering Unit Operations Branch	Units 2 Theoretical 2 hr/week Tutorial 1 hr/week Practical - hr/week
1. Definition, introducti	ion:	(2hrs)
	anisms and methods of separation	<u>n:</u> (2hrs)
3. Micro organisms gen		(4hrs)
4. <u>Biotechnological met</u> Liquid systems, b Fermentation's	atch and continuous cultures.	Breaded cultures Solid
5. Single cells proteins:		(6hrs)
	d, treatment of waste, production	
6. Bakery yeast:		(3hrs)
Dry and ready effe	ective types.	(3hrs)
Clotamic acid.		(1hr)
8. Organic acids: Acetic acid.		
9. Enzymes:		(1hr)
10. Antibiotics: Microorganisms prod	lucing antibiotics, Penicillin	(1hr)
11. Industrial Alcohol:	deling until of other states of the states o	(1hr)
	on, microorganism used, production	n method. (1hr)
12. Animal cell cultivati		(2hrs)
13. Plant cell Cultivatio	ning using biotechnology:	(1hr)
	ter treatment. Organic Pollutants. F	Petroleum wastes. (2hrs)

University of Technolog	gy Chemical Engineering Department	Third Year B.Sc. Syllabus
CD (42)		Units 2
CE 643	Environmental Engineering	Theoretical 2 hr/week Tutorial 1 hr/week
	Unit Operations Branch	Practical - hr/week

1. Introduction:

What is pollution and a historical brief on international concern with pollution. Types of pollutants and measurements. Pollution allowable limits in the environment. (2hrs)

2. Cleaning of air and industrial gases:

Sources and effects of air pollution. Methods of control. Designing pollution equipment. Settling chamber, Cyclone separator, Venture scrubber, spray tower, cyclone scrubber, packed bed, electrostatic precipitator. Economic evaluation of gas cleaning.

(10hrs)

3. Water and heavy water treatment:

Physical methods of treatment. Occupation methods. River water cleaning, heavy water treatment. Desalination.

(10hrs)

4. Industrial and sewage treatment:

Industrial water treatment. Sewage water treatment.

(4hrs)

5. Solid pollutants:

Sources and effects. Methods of control of solid pollutants.

(4hrs)

University	of Technology	Chemical Engineering Department	Third Y	ear B.Sc.	<u>Syllabus</u>
	, , , , , , , , , , , , , , , , , , , 	<u> </u>		Units	2
CE 543	Storage and Tr	ansport of Crude and Petroleum Pi	<u>oducts</u>	Theoretical	*
				Tutorial	1 hr/week
		Rofinory Fuginooring Branch		Practical	- hr/week

1. Fluid Flow:

Petroleum liquids flow. Petroleum gases flow. Designing pipe network from optimum velocity and diameter viewpoint. Crude and petroleum products pipelines, Heating and pumping stations, Increasing pumping rates by looping. Increasing pumping rates by polymer additives. Two-phase flow. Unsteady state flow and hammering.

(14hrs)

2. Pressure Vessels / Tanks:

Stress systems of shells. Bending and membrane shells. Equilibrium equation for a shell element and LPG tanks, Rotationally symmetrically loaded membrane shell. Crude oil huge tanks.

(8hrs)

3. Selection of Materials for Vessels and Pipelines:

Selection of materials for vessels to limit corrosion. Selection of materials for pipelines to limit corrosion. Catholic protection.

(8hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
CE 643	Combustion Engineering	Units 2 Theoretical 2 hr/week
	Refinery Engineering Branch	Tutorial I hr/week Practical - hr/week
1. Combustion reaction	stoichiometry:	
		(8hrs)
2. Excess air and flame	temperature:	
3. Combustion product	_	
4. Gas, solid, and liquid	<u>l fuels:</u>	
5. Flammability limits:		(Oh.ma)
6. Flame and combustic	on speed.	(8hrs)
7. Flame stability:	on speed.	
8. Gas, liquid and solid	fuels burners:	
		(6hrs)
9. Combustion zones an	id temperature profiles:	, ,
10. Radiation and conv	ection rooms in furnaces:	
11. Furnaces and the tu		
12. Chimney height cale	<u>eulation:</u>	/nt >
40.77		(8hrs)
	and heat loss calculations:	
14. Furnace wall layers	and retractories:	

15. Tube layers in furnaces:

University of Technology	Chemical Engineering Department	Fourth Year B.Sc.	<u>Syllabus</u>
CE 144	Project	Units Theoretical	•
		Tutorial Practical	- hr/week

1. Introduction:

Literature, Methods of manufacturing, Capacity of the Process Physical and Chemical Properties

2. Material and Energy Balances:

Review of Material and Energy balances, Flow sheet Symbols, PFD In formation in Flow Diagram

3. Choice of Process Devices and Equipments:

Pumps, Valves, Pipes, Measuring Devices and Controllers, Material of Construction, PID Plot

4. Choice of Plant Location and Layout Standards:

5. Industrial Services:

Cooling Tower, Boilers, Study Problems from Industrial Waste Disposals, and How to Prevent the Environment, pollution.

6. Use a Computer Program to Design:

Introducing a Complete Project with Material and Energy Balances Design of Devices, Calculation of Costs, Choice of Location

	University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
Į	CE 244	Transport Phenomena	Units 7 Theoretical 3 hr/week Tutorial 1 hr/week
			• Practical 3 hr/week

1. Boundary Layer and analogies:

Boundary Layer in Laminar and turbulent Flow, Boundary Layer in Laminar sub Layer, Transition layer, Velocity distribution on Surfaces and Pipes, Momentum, heat and mass transfer molecular diffusion, Eddy transfer, Reynolds analogy, Modified Reynolds analogy, Chilton and Colburn analogy.

(20 hrs)

2. Evaporation:

Introduction, Types of Evaporations, Evaporation Equipment, Heat transfer in Evaporation Process, single, double and Multi effect Evaporators, Design of evaporators, Comparison of Forward, backward and Parallel effect evaporators, boiling Point rise.

(12 hrs)

3. Drying:

Introduction and general Principle in drying, Rate of drying, the mechanism of moisture movement, Calculation of rate of drying, moisture transport in Solids at Constant and falling rate Period, Capillary movement, Material and Energy Balances in Continuous dryers, Types of Dryers

(8 hrs)

4. Humidification, dehumidification and Cooling towers:

Humidification, dehumidification, Temperature humidity Chart for air – water system. Enthalpy – humidity –temperature chart. Addition of Vapor or liquid Stream to a gas Stream. Mechanism of dehumidification Evaluation of heat and Mass transfer Coefficient, Cooling tower, height of Packing in Cooling towers, Minimum gas Condition, Change of Condition of liquid and gas in Cooling towers.

(14 hrs)

5. Extraction:

Definition, Extraction process, Equilateral Triangular coordinates (Ternary Diagram), system of three liquid one pair partially soluble, system of three liquid two, pairs partially soluble, choice of solvent, Equipment in extraction cross current extraction, multi-stage Cross Current extracting cross current for insoluble Liquid, Continuous Counter current extraction, Continuous Counter Current in Soluble, Liquid, Minimum Solvent, Counter Current extraction with reflux.

(22 hrs)

Note:

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Omversity of Technology	Chemical Engineering Department	rourin fear b.sc. synabus
CE 244	Transport Phenomena	Units 7 Theoretical 3 hr/week
		Tutorial 1 hr/week Practical 3 hr/week

6. Filtration:

Type of Filters, Filtration theory, Plate and frame filter press, leat filter, filtration at Constant ΔP , Filtration at Constant rate, washing Time.

(10 hrs)

7. Crystallization:

Introduction Equilibrium Solubility in crystallization, Classification of Crystallizes, Nucleation theory, Rate of Crystal Growth, Material and Energy balance in Crystallization.

(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
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CE 344	Process Control	Theoretical 2 hr/week

Process Control

Theoretical 2 hr/week
Tutorial 1 hr/week
Practical 3 hr/week

1. Revision of Laplace Transformation:

(4 hrs)

2. Process Dynamic and Transient Response of the system:

Dynamic and Control For chemical Process, Dynamic behavior for 1st order system, lineraization, Interacting and non-interacting, Second order system, Transportation.

(16 hrs)

3. Characteristics of the closed loop System:

Over all closed loop transfer functions and block diagram algebra Transient of Simple Closed loops System, Stability of Control, Introduction to the Frequency analysis and design technique.

(20 hrs)

4. Industrial Controller:

Actions selection Criteria for Various Control modes, final Control element, Dynamics and Control of Chemical reactor System.

(10 hrs)

5. Dynamic and Control of some Complex Chemical Process:

Heat exchanger, distillation Column, Introduction to Computer Control of Chemical Process.

(10 hrs)

Note:

• 2nd term

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
6D 141		Units 6
CE 444	Chemical Industries	Theoretical 2 hr/week
	Unit Operation Branch	Tutorial 1 hr/week Practical 3 hr/week
	Onu Operation Branch	Tractical 3 III/ WEEK
1. Chemical Processing	:	
	ng Conditions, Type of Process.	
, _ 		(3 hrs)
2. Sulpher and Sulpher	ic acid:	(3 1115)
	ch Process, Iead Chamber Proce	ss Contact Process Oleum
Production.	en Trocess, read Chamber Troce	ss, Contact 1 10ccss, Octum
Froduction,		(A tours)
4 • 1374 •	• •	(4 hrs)
3. Ammonia and Nitric		
Raw materials, Produ	ction Steps, Production Process.	
		(6 hrs)
4. Nitrogenous Fertilize	ers:	
Ammonium Sulphate	, Ammonium nitrate, Urea.	
		(4 hrs)
5. Phosphoric acid:		
	oric and, Electric Furnace	
		(4 hrs)
6. Phosphate Fertilizers	g•	(1.22)
Superphosphate, Trip		
Superphosphate, Trip	ne Superphosphate.	(A hara)
7 Electric buttle Industri		(4 hrs)
7. Electrolytic Industric		D 1
Principle and theory,	Caustic Soda Production Chlorine	
		(4 hrs)
8. <u>Industrial Salts:</u>		
Sodash Production		
		(3 hrs)
9. Ceramic Industries:		
Raw materials, Chem	ical Conversions, Porcelain, Fire C	Clay
· ·	,	(4 hrs)
10. Cement Industrie	s:	` '
Raw materials, Portla	 -	
Tan many and I of the	and Commit	(4 hrs)
11. Glass Industries:		(4 1115)
	facture of Class	
Raw materials, Manu	racture of Grass	(4.1)
		(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 444	Chemical Industries	Units 6 Theoretical 2 hr/week
	Unit Operation Branch	Tutorial 1 hr/week Practical 3 hr/week
12. Surface Coating I Paint, varnishes, lacq		
•	uers	(4 hrs)
13. Oils and Fats: Fatty acids, Manufac	ture of Vegetable Oils and Waxes	/41 N
14. Soap and Deterge		(4 hrs)
Soap Production, Det	ergents Production	(4 hrs)
15. Sugar Industries: Cane Sugar, Beet Sugar,		
	,	(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 234	<u>Optimization</u>	Units 3 Theoretical 2 hr/week Tutorial 1 hr/week Practical - hr/week
1. Introduction to Opti	mization Methods:	(A hus)
2. Organization of Opt	imization problems:	(4 hrs)

3. Single Variable:

Analytical methods, numerical methods, graphical methods, numerical Search, restriction Function, unrestricted Function, Direct search, Dichotomous Search, golden Search, Fibonacci Search

(12 hrs)

(4 hrs)

4. Multivariables Optimization Methods:

- Necessary Conditions For Extreme Values in graphical Cases.
- Solution by graphical method.
- Simplex method.
- Linear Programming and application in Chemical engineering (transportation mixing)

(10 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabu
	<u> </u>	Units 2
CE 644	Engineering of Catalysts	Theoretical 2 hr/wee Tutorial 1 hr/wee Practical - hr/wee
Part One		
1. <u>Catalysts:</u> Definitions, Properties	es of Catalysts	
2. Steps in a Catalytic I	•	(2 hrs
	Reaction, Decoration, Rate Equation	•
3. Finding Rate Law an Equations of Rate Lir	· · · · · · · · · · · · · · · · · · ·	(4 hrs
•	-	(4 hrs
Basic Guidelines. Des	<u>r Gas – Solid Reactions:</u> sign Equations.	
		(8 hrs

5. Heterogeneous Data Analysis for Reactor Design:

Deducing a Rate law from the Experimental Data, Finding a Mechanism Consistent with Experimental Observations, Evaluation of the Rate law Parameters, Reactor Design

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
		Units 2
CE 644	Engineering of Catalysts	Theoretical 2 hr/week
	· · -	Tutorial 1 hr/week

Part Two

1. Introduction:

Properties of Catalysts, Determination of Surface Area, Void Volume and Solid Density, Pore Volume Distribution, Classification of Catalysts, Promoters and Inhibitors, Catalyst Poisoning.

(4 hrs)

- hr/week

Practica!

2. Rate Equations of Fluid-Solid Catalytic Reactions:

Rate of Adsorption, Desorption and Surface Reaction, Rate of the limiting Step, Mechanism, Design of Reactor for- Gas- Solid Reactions. Data Analysis for Reactor Design

 $(10 \, hrs)$

3. External Transport Processes in Heterogeneous Reactions:

Equations of Mass –and Heat Transfer, Calculations of the Concentrations on the Surface, Calculations of Temperature Differences in External Transport Processes, Fluidized Bed Reactors, slurry Reactors

(4 hrs)

4. Internal Transport Processes:

Gaseous Diffusion in Single Cylindrical Pores, Surface Diffusion, Effectiveness Factor, Thiele – Type Modulus

(4 hrs)

5. Design of Heterogeneous Calalytic Reactors:

Fixed - Bed Reactors, Fluidized Bed Reactor, Slurry Reactions

(8 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 844	Corrosion Engineering Unit Operations Branch	Units 2 Theoretical 2 hr/week Tutorial 1 hr/week Practical - hr/week
Introduction: Definition of Corrosi	on, why it happens, direct and indi	
2. Types of Corrosion: Dry Corrosion, wet C	Corrosion, Eight form of Corrosion	(2 hrs)
3. Mechanism of Crevi	ce Corrosion:	(4 hrs) (2 hrs)
4. Electrochemical cell: Electrode Potential, t	ypes of reference electrodes.	(4 hrs)
5. Corrosion rate Calcu Wet loss method, elec		(2 hrs)
6. <u>Polarization:</u> Definition, Types.		(4 hrs)
7. <u>Passivity:</u>8. Effect of Temperature	re and velocity on corrosion reac	(2 hrs)
9. Method of Protection		(2 hrs) (4 hrs)
10. <u>Instruments to mo</u>	easure Corrosion rate:	(4 hrs)

University of Tec	hnology	Chemical Engineering Department	Fourth	Year B.Sc.	Syllabus
			· · · · · · · · · · · · · · · · · · ·	Units	2
CE 134	<u>M</u>	anagement and Engineering Econo	<u>mic</u>	Theoretical	2 hr/week
				Tutorial	1 hr/week
				Practical	- hr/week

1. Principles of Management and its application in industry:

Types and duties (such as Production, marketing, Personal management, Financial duties etc)

(5 hrs)

2. Feasibility Study of industrial site:

This will include industrial Site and Factors affecting Choice of the site.

(5 hrs)

3. Production and Productivity:

Types of production, optimum Design, Production Lines, Plant Layout, Machine Balance.

(5 hrs)

4. Maintenace and its types:

Costs of maintenance machine Replacement.

(5 hrs)

5. Critcal Path Method (CPM):

Types and applications

(5 hrs)

6. Quality Control in industry:

Quality Control in industry and its applications in Production lines .ISO and its application in industry.

(5 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 744	Polymer Technology	Units 2 Theoretical 2 hr/week Tutorial 1 hr/week
· · · · · · · · · · · · · · · · · · ·	Unit Operations Branch	Practical - hr/week
1. Polymers: Nomenclature		
		(2 hrs)
2. <u>Polymers Classificat</u> Classification based Polymers, Technical Cla	on Sources, Classification base	ed on Chemical nature of
		(2 hrs)
Polymerization, Ionic (merization (condensation polycationic, Anionic) Coordination Porization by group transfer	•
1 ory merizaciony 1 ory me.	inzarion by group transiti	(6 hrs)
4. Polymerzation System		
Bulk, Solution, Suspe	ension, Precipitation	(4 hua)
5. Molecular Weight of	Polymers:	(4 hrs)
	nent of molecular weight of Polym	ers
< p. 1		(2 hrs)
6. Polymer composites:		(2 hrs)
7. Polymer blends:		(2 1113)
		(2 hrs)
8. Physical Properties of	of Polymers:	(4 L)
9. Mechanical Properti	es of Polymers:	(4 hrs)
	act strength, Creep, hardness	
		(4 hrs)
10. Polymer Processin		
Molding, Extrusion,	Thermal Forming Calendaring	

(4 hrs)

	Vancous Consenses Consense	
University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 444	Petroleum Refining	Units 6 Theoretical 2 hr/week Tutorial 1 hr/week
· · · · · · · · · · · · · · · · · · ·	Refinery Engineering Branch	Practical 3 hr/week
1. Introduction: History and Development	ment of Refining Processes Kinds	of Refineries.
2. Chaminal Canada's		(2 hrs)
2. Chemical Composition 3. Physical and Thermo	on of Petroleum: odynamic Properties of Petroleu	(2 hrs)
3. 1 nysical and 1 nermi	odynamic i ropernes of retroleu	(2 hrs)
4. Evaluation of oil stoo	eks:	,
5. Introduction to Proc	essina:	(4 hrs)
	ation, tube still heaters	
		(10 hrs)
6. <u>Fractionation of Petr</u> Atmospheric and Vac		
ranospherie and vac	dum i ractionation	(2 hrs)
7. Fractionation Tower		
Material and Energy Tower, Tower Diameter	Balances, Reflux Temperate D	
8. Treating Processes:		(8 hrs)
	ses, Sweetening Processes, improv	vement in Performance and
storage stability.		(8 hrs)
9. <u>Upgrading processes</u> Thermal Crackling,	<u>:</u> Cooking, Visbreaking, Catalytic,	•
Catalytic Reforming, All	cylation, Isomerisation.	
10. Product Blending	•	(10 hrs)
	ressure, octane number, viscosit	y, flash point aniline point,
Lan barre		(6 hrs)
11. Manufacturing of	lubricating oils:	(4 1)
12 Tuesting wasses	sain - IYadaa saa	(4 hrs)

(2 hrs)

12.

Treating process using Hydrogen:

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 544	Gas Technology	Units 2 Theoretical 2 hr/week Tutorial 1 hr/week
	Refinery Engineering Branch	Practical - hr/week
1. Classification of Gas	eous fuels thermodynamic:	
	•	(3 hrs)
2. Physical and Chemic	cal Properties for natural Gas:	
2 During of Consons F	la-	(3 hrs)
3. <u>Drying of Gaseous F</u> Drying by Adsorption	n, Drying by Absorption	(41)
4. Sweetening of Gaseo	us Fuels:	(4 hrs)
	ption, Sweetening by Absorption	
		(6 hrs)
5. Gas Processing units		ant as misture
riash Separation Proc	cesses, Distillation of multicompor	(6 hrs)
6. Liquefied Petroleum	Gases:	(0 1113)
		(2 hrs)
7. Production of Hydro	gen:	
0.0.10		(2 hrs)
8. Sultur recovery proc	esses from Acid gases streams:	(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
		Units 2
CE 744	<u>Petrochemical Industries</u>	Theoretical 2 hr/week Tutorial 1 hr/week
	Refinery Engineering Branch	Practical - hr/week
2. Basic Petrochemical		
		(1 hrs)
	racking Processes Cooling, Pur	ification Processes Cooling
	Processes Compression, Separ	-
	ources of diolefins, Butadiene	
- ·	Process For Production of	_
	higher olefins, linear, Alkyl ber	-
Basic materials Aroma	atic Sources, Separation of benz	ene - Toluene - Xylene,
Syntheses Gas, Steam re	forming, Partial Oxidation	•
		(10 hrs)

4. Intermediate Petrochemicals:

3. Synthetic Zeolites:

Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, Vinylchloride, Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, EthylbenZene, Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Terephthalic acid.

(7 hrs)

(2 hrs)

5. Petrochemical Detergents:

(2 hrs)

6. Polymers:

LDPE, HDPE, PP, PVC

(4 hrs)

7. Synthetic Fibers:

(2 hrs)

8. Petrochemical Complexes:

(2 hrs)

		Units 2
CE 844	Environmental - Pollution	Theoretical 2 hr/week
		Tutorial 1 hr/week
	Refinery Engineering Branch	Practical - hr/week

Chemical Engineering Department

1. Introduction:

University of Technology

Definition of Pollution, literature surveys on international interest of Environmental Pollution Problems. Pollutants, Type of Pollutants, Measuring Devices, Allowable level of Pollutants in Environment.

(2 hrs)

Fourth Year B.Sc. Syllabus

2. Air Pollution:

Air Pollutants, Particulate matter, SO₂, CO, NO, ...etc Sources and effect, Control Treatment Methods, Design of some Equipment: Centrifugal, Cyclone, Scrubber, Packed Bed Ventura, Spray – tower, Electrostatic Separator, and Chimney Design.

(12 hrs)

3. Water - Pollution:

Water Pollutants, Sources and effect, Organic and non – organic, Radiation activity, Suspended Solids and Sculling Salts, Control Treatment of Water Pollution for: Drinking, Industry Processes, Desalination methods, Reverse Osmosis, Electrodalysis, Solar Distillation, Primary, secondary and advanced treatment of Wastewater, Treatment of oily Pollution.

(12 hrs)

4. Pollution of Solid - Waste:

Sources, Effects, Control Treatment method of Removal Solid – Waste, Treatment by economy methods.

(6 hrs)

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