Ministry Of Higher Education

University Of Technology Chemical Engineering Department



B.Sc PROGRAMME IN CHEMICAL ENGINEERING

OUTLINE OF SYLLABUSES ALLOCATION OF SUBJECTS AND WEEKLY LOAD

2004 - 2005

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INTRODUCTION

Chemical Engineering is distinguished from other branches of engineering by its strong dependence on chemistry. This enables the chemical engineer to understand properties involving changes in physical state, chemical composition, or energy content for systems ranging in scale from molecules to full sized manufacturing plants.

The chemical engineer may be employed in an established industry producing chemicals, petroleum products, petrochemicals, pharmaceuticals, synthetic fibers, foods, plastics or metals. These products are steadily needed in an increasing amounts due to the expanding growth of population. His function may be involved in making innovations in operations, doing research, development and analysis of existing plants, technical services, or scale up.

Due to his board knowledge, the chemical engineer often occupies a dominant position in the above mentioned industries.

He also apply his knowledge in such diverse areas as air and water pollution or biochemical research.

The formal course work for B.Sc. programme involves mathematics, including both analysis and computer application, chemistry, mechanics, electricity chemical engineering principles and practice subjects including fluid dynamics, heat, unit operation, chemical reactor design, thermodynamics, and chemical processes. In addition the programme involves English Language which is essential for any engineering student to enable him/her to benefit from thousands of textbooks and periodicals.

A considerable emphasis is laid, in this department, on practical training and laboratory work, which helps students in getting better understanding of the theoretical part of their curriculum.

Moreover students are required to spent twelve weeks, training in industry during summer vacations.

This booklet contains the academic programme and the syllabus for subjects which are taken by the chemical engineering students during their four years study.

Prof. M. A. Zablouk Chairman of Chemical Eng. Dept.

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		Hours / Week						
Subject Symbol	Subject	F	irst Term		Se	Units		
<i>5,111001</i>		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 111	English Language	2	-	-	2	-	-	4
CE. 121	Mathematics	3	-	1	3	-	1	6
CE. 141	Principle of Chemical Engineering	3	-	1	3	-	1	6
CE. 221	Chemistry	3	2	1	3	2	1	8
CE. 131	Engineering Drawing and AutoCAD	1	2	-	1	2	-	4
CE. 231	Mechanics and Strength of Materials	2	-	1	2	-	1	4
CE. 331	Electrical Technology	1 .		1	1	-	1	2
CE. 321	Computer Science	2	2	-	2	2	-	6
CE. 431	Workshop	-	6	-	_	. 6	-	4
	Total	17	12	5	. 17	12	5	44

Note: The student is required to complete twelve weeks summer training in industry

Number OF Hours / Week

First Term	Second Term
34	34

Key of subject symbol (from lift):

CE: Chemical Engineering

First No.: Order of subject with specialization

Second No.: specialization

1: Human subject

2 : Basic subject

3 : General subject

4 : Specific subject

Third No.: Year of Study

-U: Unit Operation OF Chemical Industry Branch

-R: Refinery Engineering of Oil and Gas Branch

		Hours / Week							
Subject Symbol	Subject	F	irst Term		Se		Units		
0,111001		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial		
CE. 112	Human Rights	2		*	. 2	-	-	4	
CE. 122	Mathematics	2	•	1	2	-	1	4	
CE. 142	Fluid Flow	3	2	1	3	2	1	8	
CE. 132	Properties of Materials	1	-	. 1	1	2	1	3	
CE. 222	Computer Programing & MATLAB	2	2	_	2	2	-	6	
CE. 342	Fuels Technology	1	2	1	1	-	1	3	
CE. 242	Thermo dynamics	2	-	1	2	- ' '	1	4	
CE. 232	Physical Chemistry	2	2	1	2	2	1	6	
	Total	15	8	6	15	8	- 6	38	

Number OF Hours / Week

F	irst T	erm	: 3	7.32.30		W 10
	29			29	•	

A-Unit Operation OF Chemical Industry Branch

		Hours / Week						
Subject Symbol	Subject	First Term			Se	Units		
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 113	Democracy	2	_	-	2	-	-	4
CE. 133	Applied Mathematics	3	_	1	3	-	1	6
CE. 143	Mass Transfer	3	2	1	3	-	1	7
CE. 233	Statistics and Instruments Measurement	2	.		2	•	-	4
CE. 243	Reactor Design	3	-	1	3	-	1	6
CE.,343	Heat Transfer	3	-	1	3	2	1	7
CE. 443	Equipment Design by Computer	2	1 1	-	2	1	- -	6
CE. 543-U	Biochemical Engineering	2	-	1	-	-	-	2
CE. 643-U	Environmental Engineering	-	· -	-	2	-	1	2
	Total	20	3	5	20	3	5	44

Number OF Hours | Week

First Term	Second Term
28	28

B-Refinery Engineering of Oil and Gas Branch

•		Hours / Week						
Subject Symbol	Subject	First Term			Se	Units		
5 , • • •		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 113	Democracy	2	•.	-	2	-	_	4
CE. 133	Applied Mathematics	3	-	1	3	-	1	6
CE. 143	Mass Transfer	3	2	1	3	-	1	7
CE, 233	Statistics and Instruments	2	-	-	2	-	-	4 -
CE. 243	Reactor Design	3	-	1	3	-	1	6
CE, 343	Heat Transfer	3	-	1	3	2	1	7
CE. 443	Equipment Design by Computer	2	1	_	2	1	-	6
CE. 543-R	Storage and Transport of Crude and Petroleum Products	2	-	1	-	-		2
CE. 643-R	Combustion Engineering	-	-	-	2	-	1	2
	Total	20	3	5	20	3 .	5	44

Number OF Hours | Week

First Term	Second Term
28	28
	1

A-Unit Operation of Chemical Industry Branch

Cartina		Hours / Week						
Subject Symbol	Subject	First Term			Se	Units		
·		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 144	Project	-	2	-	_	2	-	2
CE. 244	Plant Design	1	1	-	1	1	_	3
CE. 344	Transport Phenomena	3	3	1	3	-	1	7
CE. 444	Process Control	2	-	1	2	3	1	5
CE. 544-U	Chemical Industries	2	3	1	2	3	1	6
CE. 134	Optimization	- .	· -	- '	. 2	_	1	2
CE. 644-U	Engineering of Catalyst	2	· ·	1	-	-	-	2
CE. 744-U	Corrosion Engineering	-	_	<u>.</u>	2	_	1	2
CE. 844-U	Technology of Oil and Gas	· -	-	-	2	-	1	2
CE. 234	Industrial Management and Safety	2	-	1	-	-	-	2
CE. 944-U	Polymers Technology	2	-	1	_	_	_	2
	Total	14	9	6	14	9	6	35

Number OF Hours / Week

First Term	Second Term
29	29

B-Refinery Engineering of Oil and Gas Branch

_		Hours / Week						
Subject Symbol	Subject	First Term		Second Term			Units	
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 144	Project	-	2	-	-	2	_	2
CE. 244	Plant Design	. 1	1	-	1	1	-	3
CE. 344	Transport Phenomena	3	3	1	3	-	. 1	7
CE. 444	Process Control	2	_	1	2	3	1	5
CE. 544-R	Petroleum Refining	2	3	1	2	3	1	6
CE. 134	Optimization	-	-	_	2	_	1	2
CE. 644-R	Engineering of Catalyst	-	-	_	2	_	1	2
CE. 744-R	Petrochemical Industries	-	-	-	2	-	1	2
CE. 844-R	Gas Technology	2	-	1	-	-	<u>-</u>	2
CE. 234	Industrial Management and Safety			-	2	_	1	2
CE. 944-R	Environmental Pollution	2	-	1	-	-	-	2
	Total	14	9	6	14	9	6	35

Number OF Hours I Week

Pirst Term	Second Term
29	29

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 111	<u>English</u>	Units 4 Theoretical 2 hr/week Tutorial - hr/week Practical - hr/week
First Semester		
1- Comprehension Reading Scientific Pas A- Solids, liquids a B- The Control of I C-Sir Isaac Newton	nd Gases Electric Currents	
3.C (1.1.D.)		(18hrs)
2- Grammatical Points A-The Sentence 1- Affirmative 2- Interrogative 3- Negative		
5- Negative		(4hrs)
B-Parts of Speech 1- Nouns A-Function B-Countibility C-non-Countibili	ty	(11)
2-Pronouns: Types and function		(1hrs)
3-Adjectives A-Degree/Comparis	son	(1hrs)
B- Order		(1hrs)
4-Verbs A-Finite B-Non- Finite C-Transitive D-Intransitive	v.	
5-Adverbs A-Classification B-Formation C-Order		(1hrs)
2 2- 2-2		(1hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 111	<u>English</u>	Units 4 Theoretical 2 hr/week Tutorial - hr/week Practical - hr/week
6-Prepositions: A general idea		
7-Conjunctions A-Main Conjunctions B-Subsidiary	s	(1hrs)
8-Interjections: Common ones		(11115)
		(1hrs)
Second Semester		•
1- Comprehension Reading Scientific Pass 1-Electricity in Early 2-Comets	y Days	
3-Exploring the mod 2-Grammatical Points	on	(16hrs)
1-Articles A-definite B-indefinite		
2-The Sentence		(2hrs)
A-Structure B-Types C-Main Parts: 1-Sul	bject edicate	
D-Expansion	carcate	
3-Written and Oral Tra	nslation	(8hrs)
		(4hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 121	<u>Mathematics</u>	Units 6 Theoretical 3 hr/week Tutorial 1 hr/week Practical - hr/week
Range, Inverse of funct	nt line, Trigonometric functions ar ions, Absolute value, limits, L ion) Conic sections (general defini	imits, applications, Pola
•	•	(12 hrs
2- Differentiation and Interest Algebraic functions and	trigonometric functions, Application	on (limits, Hopitals rule) (6 hrs
3-Transcendental Function		
	ric functions, Natural logarithmic	
unctions (Definitions, Pro	perties, Sketches, Derivatives and I	(12 hrs
(11		
- Hyperbolic Functions: Definitions, Properties,	Sketches, inverse and Derivatives a	and Integrals. (8 hrs

Area between two curves, Volumes, Length of curves, Surface area.

(12 hrs)

7- Moods of integration:

Substitution, Integration by parts, The Substitution $U = \tan \frac{X}{2}$, Power of trigonometric functions, Integrals involve $\sqrt{a^2-\chi^2}$, $\sqrt{a^2+\chi^2}$, $\sqrt{\chi^2-a^2}$, integrals of $ax^2 + bx + c$, Partial functions, Improper Integrals (conv and dity)

(16 hrs)

8- Complex Numbers:

Definitions, Argands diagram, Addition and subtraction and multiplication and dividing numbers, Trigonometric and exponential formula of complex number, Demoivors theorem, Roots of equation, Power of complex numbers.

(9 hrs)

9- Vector Analysis:

Definition, Addition and subtraction of vectors, Multiplication by scalars, Direction, products of vector (scalar product and vector product).

(9 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 141	Principle of Chemical Engineering	Units 6 Theoretical 3 hr/week Tutorial 1 hr/week Practical - hr/week
·	Engineering: onversion factors, Conversion equation analysis, Chemical equations.	n, Temperature, Pressure, (24 hrs)

2- Material Balance Calculation:

Without and with chemical reaction, Recycle, By pass and purge calculations for steady state, Combustion Calculation.

-(30 hrs)

3-Ideal Gas laws:

Definitions of Ideal gas, Real gas, Ideal gases mixture, Real gas mixture.

(9 hrs)

4- Energy Balance:

General energy balance, Enthalpy, Heat capacities their predications and variation with temperature, Heat effects.

(18 hrs)

5-Simultaneous, Mass and Energy Balance:

(9 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus

CE 221

Chemistry

Units 8

Theoretical 3 hr/week
Tutorial 1 hr/week

Practical 2 hr/week

Part -1- Analytical Chemistry

1- Introduction:

Atomic weight, Molecular formula, Chemical equations, Mole concept, and Chemical equilibrium.

(4 hrs)

2- Solution:

Definition, Preparation and properties, Molarity, Normality, Formality, PH, POH, Solubility

(8 hrs)

3- Analytical Methods of Analysis:

a-Qualitative Analysis

b-Quantitative Analysis.

- i. Volumetric (titrimetric) & analysis,
 Acid- base, Redox, Precipitation, Complex titration, Methods of calculation,
 Titration curves
- Gravimetric Analysis
 Precipitation reactions, Direct and indirect methods of analysis, Ksp.
- iii. instrumental Methods of Analysis
 Photometric,, Colourimetic, Potentiometric titration,, PH-measurements
 Chromatography, Atomic absorption.

(33 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 221	Chemistry	Units 8 Theoretical 3 hr/week Tutorial 1 hr/week Practical 2 hr/week
Part -2- Organic Chem	istry	·
Introduction:	tion of avania compounds	· ,
Definition and classification	tion of organic compounds.	(3 hrs)
Aliphatic Compounds:	•	(3)
	on, Properties, Reactions.	•
a) Alkanes		
b) Alkenes		••
c) Alkynes		
 d) Aliphatic Derivative Esters Carboxylic Mercaptans, Thiophe 	ves: Alkyl halides Alcohol's, Eth acids, Amines, Mercaptans, Ca nes Disumbide	ers, Aldehydes, Ketones, arboxylic acids, Amines.
wiereaptans, imophe	nes, Discipline.	(20 hrs)
Aromatic Compounds:		
Structural formula of berreactions; Nitration, Su	nzene ring, Nomenclature, Preparat lphonation, Halogenation, Alkylati chemical reactions of – Toluene	on, Benzene- homologues
,,,		(10 hrs)

4- Heterocylic Compounds:

Pyridine, Furan, Pyrrole.

(3 hrs)

5- Introduction to Polymer and Carbohydrate:

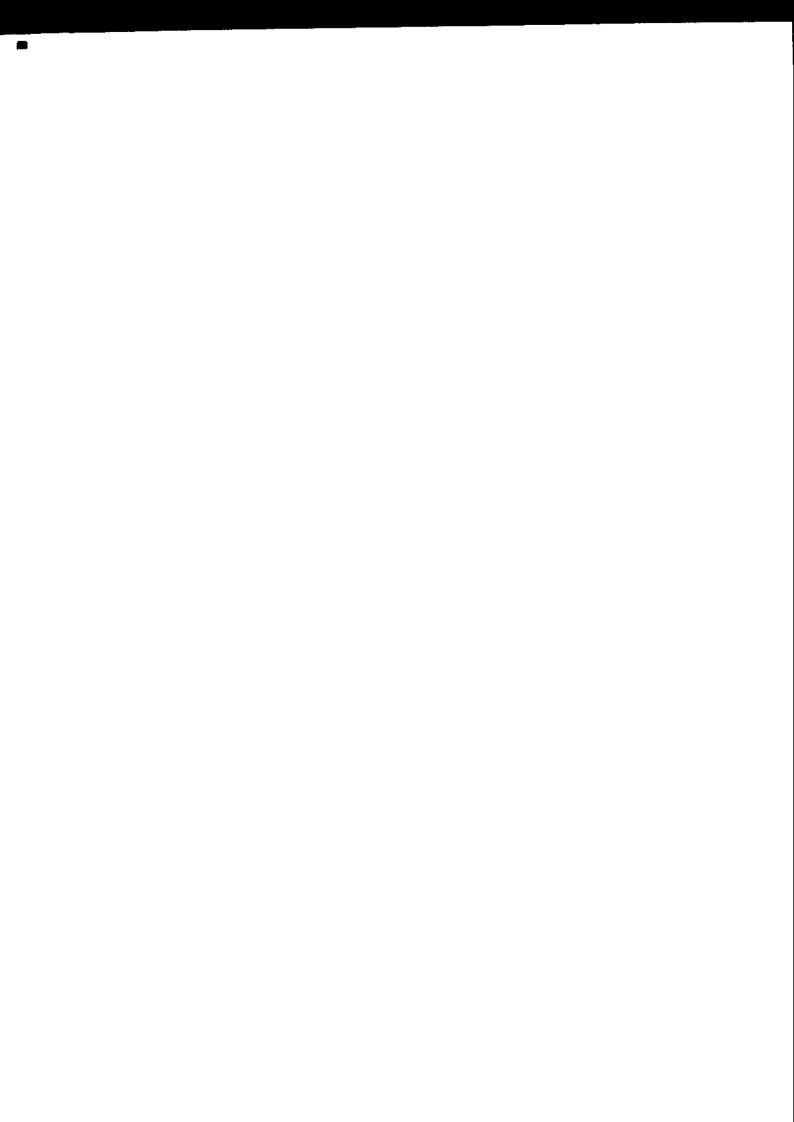
(5 hrs)

6- Organo-Metallic Compounds:

(4 hrs)

University of Technology	Chemical Engineering Department	First Y	ear B.Sc.	Syllabus
CE 131	Engineering Drawing and AutoCAD		Units Theoretica Tutorial Practical	4 al 1 hr/week - hr/week 2 hr/week
Part -1- Engineering 1	Drawing			
	a bout engineering drawing and i	its imp	ortance,	Drawing
instruments, Types of fine	s used in drawings, Hidden lines.			(2 hrs)
2- Geometrical construct 3- Pictorial Drawings:	tion:	· ·		(2 hrs)
(Isometric and Oblique)			÷	(2 hrs)
4- Projection Drawings: First and Third Angle P	rojection.		•	(4 hrs)
5- The Finding of 3 rd vi	ew:			(6 hrs)
6- Sectioning:	·			(6 hrs)
7- Fasteners:				(2 hrs)
8- Assembly and detail d	rawing:			(4 hva)

(4 hrs)



University of Technology Chemical Engineering Department

First Year B.Sc. Syllabus

CE 131

Engineering Drawing and AutoCAD

Units

Theoretical 1 hr/week

Tutorial - hr/week Practical 2 hr/week

Part -2- AutoCAD

1- Definition of AutoCAD programs, its applications, and explain the connection between its program and engineering drawing.

- 2- Explain in details the command of AutoCAD drawing (line, circle, ...atc.), and giving the applications examinations.
- 3- Explain the drawing of projections and sections (2D) in engineering shape, using the command of Layer and Dimensions.
- 4- Giving a simple idea for three dimensions drawing (3D).
- 5- Forming a simple shape for three dimensions drawing (3D).
- 6- Explain the method of forming for three dimensions shape, and giving the applications examinations.

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 231 <u>Me</u>	echanics and Strength of Materials	Units 4 Theoretical 2 hr/week Tutorial 1 hr/week Practical - hr/week
A-Mechanics		
1- Principles of static's:		(2.1
2- Resultants of Force syst	ems:	(2 hrs)
		(5 hrs)
3- Equilibrium of force sys	stems:	(5 hrs)
1- Friction:		
5- Centurions and Centers	s of Gravity:	(6 hrs)
· -		(5 hrs)
6- Moment of Inertia:		(4 hrs)
7- Principle of Movement:		(3 hrs)
		(5 1118)
B-Strength of Materials		
l- Internal Forces in Rigid	Radies	
t- internal Porces in Rigid	Douits.	(1 hrs)
2- Definition of stress and Types of stresses and stra		
	•	(1 hrs)
8- Stress- Strain Diagrams Stress- strain diagrams fo	: r ductile and brittle materials	
		(2 hrs)
Froportional limits: Elastic limit, stiffness elastic	sticity, plasticity, Toughness, Resili	ience, Hardness,
Working stress.	, , , , , , , , , , , , , , , , , , ,	
		(2 hrs)

University of Tech	nology Chemical Engineering Department	First Year B.Sc. Syllabus
CE 231	Mechanics and Strength of Materials	Units 4 Theoretical 2 hr/week Tutorial 1 hr/week Practical - hr/week
5- Hook law:		(2 hrs)
6- Poisson's ratio,	Composite stresses:	(4 hrs)
7- Volumetric stre		(4 hrs)
8- Thin walled Cyl		(4 hrs)
9- Thermal stress:	<u>.</u>	(4 hrs)
10- shear and Mom Deflection	nents in Beam:	
		(4 hrs)
11- Wind Effect on	the Tower high Tower:	(2 hrs)

•	University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
	CE 331	Electrical Technology	Units 2 Theoretical 1 hr/week Tutorial 1 hr/week Practical - hr/week
1	- Semiconductors Equip	ment:	(3 hrs)
	- Rectifiers and Detector	<u>s:</u>	(2 hrs)
	- Electronic Amplifiers: - D.C. Circuits:		(4 hrs)
1	5- D.C. Generators and M	lotors:	(2 hrs)
<u> </u>	6- A.C Circuits:		(3 hrs)
1 2	7- Polyphases Circuits:		(2 hrs)
	3- Transformers and Indu	iction Motors:	(3 hrs)
1)- <u>Starters</u> :)- Integrated Circuits, Me	easuring Instruments, Tranducer	(2 hrs) s, Transmitter:
_	- Electrical Heating App		(4 hrs)
	·		(2 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc.	Syllabus
CE 321	Computer Science	Units Theoretical Tutorial Practical	6 2 hr/week - hr/week 2 hr/week
I- Introduction: Component (hard ware), co	mputer work, computer, composition	on, Application	(2 hrs)
, 12	lature, MSDos, External and interna	al commends.	(4 hrs)
	iter-icons, close window, stand by.		(2 hrs
4- Folders: Size and cascade, window file or folder copy file or fo	ws folder construction, construction lder.	choose file or fo	
•	windows, format & Test Floppy di	isk, Arrange Icons	(2 hrs) s, Run,
Help. 5- Word Program:			(2 hrs)
File commands (New,	Open, Close, Save, Save as, Page Paste, Clear, Select All, Find, Repl		o typing (2 hrs
7- Menu Table:	nd footnote, Tool bar Insert (Picture	e, Footnote, Syml	

Cells, Select Row, Select column) (Table Autoformat Sort, Formula).

(6 hrs)

8- Program Excel:

Open, Taskbar, Toolbar, Format, Sheet, Cell, Construction of sheet, Open file, Clear (Delete), Save (Save as), Close, cascade, Copy and Paste, Add, Delete, Find, Replace, Auto summation, Insert, equation, Format sheet, Auto format, Print, Print preview, Draw.

(12 hrs)

First Year B.Sc. Syllabus

CE 321

Computer Science

Units

Theoretical 2 hr/week
Tutorial - hr/week

Practical 2 hr/week

9- Power Point:

Definition

FILE: New, Open, Open, Close, Save, Save as, Page set up, Print, Exit)

EDIT: Undo, Repeat, Cut, Copy, Paste, Select All, clear Duplicate, Delete slide,

find, Replace

FORMAT: Font, Alignment, Text Direction slide, Layout, Background, Apply Slid Show, View show, Rehearse timing design set up show, present Animation, Animation Preview, Slide Transition).

(11 hrs)

10- Auto Cad:

Introduction, define screen draw, select point in the screen Limits, Status, Circle, Erase, Line, Color, Point, zoom, Redraw, Arc, Ellipse, Polygon, Donut, Solid, Fill Break, Copy, More, Pan, Mirror Trim, Extend, Charge, Line type Ltscale, Dis Area, Trace, Hatch text, Snap, Grid, Rotate, Cops, Ortho, Fillet, Dim Layer.

(14 hrs)

11- Internet:

(6 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
		Units 4
CE 431	Workshop	Theoretical - hr/week Tutorial - hr/week

Practical

6 hr/week

Refer to Centralize Curriculum

University of Technol	ogy Chemical Engineering Department	Second Year B.Sc. Syllabus
		Units 4
CE 112	Human Rights	Theoretical 2 hr/week
		Tutorial - hr/week

Standard Syllabus

		Units	4
CE 122	<u>Mathematics</u>	Theoretical	2 hr/week
		Tutorial	1 hr/week

Chemical Engineering Department

1. Partial Differentiation:

University of Technology

Function of two or more variables, limits and continuity, partial derivatives, chain rule, Gradients, Directional derivatives, Higher order derivatives, Maxima, minimum and saddle points, Lagrange multipliers.

(10hrs)

- hr/week

Second Year B.Sc. Syllabus

Practical

2. Multiple integrals:

Double integrals, Area, Triple integrals in rectangular coordinates, Physical application of double and triple integration.

(9hrs)

3. Ordinary Differential Equations:

Solution of first order ordinary differential equations, Solution of second order ordinary differential equations, and higher order differential equation.

(9hrs)

4. Infinite Series:

Power series of functions, Taylor's theorem, Integration, Differentiation, Fouries series, Even and odd functions, Half range expansion, Periodic functions.

(8hrs)

5. Function and Definite Integrals:

The error function, The gamma function, The beta function, factorial function. (6hrs)

6. Vector Analysis:

Products of three vectors, Equation of lines and planes.

(5hrs)

7. Complex Algebra:

Continuity, Derivation of Complex Variables, Analytic function, Integration of functions of complex variables and cauchy's theorem.

(5hrs)

8. Matrices:

The Matrix, Matrix algebra, The transpose of matrix, The inverse of matrix, Eigen Values and Eigen vectors, solution of linear equation by matrix.

(8hrs)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
·	•	

CE 142 <u>Fluid Flow</u>

Theoretical 3 hr/week
Tutorial 1 hr/week
Practical 2 hr/week

Units

1. Introduction:

Physical properties of fluid, Definition of type fluid: Newtonian's, non-Newtonian, Incompressible, compressible fluid, static, Dynamic fluid, shear stresses in fluid.

(4hrs)

2. <u>Dimensional- Analysis:</u>

Rayleigh's method (Power series), Buckingham's ∏-theorem.

(6hrs)

3. Fluid- Dynamic:

Fluid dynamic system, Boundary layer, Continuity equation, Momentum equation, Bernoulli's equation, Euler's equation, Application of Bernoulli's equation and continuity equation.

(10hrs)

4. Newtonian's Fluid-(Incompressible Fluid):

Laminar and Turbulent flow, Reynolds's number, Friction losses and Pressure drop in pipes and Fitting, Determine of pipe diameter, Flow in Branch pipes, Friction losses in non-circular pipes, velocity Distribution.

(10hrs)

5. Non- Newtonian Fluid:

Definition, Type of fluid depend on time, Calculation of friction and pressure drop for general time independent in laminar and Turbulent flow.

(6hrs)

6. Compressible Fluid in pipes:

Pressure wave, Mach number, General equation, General equation for isothermal and adiabatic condition, work for isothermal and adiabatic of Compressor.

(10hrs)

7. Flow-Measurement:

Pilot-tube, Orifice meter, Venturi meter, Rotameters, Notch air Weirs, Special Flowerets (Hot wire anemometer, Gas meter).

(12hrs)

8. Pumping of Liquids:

Calculation of Total head, NPSH, Performance characteristics curve, Calculation of horsepower, equipment, and cost.

(8hrs)

9. Flow of Fluid Through Granular Beds:

Beds & Packed Column, Fixed beds, Fluidized Bed.

(8hrs)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 142	Fluid Flow	Units 8 Theoretical 3 hr/week Tutorial 1 hr/week Practical 2 hr/week
10. Mixing of liquids in Mixing, Mixers groups, Power consump	type, Small blade area, large	
11. Two- Phase Flow a	nd Pressure drop in pipes :	(6hrs)
(4hrs		
12. Introduction in uns	steady Flow:	(6hrs)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 132	Properties of Materials	Units 3 Theoretical 1 hr/week Tutorial 1 hr/week • Practical 2 hr/week
1. Engineering require	ment for materials:	-1. distantista vorma

Mechanical Properties, stress-strain curve, elasticity, plasticity, ductility, young modulus, tensile stress, yield stress, bricking stress, true and engineering stress-strain diagram.

(4hrs)

2. Atoms and atomic coordination:

Ionic bond, inter-atomic distance attraction forces between atoms, coordination number, covalent bond, Metallic bond.

(4hrs)

3. Crystal structure:

Crystal structure, unit cell, types of unit cells simple cubic, Face centered cubic, body centered cubic, atomic packing factor, Previous lattice, Miller index, crystalline Planes, X-ray diffraction.

(4hrs)

4. Atomic disorder in solid:

impurities in solids, solid solutions in metals, Imperfection incrystals, point defect, line dislocation, surfaces + grain boundaries.

(4hrs)

5. Atomic movement in solid state:

Atomic diffusion, self-diffusion, diffusivity Fick's first law, diffusion constant, diffusivities versus temperature, Factors effect diffusivity.

(4hrs)

6. Phase diagrams:

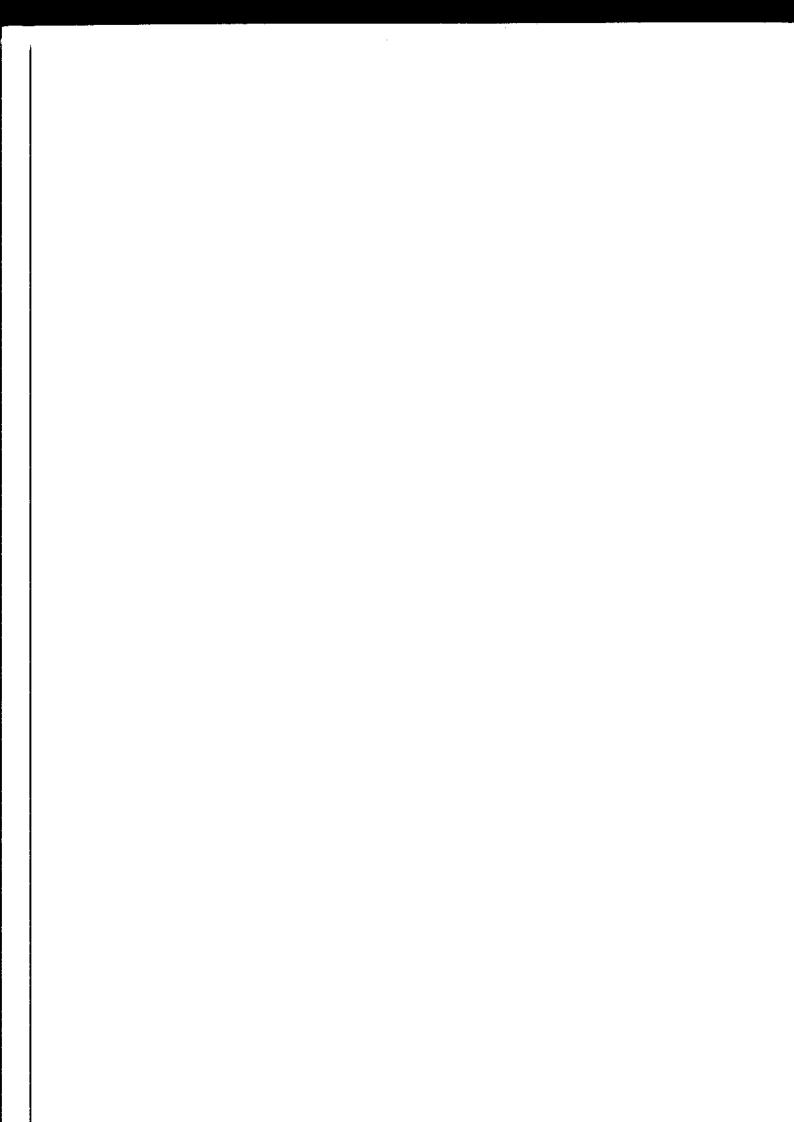
Definition of alloys, binary alloys, phase diagrams (equilibrium thermal diagrams), eutectic; solid solution and combination type diagram, Iron-carbon face diagram, allotrope Trans formation (polymorphism).

(4hrs)

7. Corrosion:

Definition, why it happen, Type of corrosion, Form of corrosion, corrosion reactions (anode + cathode) corrosion rate measurement, methods of prevention.

(4hrs)



University of Technolog	y Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 222	Computer Programming & MatLA	Units 6 Theoretical 2 hr/week Tutorial - hr/week Practical 2 hr/week
Part -1- Visual Basic	<u>e</u>	
1. <u>Introduction to Vi</u> Open, Save File A	isual Basic: s Standard EXE file.	
2. Menus:		(2hr)
Form window,	Project Explorer window, Propertie box window, Debug window.	
3. Using Toolbox:		(4hr)
Simple examples of	on using toolbox items.	(4hr)
	and string data, variables, Assignme	ent statement, mathematical
and logic operations.	•	(4hr)
5. Arithmetic operat	ors:	(2hm)
	nent, multiple If Then statement, I	(2hr) If Then else, multiple
IfThen Else, Se	elect case statement.	(4hr)
7. Messages: Msgbox, InputBox		
8 Loon.		(4hr)
_	Do while Loop, Do Loop Whi	ile, Exit For , Exit Loop. (4hr)
9. Array: Array definition, operations on Arrays.	Declaring Arrays, Using for next Lo	op with Arrays, Arrays and
,		(6hr)

• <u>Database:</u>
Working with database, Adding and deleting records, Displaying data.

(8hr)

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University of Technology	Chemical Engineering Department	Second Yea	r B.Sc. Syllab <u>us</u>
CE 222	Computer Programming & MatL	<u>AB</u>	Units 6 Theoretical 2 hr/week Tutorial - hr/week Practical 2 hr/week
Part -2- MatLab			
1. Introduction to Ma	tlab:		
M-files, getting help	o. ·		(2hr)
2. Algebra:			
2.1 <u>Variable Names</u>		1	مسمعما المالة
Variable names must st	tart with a letter and the intial letter	can be folio	owed by letters (1hr)
2.2 Basic Arithmeti	<u>c</u>		•
Addition +, Subtraction	n -, Multiplication *, Division /, Ex	ponentiation	
		· ·	(1hr)
2.3 <u>Functions</u>			
- abs: absolute value ab			
- sqrt: square root sqrt(•		
_	in an array max([132,129,66,120]) t in an array min([132,129,66,120])	·.	•
- factorial: factorial fun			
	n) to an integer round(3.5)		
- floor: round down to	The state of the s		
- ceil: round up to an ir	- ' '		
- exp: exponential exp(- , , ,		
- log: natural logarithm	log(exp(2))		
- log10: logarithm to ba	ase 10 log10(100)		
			(2hr)
2.4 Complex Numb	· · · · · · · · · · · · · · · · · · ·		
MATLAB can also har	ndle complex numbers		
2.5 <u>Loops</u>			241
2 Cabin F4			(1hr)
3. Solving Equations:	you can use the command salve		
3.1 Representing a 1	you can use the command solve.		
	, use the command inline.		
1 0 1 - P1 - V1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	, ase the communication in the		(1hr)
3.2 <u>Differentiation</u>			(
	ion, we use the command diff		
	•		(1hr)
•			

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 222	Computer Programming	Units 6 Theoretical 2 hr/week Tutorial - hr/week Practical 2 hr/week
3.3 <u>Integration</u> We can use MATLAI command int.	B for computing both definite and i	
3.4 <u>Differential Equ</u> We can use MATLA symbolic solution is or represented by Dy.	ations B to solve differential equations. T dsolve. For that command, the d	(1hr) The command for finding the erivative of the function <i>y</i> is
represented by Dy.		(1hr)
3.5 <u>Multiple integra</u>	<u>ls</u>	(1hr)
3.6 <u>Limits</u> You can use limit to co	ompute limits, left and right limits a	as well as infinite limits.
4. Graphics: 4.1 Two-Dimension	nal Plots	
	aph a function is to use the comma	nd ezplot (2hr)
	-Dimensional Space	
	ving 3D curves is ezplot3.	(2hr)
spaces inside the squar	tor by typing a list of numbers se	parated by either commas or
dot (dot product)cross (cross product)norm (norm or length	of a vector)	(21)
5.2 Matrices		(2hr)
A+B (matrix additionA-B (matrix subtractionc*A (scalar multiplic	ion)	
- A*B (matrix multipli- A' (conjugate transpo	cation) se of A)	
det(A) (determinant of ank(A) (rank of A)	of square matrix A)	
		(2hr)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
		Units 3
CE 342	Fuels Technology	Theoretical 1 hr/week
		Tutorial 1 hr/week • Practical 2 hr/week
different Product. 2. Properties of oil stor	eks and prepare fractions oil:	(3hr)
	al processes which applied from	n the well to the different
	ation and the processes on fraction	
		(4hr)
3. Petroleum Fractions	s properties :	
Benzene, Kerosene,	Jet Fuel, Diesel, lubricants oil Asp	halt, Fuel oil.
		(9hr)

4. Gaseous Fuels and Liquefied gases:

Energy source types gases fuel types and liquefied gases which source of energy in industrial and houses.

(4hr)

5. Combustion Calculation:

Different type of equations about the subject and from it found the component which burner and the combustion equation and guess the flue gases and study behavior of combustion.

(6hr)

6. Burner of gases fuel:

The burner types of gases fuel and liquid fuel properties arrange it in Furnace.

(2hr)

7. Combustion in Furnaces:

Furnace types in oil industrial, Boilers and steam production that used in different industrial.

(2hr)

Note:

• 1st term

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 242	<u>Thermodynamics</u>	Units 4 Theoretical 2 hr/week Tutorial 1 hr/week Practical - hr/week
1. Introduction: Fundamental quantit	ies Time, Length, Mass, Force, Te	mperature.

2. First Law and Other Basic Concepts:

Joules experiments, Internal Energy Formulation of the first law, Thermodynamics state and state functions Enthalpy, The steady state flow processes, The reversible processes, Heat Capacity and specific Heat, Equilibrium, the phase rule.

(4hrs)

(2hrs)

3. <u>Applications of the Equations of Ideal Gases, Generalized Correlation and the Acentric Factor</u>.

(8hrs)

4. Heat Effects:

Heat capacities of gases as a function of temperature, solids and liquids, Heat change accompanying phase change, Heat of industrial reactions.

(6hrs)

5. The Second Law of Thermodynamics:

Heat engine, Entropy, Second law limitation and real process Entropy change and irreversibility and probability (statistical thermodynamics), Third law of thermodynamics.

(6hrs)

6. Thermodynamics Properties of Fluids:

Relationships among the thermodynamics properties (including Helomtz and Gibes free energies and chemical potential), Steam formation and two phase system, Saturated temperature and pressure, Triple point, Wet vapor and dryness fraction, Types of thermodynamic tables and diagrams, Steam power plant cycle and analysis, Barometric condenser, Metering and throttling processes, Steam and as turbines.

(10hrs)

7. Refrigeration and Liquefaction:

Refrigeration cycles (Carnot, Air, Vapor – Compression) and Comparisons, Choice of refrigerant, Absorption Refrigeration, The heat pump, Liquefaction process.

(10hrs)

	•		
		Units	4
CE 242		Thermodynamics Theoret	ical 2 hr/week = 1 hr/week

Chemical Engineering Department

8. Phase Equilibrium:

University of Technology

The nature and criteria of equilibrium, Binary system, Vapor pressure of an ideal solution and non ideal solutions, Henrys law, Activity and activity coefficients, Flash separation calculations.

(8hrs)

Second Year B.Sc. Syllabus

9. Chemical Reaction Equilibrium:

Thermodynamics of ideal gases and mixtures, Derivation of the general equilibrium expression, Chemical equilibrium of ideal and non-ideal gases, Reaction equilibrium in solution, effect of temperature on chemical equilibrium.

(6hrs)

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		· <u> </u>	· .		· 		
	•					Units	6
CE 232		<u>Ph</u>	ysical Che	<u>mistry</u>		Theoretical	2 hr/week
						Tutorial	l hr/weck

Chemical Engineering Department

1. Change of State:

University of Technology

One component systems: Water, carbon dioxide and helium, Vapor pressure and external pressure, Liquid-vapor composition diagram, Liquid-Liquid binary solution mixture, Roulette's law for ideal mixtures, Non-ideal Liquid mixtures, The principle of distillation, Henry's law, solubility of gases in Liquids.

(10hrs)

Practical 2 hr/week

Second Year B.Sc. Syllabus

2. Surface Chemistry:

Pressure difference across curved surfaces, Surface tension, and capillary rise, Langmuir adsorption isotherm and langmuir theory of adsorption, Calculations of surface area of the adsorbent, BET equation for surface area calculation.

(12hrs)

3. Chemical Kinetics:

The rate of chemical reactions, order of reaction and rate constant, zero, first, second and third order rate equations, Reversible reactions, Consecutive reactions, parallel reactions, Determination of the order, Reactions in flow systems, Effect of temperature on reaction rate, The transition-state theory.

(16hrs)

4. Catalysis:

Reaction of catalyst in homogeneity system, Enzyme reactions and Kinetic of reactions

(6hrs)

5. Electrochemistry:

Conductivity measurements, Diffusion and ionic mobilities, Activity and ionic strength, Determination of activity coefficient from solubility, The Debye-Hackle theory, Acid-base catalysis and their dissociation constant.

(6hrs)

6. Electrochemical Cells:

Electromotive force (EMF) of a cell, Measurements of EMF- the potentiometer, The polarity of electrodes, The cell reactions and reversible cells, Free energy and reversible cells, Types of half cells and classification EMF, Standard electrode potentials, Standard free energy and entropy of aqueous ions, Calculation of EMF of a cell, Oxidation-reduction reactions, Concentrations cells, Electrolysis, Corrosion. Kinetic of reactions.

(8hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
		Units 4
CE 113	<u>Democracy</u>	Theoretical 2 hr/week
		Tutorial - hr/week
		Practical - hr/week

Standard Syllabus

University of Technology	Chemical Engineering Department	Third Year B.Sc. S	<u>yllabus</u>
<u>, , , , , , , , , , , , , , , , , , , </u>		Units 6	
CE 133	Applied Mathematics	Theoretical 3	hr/week
•		Tutorial 1	hr/week

1. Revision ordinary differential equations:

First and second order, simultaneous differential equations, Application for chemical engineering.

(18hrs)

2. Solution of differential equations by series:

Simple series, method of Frobeins, Bassel's equation. Application for chemical engineering.

(18hrs)

3. The Laplace transformation:

The Laplace transformer, Properties of the Laplace transformation. The inverse transformation. Inversionby partial fraction. Convolution. Solution of differential equations. The transforms of special functions. Step function, staircase function, periodic function, Trigonometric function.

(18hrs)

4. Partial Differential Equations:

Formulating of partial differential equations. Solution of partial differential equations. Method of separating variables and Laplace transformation method.

(18hrs)

5. Numerical Analysis:

The difference operators. Interpolation. Finite difference equations. Solution of first and second order differential equations.

(18hrs)

l	University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
		<u></u>	
			Units 7
	CE 143	Mass Transfer	Theoretical 3 hr/week
			Tutorial 1 hr/week
	•		Practical - 7 hr/week

1. Introduction:

(4hrs)

2. Mass transfer fundamentals:

Steady state mass transfer. Diffusion in binary and multicompouent gases. Diffusion in binary and multicompouent liquids. Connective mass transfer. Mass transfer coefficients. Two-film theory. Penetration theory. Unsteady state mass transfer methods

(25hrs)

3. Methods of gas-liquid separation (absorption and stripping):

Determination of interfacial mole fractions. Individual and overall mass transfer coefficients. Effect of gas solubility on equilibrium curve. Single stage absorption, multistage absorption. Finding number of stages graphically and through use of correlations for absorption and stripping. Packed columns limits of operation. Design of packed beds for dilute and concentrated solutions. Finding number of transfer units. Absorption with chemical reaction.

(16hrs)

4. Binary distillation:

Introduction, (T-x,y) and (x,y) equilibrium curves. Flash or equilibrium distillation for binary and multicomponent mixtures. Distillation with reflux using Mckabe-Thiele. Sorel method to find number of stages. Fenske equation to find minimum reflux ratio and minimum theoretical stages. Column efficiency. Thermal method for finding number of ideal stages (Ponchoue-Savarit Method). Multicomponent differential distillation.

(25hrs)

5. Multicompouent distillation:

Dew and bubble points calculation methods. Multicomponent equilibrium relationships. Minimum reflux and minimum number of stages. Short-cut method. Checking validity of reflux ratio used. Operating limits of distillation columns.

(14hrs)

6. Mechanical separation process and size reduction:

Classification of equipment. Particle shape factor. Power requirements in crushers. Particle size analysis.

(6hrs)

Note:

• 1st term

University of Technology Chemical Engineering	Third Year B.Sc. Syllabus
CE 233 Statistics and Instrumen	Units 4 Theoretical 2 hr/week Tutorial - hr/week Practical - hr/week
Part A: Statistics	
1.Introduction: Qualitative and conclusive statistics, popu of data.	
2 Toble and graphical representations	(4hrs)
2. <u>Table and graphical representation:</u> Primary data, repeatability, normal distribu	tion.
	(4hrs)
3. Media determination:	(4hrs)
4. Scatter determination:	(into)
Standard and relative deviations.	(41)
5. Standard distributions:	(4hrs)
S. Standard distributions.	(8hrs)
6. Chi distribution:	
Confidence limits, Chi test, degree of fro	eeaom. (4hrs)
7. Curve fitting:	
Least square method, linear relations,	, exponential equations, gradient, and
correlation coefficient.	(4hrs)
	(1442)
Part B: Measurements	
1. Measurements:	
Absolute and relative. Environmental	conditions. Application on physical
properties.	(4hrs)
2.Errors:	(1113)
Types of errors, determination of errors.	(01)
3.Instruments:	(8hrs)
For temperature, Pressure, level, composition	on, velocity, flow rate.
	(16hrs)
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University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
	· · · · · · · · · · · · · · · · · · ·	Units 6
CE 243	Reactor Design	Theoretical 3 hr/week Tutorial 1 hr/week Practical - hr/week
1. Introduction:		
Stoichiometry, rea	action rate constant, order of reaction, yield, selectivity.	ction, reversible and non-
		(9hrs)
2. Types of reactors:		
Batch reactors, continu	uous flow reactors i.e. CSTR, PFR.	
3. Bach reactors:		(9hrs)
	eaction time, maximum producti	on rate, non-isothermal
operation, adiabatic oper		(18hrs)
4. Tubular reactors:	•	· · · · · · · · · · · · · · · · · · ·
- · · -	isothermal and non-isothermal or of tubes, pressure drop, space velo	•
	71	(18hrs)
5. Continuous stirred ta		
	for isothermal operation, scale-up of STR's in parallel and in series.	of liquid-phase batch data
_		(15hrs)
6. Unsteady state operation		
Start up of CSTR,	semi-batch reactor.	
77 N.T. 142-1		(9hrs)
7. Multiple reactions: Operation in para maximum amount of desi	llel and in series for single and m	ultiple reactions to obtain
maximum amount or ges.	red product.	(6hrs)
8. Comparison between	reactors:	(3.110)

8. Comparison between reactors:

Comparison between batch, tubular, and CSTR for single and multiple reactions

(6hrs)

University of Technology	Chemical Engineering Department	Inird Year B.Sc. Synabus
	Well the state of	
		Units 7
CE 343	Heat Transfer	Theoretical 3 hr/week
	· · · · · · · · · · · · · · · · · · ·	Tutorial 1 hr/week
<u>.</u>		 Practical 2 hr/week
		- · · · · · · · · · · · · · · · · · · ·

1. Modes of heat transfer:

Material properties of importance in heat transfer.

(2hrs)

2. Steady state heat conduction in one dimension:

Plane wall, radial systems, heat source systems. Boundary surrounded by fluids. Overall heat transfer coefficient. Extended surface, conduction-convection systems, fins.

(24hrs)

3. Principles of convection:

Transport equations. Fluid mechanism aspect of convection. Laminar boundary layer. Thermal boundary layer. Empirical and practical relations for pipe and tube flow. Flow normal to single and tube banks.

(18hrs)

4. Heat exchangers:

Various types and their general characteristics. Fouling factor. Heat exchanger mean temperature differences. Co-current and counter-current flow.

(12hrs)

5. Shell and Tube Exchangers:

Types and various specifications. Design calculations by conventional and by effectiveness (NTU) methods. Optimum design calculation.

(6hrs)

6. Condensation and Boiling Heat Transfer:

Condensation of single vapors. Design calculations for condenser, condensersubcooler and superheater condenser.

(9hrs)

7. Radiation and furnace design:

Radiation properties, shape factor, heat exchange for non black bodies, parallel planes, shields, gas tradition, boiler.

(9hrs)

8. Unsteady state heat transfer:

Temperature as a function of time, lumped capacity system, quenching of small bodies, heating of tank reactor.

(10hrs)

Note:

• 2nd term

University of Technology	Chemical Engineering Department	Imru Tear B.Sc.	Synanus
CE 443 <u>E</u>	quipment Design using Computers	Units Theoretical Tutorial	- hr/week

Part (A)

1. Process Planning:

Scheduling and flow sheet design. Flow sheet types and designation. Block diagram. Process flow sheet. Piping and instruments diagram. Utility flow sheet. Equipment layout and plot plan. Project evaluation. Feasibility studies. Site selection. Project cost estimation.

(6hrs)

2. Piping networks and pumps:

Pipe fittings, valves. Steam traps. Piping design standards. Piping material and selection. Pump specifications and data sheets. Optimum diameter. Types of pumps. Hydraulic characteristics for centrifugal pumps.

(6hrs)

3. Vessels and Tanks:

Types of vessels. Flash drums. LPG tanks. Criteria in vessel design. Stress considerations. Design of tall vertical vessel. Design of pressure vessels. Supports and foundations

(6hrs)

4. Heat transfer equipment:

Types of exchangers and applications. Exchanger ratings. Exchanger design. Exchanger specification sheets. Furnaces. Convection and radiation zones. Types of fuels. Burners and arrangements. Steam boilers, types of boilers.

(6hrs)

5. Mass Transfer Equipment:

Types of columns. Plate and packed Types of plates and packing. Design features. Pressure drops in columns.

(6hrs)

University of Technology	Chemical Engineering Department	Third Y	<u>ear B.Sc</u>	<u>. Syllabus</u>	
·		·····			
			Units	6	
CE 443	Equipment Design using Computers		Theoretica Tutorial	al 2 hr/week - hr/week	

Part (B): Complete equipment design of each of the following units:

1. Pressure vessels, pumps and compressors:

Flash drum. Gas-liquid separator, LPG storage tanks. Cyclone separators. Gas movers and compressors. Pumps. Piping network. etc.

(8hrs)

Practical 1 hr/week

2. Heat Equipment:

Shell and tube heat exchangers. Plate heat exchanger. Boilers. Furnaces. Vaporizers. Air coolers. Coil type heaters. Condensers. etc.

(10hrs)

3. Mass transfer equipment and reactors:

Distillation column. Absorber. Extractor. Dryer. Absorption towers. Leaching equipment. Scrubbers. Stripping towers. Reactors.

(12hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
	· · · · · · ·	Units 2
CE 543-U	Biochemical Engineering	Theoretical 2 hr/week
		Tutorial 1 hr/week
	Unit Operations Branch	Practical - hr/week
1. Definition, introducti	ion:	
		(2hrs)
2 Industrial micro orga	anisms and methods of separation	
	might and methods of separation	
4 3.51		(2hrs)
3. Micro organisms gen	etics:	
	•	(4hrs)
4. Biotechnological met	hods of growths:	
Liquid systems, b		Breaded cultures Solid
<u> </u>	atch and continuous cultures.	Dieaded Cultules Solid
Fermentation's		
		(6hrs)
5. Single cells proteins:		
	d, treatment of waste, production	of single cell protein Diet
	i, ireadificial of waste, production	of single cen protein. Diet
value	•	
		(3hrs)
6. Bakery yeast:		
Dry and ready effe	ective types.	
	J. C.	(3hrs)
7 Amino saida.		(31118)
7. Amino acids:		
Clotamic acid.		
		(1hr)
8. Organic acids:		
Acetic acid.	•	
Acetic acid.		/11 \
		(1hr)
9. Enzymes:		
		(1hr)
10. Antibiotics:		,
	uging antihistica Danigillin	
wheroorganisms produ	ucing antibiotics, Penicillin	
		(1hr)
11. <u>Industrial Alcohol:</u>	·	
Alcoholic fermentatio	n, microorganism used, production	method.
	, 2	(1hr)
12 Animal call cultivation		(1111)
12. Animal cell cultivation	<u>on:</u>	
•		(2hrs)
13. Plant cell Cultivation	<u>n:</u>	
•		(1hr)
14. Environmental clear	ning using biotechnology:	()
· · · · · · · · · · · · · · · · · · ·	er treatment. Organic Pollutants. Pe	etroleum wostes
in the netus of wat	er a caunem. Organic Politicants. F	
		(2hrs)
		•

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
<u> </u>		Units 2
CE 643-U	Environmental Engineering	Theoretical 2 hr/week
·		Tutorial 1 hr/week
•	Unit Operations Branch	Practical - hr/week

1. Introduction:

What is pollution and a historical brief on international concern with pollution. Types of pollutants and measurements. Pollution allowable limits in the environment. (2hrs)

2. Cleaning of air and industrial gases:

Sources and effects of air pollution. Methods of control. Designing pollution equipment. Settling chamber, Cyclone separator, Venture scrubber, spray tower, cyclone scrubber, packed bed, electrostatic precipitator. Economic evaluation of gas cleaning.

(10hrs)

3. Water and heavy water treatment:

Physical methods of treatment. Occupation methods. River water cleaning, heavy water treatment. Desalination.

(10hrs)

4. Industrial and sewage treatment:

Industrial water treatment. Sewage water treatment.

(4hrs)

5. Solid pollutants:

Sources and effects. Methods of control of solid pollutants.

(4hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
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Units CE 543-R Storage and Transport of Crude and Petroleum Products Theoretical 2 hr/week 1 hr/week Tutorial Refinery Engineering Branch

Practical - hr/week

1. Fluid Flow:

Petroleum liquids flow. Petroleum gases flow. Designing pipe network from optimum velocity and diameter viewpoint. Crude and petroleum products pipelines, Heating and pumping stations, Increasing pumping rates by looping. Increasing pumping rates by polymer additives. Two-phase flow. Unsteady state flow and hammering.

(14hrs)

2. Pressure Vessels / Tanks:

Stress systems of shells. Bending and membrane shells. Equilibrium equation for a shell element and LPG tanks, Rotationally symmetrically loaded membrane shell. Crude oil huge tanks.

(8hrs)

3. Selection of Materials for Vessels and Pipelines:

Selection of materials for vessels to limit corrosion. Selection of materials for pipelines to limit corrosion. Catholic protection.

(8hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
		Units 2
CE 643-R	Combustion Engineering	Theoretical 2 hr/week
	Refinery Engineering Branch	Tutorial 1 hr/week Practical - hr/week
1. Combustion reaction	stoichiometry:	
		(8hrs)
2. Excess air and flame	temperature:	
3. Combustion product	<u>s:</u>	·.
4. Gas, solid, and liquid	l fuels:	
5. Flammability limits:	•	
•		(8hrs)
6. Flame and combustion	on speed:	
7. Flame stability:	•	•
8. Gas, liquid and solid	fuels burners:	
	·	(6hrs)
	nd temperature profiles:	
·	ection rooms in furnaces:	
11. Furnaces and the tu		
12. Chimney height cal	culation:	(01,)
12 Eumana officiares	and heat loss salaulations	(8hrs)
	and heat loss calculations:	
14. Furnace wall layers15. Tube layers in furn		
13. Tube layers in Turns	accs.	

University of Technology	Chemical Engineering Department	<u>Fourth</u>	Year B.Sc.	<u>Syllabus</u>
				<u></u>
		<u> </u>	Units	2
CE 144	<u>Project</u>			l - hr/week
			Tutorial	- hr/week

Practical 2 hr/week

1. Abstract:

Brief: conclusion of experimental runs, discussion, empirical correlations and programming computer.

2. First Chapter:

Introduction, the aim of the project.

3. Second Chapter:

Literature Survey, theoretical concepts.

4. Third Chapter:

Experimental works, Equipments and materials, the procedure of experimental and description of computer programming.

5. Fourth Chapter:

Results and Tables.

6. Fiveth Chapter:

Discussion of results by presentation the results in figure relationships.

7. Conclusion and Recommendation:

8. Reference and Appendices:

University of Technology	Chemical Engineering Department	Fourth Y	Year B.Sc.	<u>Syllabus</u>
			Units	3
CE 244	Plant Design		Theoretical Tutorial	1 hr/week - hr/week

1. Introduction:

Literature, Methods of manufacturing, Capacity of the Process Physical and Chemical Properties.

2. Material and Energy Balances:

Review of Material and Energy balances, Flow sheet Symbols, PFD In formation in Flow Diagram.

3. Choice of Process Devices and Equipments:

Pumps, Valves, Pipes, Measuring Devices and Controllers, Material of Construction, PID Plot.

4. Choice of Plant Location and Layout Standards:

5. Industrial Services:

Cooling Tower, Boilers, Study Problems from Industrial Waste Disposals, and How to Prevent the Environment, pollution.

6. Use a Computer Program to Design:

Introducing a Complete Project with Material and Energy Balances Design of Devices, Calculation of Costs, Choice of Location.

Units	7
Theoretical	3 hr/week
Tutorial	
1	Theoretical

Chemical Engineering Department

1. Boundary Layer and analogies:

University of Technology

Boundary Layer in Laminar and turbulent Flow, Boundary Layer in Laminar sub Layer, Transition layer, Velocity distribution on Surfaces and Pipes, Momentum, heat and mass transfer molecular diffusion, Eddy transfer, Reynolds analogy, Modified Reynolds analogy, Chilton and Colburn analogy.

(20 hrs)

Fourth Year B.Sc. Syllabus

2. Evaporation:

Introduction, Types of Evaporations, Evaporation Equipment, Heat transfer in Evaporation Process, single, double and Multi effect Evaporators, Design of evaporators, Comparison of Forward, backward and Parallel effect evaporators, boiling Point rise.

(12 hrs)

3. Drying:

Introduction and general Principle in drying, Rate of drying, the mechanism of moisture movement, Calculation of rate of drying, moisture transport in Solids at Constant and falling rate Period, Capillary movement, Material and Energy Balances in Continuous dryers, Types of Dryers

(8 hrs)

4. Humidification, dehumidification and Cooling towers:

Humidification, dehumidification, Temperature humidity Chart for air – water system. Enthalpy – humidity –temperature chart. Addition of Vapor or liquid Stream to a gas Stream. Mechanism of dehumidification Evaluation of heat and Mass transfer Coefficient, Cooling tower, height of Packing in Cooling towers, Minimum gas Condition, Change of Condition of liquid and gas in Cooling towers.

(14 hrs)

5. Extraction:

Definition, Extraction process, Equilateral Triangular coordinates (Ternary Diagram), system of three liquid one pair partially soluble, system of three liquid two, pairs partially soluble, choice of solvent, Equipment in extraction cross current extraction, multi-stage Cross Current extracting cross current for insoluble Liquid, Continuous Counter current extraction, Continuous Counter Current in Soluble, Liquid, Minimum Solvent, Counter Current extraction with reflux.

(22 hrs)

Note:

• st term

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
		·
		Units 7
CE 344	Transport Phenomena	Theoretical 3 hr/week
		Tutorial 1 hr/week

6. Filtration:

Type of Filters, Filtration theory, Plate and frame filter press, leat filter, filtration at Constant ΔP , Filtration at Constant rate, washing Time.

(10 hrs)

3 hr/week

Practical

7. Crystallization:

Introduction Equilibrium Solubility in crystallization, Classification of Crystallizes, Nucleation theory, Rate of Crystal Growth, Material and Energy balance in Crystallization.

(4 hrs)

University of Technology	Chemical Engineering Department	Fourth '	<u>Year B.Sc.</u>	<u>Syllabus</u>	ĺ
·					
			Units	5	
CE 444	Process Control		Theoretical	2 hr/week	
	• • • • • • • • • • • • • • • • • • • •		Tutorial	1 hr/week	
			D (1) 1	0.1 (1	

1. Revision of Laplace Transformation:

(4 hrs)

2. Process Dynamic and Transient Response of the system:

Dynamic and Control For chemical Process, Dynamic behavior for 1st order system, lineraization, Interacting and non-interacting, Second order system, Transportation.

(16 hrs)

3. Characteristics of the closed loop System:

Over all closed loop transfer functions and block diagram algebra Transient of Simple Closed loops System, Stability of Control, Introduction to the Frequency analysis and design technique.

(20 hrs)

4. Industrial Controller:

Actions selection Criteria for Various Control modes, final Control element, Dynamics and Control of Chemical reactor System.

(10 hrs)

5. Dynamic and Control of some Complex Chemical Process:

Heat exchanger, distillation Column, Introduction to Computer Control of Chemical Process.

(10 hrs)

Note:

• 2nd term

University of Technology Chemical Engineering Department Fourth Year B.Sc	<u>Syllabus</u>
CE 544-U Chemical Industries Theoretica Tutorial Unit Operation Branch Practical	6 1 2 hr/week 1 hr/week 3 hr/week
1. <u>Chemical Processing:</u> Introduction, Operating Conditions, Type of Process.	(3 hrs)
2. <u>Sulpher and Sulpheric acid:</u> Raw materials, Frasch Process, Iead Chamber Process, Contact Process Production.	s, Oleum
3. <u>Ammonia and Nitric acid:</u> Raw materials, Production Steps, Production Process.	(4 hrs)
4. Nitrogenous Fertilizers: Ammonium Sulphate, Ammonium nitrate, Urea.	(6 hrs)
5. Phosphoric acid: Wet Process Phosphoric and, Electric Furnace	(4 hrs)
6. Phosphate Fertilizers: Superphosphate, Triple Superphosphate.	(4 hrs)
7. Electrolytic Industries: Principle and theory, Caustic Soda Production Chlorine Production	(4 hrs)
8. <u>Industrial Salts:</u> Sodash Production	(3 hrs)
9. <u>Ceramic Industries:</u> Raw materials, Chemical Conversions, Porcelain, Fire Clay	(4 hrs)
10. <u>Cement Industries:</u> Raw materials, Portland Cement	(4 hrs)
11. Glass Industries: Raw materials, Manufacture of Glass	(4 hrs)

University of Technology	Chemical Engineering Department	Fourth	Year B.Sc. Sylla
		; <u>,</u> -,-	Units 6
CE 544-U	Chemical Industries		Theoretical 2 hr/v Tutorial 1 hr/v
·	Unit Operation Branch	· · · · · · · · · · · · · · · · · · ·	Practical 3 hr/v
12. Surface Coating Paint, varnishes, laco		*	
13. Oils and Fats:			(4 l
	ture of Vegetable Oils and Waxes		(41
14. Soap and Detergor Soap Production, De			
15. Sugar Industries:			(4)
Cane Sugar, Beet Su	gar		(4 ł

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus		
CE 134	Optimization		Units 2 Theoretical 2 hr/week Tutorial 1 hr/week Practical - hr/week	
1. Introduction to Opti			(4 hrs)	
2. Organization of Opt	imization problems:		(4 hrs	

Analytical methods, numerical methods, graphical methods, restriction Function, unrestricted Function, Direct search, Dichotomous Search, golden Search, Fibonacci Search

(12 hrs)

4. Multivariables Optimization Methods:

- Necessary Conditions For Extreme Values in graphical Cases.
- Solution by graphical method.
- Simplex method.
- Chemical engineering application Linear Programming and in (transportation mixing)

(10 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
		Units 2
CE 644-U	Engineering of Catalysts	Theoretical 2 hr/week Tutorial 1 hr/week
	Unit Operations Branch	Practical - hr/week
1. <u>Catalysts:</u>		
Definitions, Propertie		(2 hrs)
2. Steps in a Catalytic landsorption. Surface	Reaction: Reaction, Decoration, Rate Equation	ons for each Step.
3. Finding Rate Law as		(4 hrs)
Equations of Rate Li		(4 hrs)
	r Gas - Solid Reactions:	
Basic Guidelines. De	sign Equations.	
		(8 hrs)
5. <u>Heterogeneous Data</u>	Analysis for Reactor Design:	(8 hrs)
5. Heterogeneous Data Deducing a Rate	Analysis for Reactor Design: law from the Experimental Damental Observations, Evaluation	ata, Finding a Mechanism

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus	١
		Units 2	_
CE 644-R	Engineering of Catalysts	Theoretical 2 hr/week Tutorial 1 hr/week	
	Rafinary Engineering Reauch	Practical - hr/week	

1. Introduction:

Properties of Catalysts, Determination of Surface Area, Void Volume and Solid Density, Pore Volume Distribution, Classification of Catalysts, Promoters and Inhibitors, Catalyst Poisoning.

(4 hrs)

2. Rate Equations of Fluid-Solid Catalytic Reactions:

Rate of Adsorption, Desorption and Surface Reaction, Rate of the limiting Step, Mechanism, Design of Reactor for- Gas- Solid Reactions. Data Analysis for Reactor Design

(10 hrs)

3. External Transport Processes in Heterogeneous Reactions:

Equations of Mass –and Heat Transfer, Calculations of the Concentrations on the Surface, Calculations of Temperature Differences in External Transport Processes, Fluidized Bed Reactors, slurry Reactors

(4 hrs)

4. Internal Transport Processes:

Gaseous Diffusion in Single Cylindrical Pores, Surface Diffusion, Effectiveness Factor, Thiele – Type Modulus

(4 hrs)

5. Design of Heterogeneous Calalytic Reactors:

Fixed - Bed Reactors, Fluidized Bed Reactor, Slurry Reactions

(8 hrs)

University of Technology	Chemical Engineering Department	Fourth Y	<u> ear B.Sc. Syllabus</u>
CE 744-U	Corrosion Engineering		Units 2 Theoretical 2 hr/week Tutorial 1 hr/week
	Unit Operations Branch		Practical - hr/week
1. Introduction:			
	on, why it happens, direct and ind	lirect disad	lvantages.
			(2 hrs
2. Types of Corrosion:			
Dry Corrosion, wet C	Corrosion, Eight form of Corrosion	1	
			(4 hrs)
3. Mechanism of Crevi	ce Corrosion:		(2 hm
4. Electrochemical cell			(2 hrs
	ypes of reference electrodes.		
Electrode Fotelliai, t	ypes of reference electrodes.		(4 hrs
5. Corrosion rate Calcu	ulation:		,
Wet loss method, ele	ctrochemical method		
			(2 hrs)
6. Polarization:			
Definition, Types.		•	(4 hrs)
7. Passivity:		•	(4 1118)
7. I ussivity.			(2 hrs
8. Effect of Temperatu	re and velocity on corrosion rea	ction:	
			(2 hrs)
9. Method of Protection	<u>n:</u>		
			(4 hrs)
10. <u>Instruments to m</u>	easure Corrosion rate:		
·		•	(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 844-U	Technology of Oil and Gas	Units 2 Theoretical 2 hr/week Tutorial 1 hr/weel
·	Unit Operations Branch	Practical - hr/week
1 History and Develor	oment of Refinery Processes:	
Kinds of Refineries.	, ment of itemer, a seesses.	
		(2 hrs
2. Chemical Compositi	on of Petroleum:	
		(2 hrs
5. <u>Fractionation of Pet</u>	ation, Tube still Heaters. roleum:	(2 hrs
Atmospheric and Va	cuum Fractionation.	
		(2 hrs
6 Fractionation Towe	re:	(2 hrs
6. Fractionation Towe Material and Energ		
	y Balances, Reflux, Temperature I	(2 hrs Distribution in Fractionation
Material and Energ	y Balances, Reflux, Temperature I	Distribution in Fractionation
Material and Energ Towers, Tower Diameter	y Balances, Reflux, Temperature I	Distribution in Fractionation (6 hrs
Material and Energ Towers, Tower Diamete 7. Thermal Cracking:	y Balances, Reflux, Temperature I	Distribution in Fractionation

(2 hrs)

University of T	echnology	Chemical Engineering Department	Fourth Y	ear B.Sc.	Syllabus
CE 234	Industrial	Management and Safety Requirem	ents_	Units Theoretical Tutorial Practical	1 hr/week

1. Management:

Principle of management, types and classifications, management responsibility, organization responsibility.

(4 hrs)

2. Industrial organization

Site, Feasibility study, Development of efficient work method (plant layout, flow of material, material handling), workstations, Inputs and Outputs, Production planning (types of productions).

(3 hrs)

3. Maintenance:

Classifications, cost, machine replacements, case studies and examples.

(3 hrs)

4. Network Analysis:

Principles and applications, Critical Path Method (CPM), Gantt Chart, Pert techniques (examples and case studies).

(3 hrs)

5. Work Measurement Techniques

Time and Motion study.

(3 hrs)

6. Costing:

Framework of management, cost of production (row material cost, labor cost, machinery cost).

(3 hrs)

7. Quality Control:

Standardization, specification, sampling techniques, Inspections – analysis of results. Quality costs (preventive cost, appraisal cost and failure cost). Application of quality control chart – examples, Reliability.

(4 hrs)

8. <u>ISO:</u>

Requirements, applications, ISO series, Quality management system (QMS), Total Quality managements (TQM), Requirements and applications.

(3 hrs)

9. Safety Requirements:

Hazards (types e.g. industrial hazards, pollution (air pollution, water pollution, industrial pollution). Industrial by products and industrial waste, Safety requirements of industrial sites. Requirements of suitable work environment (examples with particular emphasis in chemical industry).

(4 hrs)



	·	
University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
		Units 2
CE 944-U	Polymer Technology	Theoretical 2 hr/week
	Unit On anations Buanak	Tutorial i hr/week Practic <u>al - hr/week</u>
	Unit Operations Branch	Tractical larrock
1. Polymers:		
Nomenclature		(21.)
		(2 hrs)
2. Polymers Classificat	ion:	and an Chamical nature of
	on Sources, Classification bas	eu on Chemical nature of
Polymers, Technical Cla	assification of Polymers	(2 hrs)
3. Polymerization:		(= ::::3)
	ymerization (condensation pol	lymerization) Free radical
Polymerization, Ionic ((cationic, Anionic) Coordination P	Olymerization (Steroregular
	erization by group transfer	
		(6 hrs)
4. Polymerzation System		
Bulk, Solution, Susp	ension, Precipitation	(4 hrs)
5. Molecular Weight o	f Polymers:	(4 1113)
	ment of molecular weight of Polyr	ners
		(2 hrs)
6. Polymer composites	<u>:</u>	
		(2 hrs)
7. Polymer blends:		(2.1)
O Dhawigal Dan a sadian	of Dalamana	(2 hrs)
8. Physical Properties	or rolymers:	(4 hrs)
9. Mechanical Propert	ies of Polymers:	(4 1115)
	pact strength, Creep, hardness	
		(4 hrs)
10. Polymer Process	ing:	, ,
Molding, Extrusion,	Thermal Forming Calendaring	

(4 hrs)

University of Technology Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 544-R <u>Petroleum Refining</u>	Units 6 Theoretical 2 hr/week Tutorial 1 hr/week
Refinery Engineering Branch	Practical 3 hr/week
1. <u>Introduction:</u> History and Development of Refining Processes Kinds	of Refineries. (2 hrs)
2. Chemical Composition of Petroleum:	
- Chemieu Composition	(2 hrs)
3. Physical and Thermodynamic Properties of Petroleu	<u>m oil:</u> (2 hrs)
4. Evaluation of oil stocks:	(4 hrs)
5. <u>Introduction to Processing:</u> Stabilization, dehydration, tube still heaters	
Stabilization, denydration, tube still heaters	(10 hrs)
6. Fractionation of Petroleum: Atmospheric and Vacuum Fractionation	
· · · · · · · · · · · · · · · · · · ·	(2 hrs)
7. Fractionation Towers: Material and Energy Balances, Reflux Temperate I	Distribution in Fractionation
Tower, Tower Diameter	(0 hua)
9 Tuesting Duesesses	(8 hrs)
8. <u>Treating Processes:</u> Removal of Acid Gases, Sweetening Processes, improstorage stability.	vement in Performance and
storage statistics.	(8 hrs)
9. <u>Upgrading processes:</u> Thermal Crackling, Cooking, Visbreaking, Catalytic	
Catalytic Reforming, Alkylation, Isomerisation.	
	(10 hrs)
10. Product Blending: Blending of vapor pressure, octane number, viscosi	ty, flash point aniline point,
and pour point.	(6 hrs)
11. Manufacturing of lubricating oils:	(4 hrs)
12. Treating process using Hydrogen:	(2 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
		Units 2
CE 844-R	Gas Technology	Theoretical 2 hr/week
		Tutorial 1 hr/week
	Refinery Engineering Branch	Practical - hr/week
1. Classification of Ga	seous fuels thermodynamic:	
1. Classification of Gu	Secus fuels file into a file i	(3 hrs)
2. Physical and Chemi	ical Properties for natural Gas:	()
2. Inysicai and enem	icui i roperios ioi maturas casi	(3 hrs)
3. Drying of Gaseous 1	Fuels:	(5 1112)
	on, Drying by Absorption	
Drying by Ausorptic	on, Drying by Absorption	(4 hrs)
1 Sweetening of Case	one Fuole:	(4 1113)
4. Sweetening of Gase	orption, Sweetening by Absorption	
Sweetening by Ausc	orphon, sweetening by Absorphon	(6 hrs)
E. Cas Duagosius unid	· for	(O IIIS)
5. Gas Processing unit		agent and enjetting
Fiash Separation Pro	ocesses, Distillation of multicompor	
7 T ** C . 1 D . 4 1.	Constant	(6 hrs)
6. <u>Liquefied Petroleur</u>	n Gases:	(2.1)
M TO 1 ATY 1		(2 hrs)
7. <u>Production of Hydr</u>	ogen:	
·		(2 hrs)
8. Sulfur recovery pro	cesses from Acid gases streams:	
		(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 744-R	Petrochemical Industries	Units 2 Theoretical 2 hr/week Tutorial 1 hr/week
	Refinery Engineering Branch	Practical - hr/week
1. <u>Introduction:</u> Chemical Process In	dustries	
		(1 hrs)
Processes, Purification System, Diolefins, S Production, Conversion Production methods of Basic materials Aron Syntheses Gas, Steam re	n Cracking Processes Cooling, Pu Processes Compression, Sepa Sources of diolefins, Butadien on Process For Production of f higher olefins, linear, Alkyl be natic Sources, Separation of ben eforming, Partial Oxidation	ration Operations, Energy ne Separation, Isobutylene diolefins, Higher Olefins enzene Complex, Aromatics,
3. Synthetic Zeolites:		(2 hrs)
4. Intermediate Petroc	·hemicals:	(2 1115)
Methanol, Ethyler Ethyleneglycol Acrylor	ne Oxide Acetic Acid, Eth nitrile, Adipic Acid, Methyl Tet benzene, Ethylene, Cyclohexane	rabutyl ether, EthylbenZene,
		(7 hrs)
5. <u>Petrochemical Dete</u>	rgents:	(2 hrs)
6. <u>Polymers:</u> LDPE, HDPE, PP, P	VC	
7. Synthetic Fibers:		(4 hrs)
		(2 hrs)

8. Petrochemical Complexes:

(2 hrs)

ĺ	University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syna	<u>bus</u>
ļ			Units 2	
	CE 944-R	Environmental - Pollution	Theoretical 2 hr/v	veek
			Tutorial 1 hr/v	week
		Rofinery Engineering Reauch	Practical - br/s	week

1. Introduction:

Definition of Pollution, literature surveys on international interest of Environmental Pollution Problems. Pollutants, Type of Pollutants, Measuring Devices, Allowable level of Pollutants in Environment.

(2 hrs)

2. Air Pollution:

Air Pollutants, Particulate matter, SO₂, CO, NO, ...etc Sources and effect, Control Treatment Methods, Design of some Equipment: Centrifugal, Cyclone, Scrubber, Packed Bed Ventura, Spray – tower, Electrostatic Separator, and Chimney Design.

(12 hrs)

3. Water - Pollution:

Water Pollutants, Sources and effect, Organic and non – organic, Radiation activity, Suspended Solids and Sculling Salts, Control Treatment of Water Pollution for: Drinking, Industry Processes, Desalination methods, Reverse Osmosis, Electrodalysis, Solar Distillation, Primary, secondary and advanced treatment of Wastewater, Treatment of oily Pollution.

(12 hrs)

4. Pollution of Solid — Waste:

Sources, Effects, Control Treatment method of Removal Solid – Waste, Treatment by economy methods.

(6 hrs)

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