Ministry Of Higher Education

University Of Technology Chemical Engineering Department



B.Sc PROGRAMME IN CHEMICAL ENGINEERING

OUTLINE OF SYLLABUSES ALLOCATION OF SUBJECTS AND WEEKLY LOAD

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INTRODUCTION

Chemical Engineering is distinguished from other branches of engineering by its strong dependence on chemistry. This enables the chemical engineer to understand properties involving changes in physical state, chemical composition, or energy content for systems ranging in scale from molecules to full sized munifacturing plants.

The chemical engineer may be employed in an established industry producing chemicals, petroleum products, petrochemicals, pharmaceuticals, synthetic fibers, foods, plastics or metals. These products are steadily needed in an increasing amounts due to the expanding growth of population. His function may be involved in making innovations in operations, doing research, development and analysis of existing plants, technical services, or scale up.

Due to his board knowledge, the chemical engineer often occupie; a dominant position in the above mentioned industries.

He also apply his knowledge in such diverse areas as air and water pollution or biochemical research.

The formal course work for B.Sc. programme involves mathematics, including both analysis and computer application, chemistry, mechanics, electricity chemical engineering principles and practice subjects including fluid dynamics, heat, unit operation, chemical reactor design, thermodynamics, and chemical processes. In addition the programme involves English Language which is essential for any engineering student to enable him/her to benefit from thousands of textbooks and periodicals.

A considerable emphasis is laid, in this department, on practical training and laboratory work, which helps students in getting better understanding of the theoretical part of their curriculum.

Moreover students are required to spent twelve weeks, training in industry during summer vacations.

This booklet contains the academic programme and the syllabu; for subjects which are taken by the chemical engineering students during their four years study.

Prof. M. A. Zablouk Chairman of Chemical Eng. Dept.

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		Hours / Week								
Subject Symbol Subject	Subject	F	irst Term		Se	Units				
	Theoretical	Practical	Tutorial	Theoretical	Practica	Tutorial				
CE. 111	English Language	2	-	-	2	-	-	4		
CE. 121	Mathematics	3	-	1	3	-	1	6		
CE. 141	Principle of Chemical Engineering	3	-	1	3	-	1	6		
CE. 221	Chemistry	3	2	1	3	2	1	8		
CE. 131	Engineering Drawing and AutoCAD	1	2	-	1	2	-	4		
CE. 231	Mechanics and Strength of Materials	2	-	1	2	-	1	4		
CE. 331	Electrical Technology	1	-	1	1	-	1	2		
CE. 321	Computer Science	2	2	3-3	2	2	-	6		
CE. 431	Workshop	-	6	1.7	-	6	-	4		
	Total	17	12	5	17	12	5	44		

Note: The student is required to complete twelve weeks summer training in industry

Number of Hours / Week

First Term	Second Term
34	34

Key of subject symbol (from lift):

CE: Chemical Engineering

First No.: Order of subject within specialization

Second No.: specialization

1 : Human subject

2 : Basic subject

3 : General subject

4 : Specific subject

Third No. : Year of Study

-U: Unit Operation of Chemical Industry Branch

-R: Refinery Engineering of Oil and Gas Branch

		Hours / Week									
Subject Symbol Subject	Subject	F	irst Term		Se	Units					
	Theoretical	Practical	Tutorial	Theoretical	Practical	Tu orial					
CE. 112	Human Rights	2	-	-	2	-	-	4			
CE. 122	Mathematics	2	-	1	2	-	1	4			
CE. 142	Fluid Flow	3	2	1	. 3	2	1	8			
CE. 132	Properties of Materials	1	-	1	1	2	1	3			
CE. 222	Computer Programing & MATLAB	2	2	-	2	2	-	6			
CE. 242	Fuels Technology	1	2	1	1	-	1	3			
CE. 342	Thermo dynamics	2	-	1	2	-	1	4			
CE. 232	Physical Chemistry	2	2	1	2	2	1	6			
	Total	15	8	6	15	8	6	38			

First Term	Second Term
29	29

A-Unit Operation of Chemical Industry Branch

		Hours / Week						
Subject Symbol	Subject	F	First Term			Second Term		
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 113	Democracy	2	-	-	2	-	-	4
CE. 133	Applied Mathematics	3	-	1	3	-	1	6
CE. 143	Mass Transfer	3	2	1	3	-	1	7
CE. 233	Statistics and Instruments Measurement	1	1	1	1	1	1	3
CE. 243	Reactor Design	3	-	1	3	-	1	6
CE. 343	Heat Transfer	3	-	1	3	2	1	7
CE. 443	Equipment Design by Computer	2	1	-	2	1	-	6
CE. 543-U	Biochemical Engineering	2	-	1	-	-	-	2
CE. 643-U	Environmental Engineering	-	-	-	2	-	1	2
	Total	19	4	6	19	4	6	43

Second Term
29

B-Refinery Engineering of Oil and Gas Branch

		Hours / Week						
Subject Symbol Subject	Subject	F	irst Term		Second Terri			Units
	Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial		
CE, 113	Democracy	2	-	-	2	-	-	4
CE. 133	Applied Mathematics	3	-	1	3	-	1	6
CE. 143	Mass Transfer	3	2	1	3	-	1	7
CE. 233	Statistics and Instruments	1	1	1	1	1	1	3
CE. 243	Reactor Design	3	-	1	3	-	1	6
CE, 343	Heat Transfer	3	-	1	3	2	1	7
CE. 443	Equipment Design by Computer	2	1	-	2	1	-	6
CE. 543-R	Storage and Transport of Crude and Petroleum Products	-		-	2	-	1	2
CE. 643-R	Combustion Engineering	2	-	1	-	-	-	2
	Total	19	4	6	19	4	6	43

First Term	Second Term
29	29

A-Unit Operation of Chemical Industry Branch

		Hours / Week						
Subject Symbol Subject	Subject	F	irst Term		Se	Units		
	Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial		
CE. 144	Project	1	2	-	1	2	-	4
CE. 244	Transport Phenomena	3	3	1	3	-	1	7
CE. 344	Process Control	2	-	1	2	3	1	5
CE. 444-U	Chemical Industries	2	3	1	2	3	1	6
CE. 134	Optimization	-	-	15.0	2	-	1	2
CE. 544-U	Engineering of Catalyst	2	-	1	-	-	-	2
CE. 644-U	Corrosion Engineering	-	-	-	2	-	1	2
CE. 744-U	Technology of Oil and Gas	-	-	-	2	-	1	2
CE. 234	Industrial Management and Safety Requirements	2	-	1	-	-	-	2
CE. 844-U	Polymers Technology	2	-	1	-	-	-	2
	Total	14	8	6	14	8	6	34

First Term	Second Term
28	28

B- Refinery Engineering of Oil and Gas Branch

				Hours	/ Week			
Subject Symbol	Subject	First Term		Second Term			Units	
		Theoretical	Practical	Tutorial	Theoretical	Practica	Tutorial	
CE. 144	Project	1	2	-	1	2	-	4
CE. 244	Transport Phenomena	3	3	1	3	-	1	7
CE. 344	Process Control	2	-	1	2	3	1	5
CE. 444-R	Petroleum Refining	2	3	1	2	3	1	6
CE. 134	Optimization	-	-	-	2	-	1	2
CE. 544-R	Engineering of Catalyst	2	-	1	-	-	-	2
CE. 644-R	Petrochemical Industries	2	-	1	-	-	-	2
CE. 744-R	Gas Technology	-	-	-	2	-	1	2
CE. 234	Industrial Management and Safety Requirements	-	-	-	2	-	1	2
CE. 844-R	Environmental Pollution	2	-	1	-	-	-	2
	Total	14	8	6	14	8	6	34

Second Term
28
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University of Technology	Chemical Engineering Department	First Year B Sc. Syllabus
CE 111	English	Units 4 Theor tical 2 hr/week Tutor al - hr/week Practical - hr/week
First Semester		
1- Comprehension		
Reading Scientific Pa		
A- Solids, liquids		
B- The Control of		
C-Sir Isaac Newto	n	(10hm)
2 Crammatical Paints		(18hrs)
2- Grammatical Points A-The Sentence		
1- Affirmative		
2- Interrogative		
3- Negative		
8		(4hrs)
B-Parts of Speech		
1- Nouns		
A-Function		
B-Countibility	e:	
C-non-Countibi	lity	(1hrs)
2-Pronouns:		(11113)
Types and function	n	
71		(1hrs)
3-Adjectives		
A-Degree/Compar	rison	
B- Order		/11 \
4 77 1		(1hrs)
4-Verbs		
A-Finite B-Non- Finite		
C-Transitive		
D-Intransitive		
2 minumonte		(1hrs)
5-Adverbs		, San K
A-Classification		
B-Formation		
C-Order		27. S
		(1hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 111	English .	Units 4 Theoretical 2 hr/week Tutorial - hr/week Pract cal - hr/week
6-Prepositions: A general idea		(1hrs)
7-Conjunctions A-Main Conjunction B-Subsidiary	ns	
8-Interjections:		(1hrs)
Common ones		(1hrs)
Second Semester		
1- Comprehension Reading Scientific Pass 1-Electricity in Ear 2-Comets 3-Exploring the mo	ly Days	(16hrs)
2- Grammatical Points		(10ms)
1-Articles A-definite B-indefinite		(2h.vo)
	ubject redicate	(2hrs)
D-Expansion		(8hrs)
3-Written and Oral Tr	anslation	(4hrs)

University of Technology	Chemical Engineering Department	First Year B Sc. Syllabus
CE 121	<u>Mathematics</u>	Units 6 Theoretical 3 hr/week Tutor al 1 hr/week Pract cal - hr/week
ange, Inverse of functi	nt line, Trigonometric functions and ions, Absolute value, limits, L ion) Conic sections (general definit	imits, applications, Polar
- Differentiation and Inte Algebraic functions and	egration: trigonometric functions, Applicatio	on (limits, Horitals rule) (6 hrs)
- Transcendental Function Inverse of Trigonomet unctions (Definitions, Pro	ons: ric functions, Natural logarithmic perties, Sketches, Derivatives and	c Exponential and Power Integrals. (12 hrs)
- Hyperbolic Functions: Definitions, Properties,	Sketches, inverse and Derivatives a	and Integrals. (8 hrs)
	Solution of systems of equations (Cramers Rule) (6 hrs)
Application of Definite Area between two curve	Integrals: es, Volumes, Length of curves, Sur	face area. (12 hrs)
- Moods of integration:		X Down of
Substitution, Integrations,	on by parts, The Substitution Integrals involve $\sqrt{a^2-x^2}$, $\sqrt{a^2-x^2}$	on $U = \tan \frac{\pi}{2}$. Power of $\frac{\pi}{2}$, $\sqrt{\chi^2 - \zeta^2}$, integrals
	ctions, Improper Integrals (conv an	
		1.11

7-

(16 hrs)

8- Complex Numbers:

Definitions, Argands diagram, Addition and subtraction and multiplication and dividing numbers, Trigonometric and exponential formula of complex number, Demoivors theorem, Roots of equation, Power of complex numbers.

(9 hrs)

9- Vector Analysis:

Definition, Addition and subtraction of vectors, Multiplication by scala's, Direction, products of vector (scalar product and vector product).

(9 hrs)

University of Technology	Chemical Engineering Department	First Year B Sc.	Synabus
		Units	6
CE 141	Principle of Chemical Engineering	Theor :tical Tutor al	3 hr/week 1 hr/week
		Dreet cal	- hr/week

1- Concept of Chemical Engineering:

Units, Dimensions, Conversion factors, Conversion equation, Temperature, Pressure, Composition, Chemical Analysis, Chemical equations.

(24 hrs)

2- Material Balance Calculation:

Without and with chemical reaction, Recycle, By pass and purge calculations for steady state, Combustion Calculation.

(30 hrs)

3- Ideal Gas laws:

Definitions of Ideal gas, Real gas, Ideal gases mixture, Real gas mixture.

(9 hrs)

4- Energy Balance:

General energy balance, Enthalpy, Heat capacities their predications and variation with temperature, Heat effects.

(18 hrs)

5- Simultaneous, Mass and Energy Balance:

(9 hrs)

University of Technology (Chemical Engineering Department	First Year B. Sc. Syllabus
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CE 221

Chemistry

Units 8
Theoretical 3 hr/week
Tutori 1 1 hr/week

Practical 2 hr/week

Part -1- Analytical Chemistry

1- Introduction:

Atomic weight, Molecular formula, Chemical equations, Mole concept, and Chemical equilibrium.

(4 hrs)

2- Solution:

Definition, Preparation and properties, Molarity, Normality, Formality, PH, POH, Solubility

(8 hrs)

3- Analytical Methods of Analysis:

a-Qualitative Analysis

b-Quantitative Analysis.

- Volumetric (titrimetric) & analysis,
 Acid- base, Redox, Precipitation, Complex titration, Methods of calculation,
 Titration curves
- Gravimetric Analysis
 Precipitation reactions, Direct and indirect methods of analysis, Ksp.
- iii. instrumental Methods of Analysis Photometric,, Colourimetic, Potentiometric titration, , PH-measurements Chromatography, Atomic absorption.

(33 hrs)

First Year E.Sc. Syllabus Chemical Engineering Department University of Technology Units Theo etical 3 hr/week Chemistry CE 221 1 hr/week Tuto: ial Practical 2 hr/week Part -2- Organic Chemistry 1- Introduction: Definition and classification of organic compounds. (3 hrs) 2- Aliphatic Compounds: Nomenclature, Preparation, Properties, Reactions. a) Alkanes b) Alkenes c) Alkynes d) Aliphatic Derivatives: Alkyl halides Alcohol's, Ethers, Aldehydes, Ketones, Carboxylic acids, Amines, Mercaptans, Carboxylic acids, Amines, Mercaptans, Thiophenes, Disulphide. (20 hrs)

3- Aromatic Compounds:

Structural formula of benzene ring, Nomenclature, Preparations, Properties, Chemical reactions; Nitration, Sulphonation, Halogenation, Alkylation, Benzene- 10mologues; Preparation, properties, chemical reactions of - Toluene, Xylene, Ethyl benzene, Styrene, Phenols, Anilines, Naphthalene.

(10 hrs)

4- Heterocylic Compounds:

Pyridine, Furan, Pyrrole.

(3 hrs)

5- Introduction to Polymer and Carbohydrate:

(5 hrs)

6- Organo-Metallic Compounds:

(4 hrs)

University of Technology	Chemical Engineering Department	First Year B. c. Syllabus
CE 131	Engineering Drawing and AutoCAD	Units 4 Theore ical 1 hr/week Tutorial - hr/week Practic al 2 hr/week
Part -1- Engineering I	Drawing	
	a bout engineering drawing and i s used in drawings, Hidden lines.	ts importance, Drawing
moraments, Types of me	s ased in drawings, rinden intes.	(2 hrs)
2- Geometrical construct	ion:	(1 hrs)
3- Pictorial Drawings:		
(Isometric and Oblique)		(2 hrs)
4- Projection Drawings: First and Third Angle P	rojection.	
	,	(2 hrs)
5- The Finding of 3 rd vi	ew:	(2 hrs)
6- Sectioning:		(2 hua)
7- Fasteners:		(2 hrs)

(2 hrs)

8- Assembly and detail drawing:

(2 hrs)

University of Technology	Chemical Engineering Department	First Year	B. Sc.	Syllabus
CE 131	Engineering Drawing and AutoCAL	The Total	nits neon tica ntori il racti :al	4 l 1 hr/week - hr/week 2 hr/week
Part -2- AutoCAD				
1- Definition of AutoCa		ogram and	ergir	
drawing	explain the connection between its pr	ogram and	CI. BII	neering
drawing.	explain the connection between its pr	ogram and	or gir	neering (2 hrs)
2- Explain in details th	e command of AutoCAD drawing:		or gri	
2- Explain in details th			or gi	
2- Explain in details the (line, circle,atc.), at 3- Explain the drawing	e command of AutoCAD drawing:	ons.		(2 hrs)

6- Explain the method of forming for three – dimensions shape: giving the applications examinations.

4- Giving a simple idea for three - dimensions drawing (3D):.

5- Forming a simple shape for three - dimensions drawing (3D):.

(3 hrs)

(3 hrs)

(2 hrs)

1021		
	University of Technology Chemical Engineering Department First Year B. Sc. S	Syllabus
	CE 231 Mechanics and Strength of Materials Theore ical Tutorial Practical	4 2 hr/week 1 hr/week - hr/week
	A-Mechanics	
<u>1-</u>	Principles of Statics:	(2 hrs)
2-	Resultants of Force systems:	(5 hrs)
<u>3-</u>	Equilibrium of force systems:	
4-	Friction:	(5 hrs)
		(6 hrs)
	Centurions and Centers of Gravity:	(5 hrs)
<u>6-</u>	Moment of Inertia:	(4 hrs)
<u>7-</u>	Principle of Movement:	(3 hrs)
	B-Strength of Materials	
1-	Internal Forces in Rigid Bodies:	
2-	Definition of stress and strain:	(1 hrs)
	Types of stresses and strain, shear stress.	/ · · · · ·
3-	Stress- Strain Diagrams:	(1 hrs)
	Stress- strain diagrams for ductile and brittle materials	(2 hrs)
4-	Proportional limits:	(=)
	Elastic limit, stiffness elasticity, plasticity, Toughness, Resilience, Hardness, Working stress.	
	Managaran W. C. Salatina (M.	(2 hrs)

University of Technolog	gy Chemical Engineering Department	First Year B Sc. Syllabus
CE 231	Mechanics and Strength of Materials	Units 4 Theoretical 2 hr/week Tutor al 1 hr/week Pract cal - hr/week
5- Hook law:		(2 h)
6- Poisson's ratio, Con	anosita etresses:	(2 hrs)
6- Poisson s Pano, Con	iposite stresses.	(4 hrs)
7- Volumetric stress, B	Bulk modules:	(4 hrs)
8- Thin walled Cylinde	ers:	(4 1)
9- Thermal stress:		(4 hrs)
9- Thermai stress.		(4 hrs)
10- shear and Moments	in Beam:	
Deflection		(4 hrs)
11- Wind Effect on the	Tower high Tower:	(2 hrs)

University of Technology	Chemical Engineering Department	First Year I .Sc. Syllabus
CE 331	Electrical Technology	Units 2 Theoretical 1 hr/week Tuto ial 1 hr/week Prac ical - hr/week
1- Semiconductors Equipr	ment:	(3 hrs)
2- Rectifiers and Detectors	<u>s:</u>	(2 hrs)
3- Electronic Amplifiers:		(4 hrs)
4- D.C. Circuits:	fatawa.	(2 hrs)
5- D.C. Generators and M 6- A.C Circuits:	totors:	(3 hrs)
7- Polyphases Circuits:		(3 hrs)
8- Transformers and Indu	uction Motors:	(2 hrs) (3 hrs)
9- <u>Starters</u> :		(2 hrs)
10- Integrated Circuits, Mo	easuring Instruments, Tranduce	
11- Electrical Heating App	liance:	(2 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc	. Syllabus
CE 321	Computer Science	Units Theoretic Tutor al Pract cal	6 al 2 hr/week - hr/week 2 hr/week
<u>1- Introduction:</u> Component (hard ware), co	omputer work, computer, composition	on, Application	(2 hrs)
<u>2- Files:</u> Definition, Types, Nomeno	elature, MSDos, External and interna	al commends.	(4 hrs)
3- Intoduction to window Desktop, Mouse, my comp	operating system: uter-icons, close window, stand by.		(2 hrs
4- Folders: Size and cascade, windo file or folder copy file or fo	ws folder construction, construction	n choose file or fo	
	n windows, format & Test Floppy di	isk, Arrange I :or	(2 hrs)
Help. 6- Word Program:			(2 hrs
	Open, Close, Save, Save as, Page Paste, Clear, Select All, Find, Repl		do typing (2 hrs
7- Menu Table: View (zoom, Header and numbers, Textbox, Object)	and footnote, Tool bar Insert (Picture	e, Footnote, Sym	

<u>7-</u>

Format (Paragraph, Font, Borders, Bullets & numbering Spelling and grammar Table (shading, Columns, Insert, Table Insert Rows, Language, Delete, Cells, Split Cells, Select Row, Select column) (Table Autoformat Sort, Formula).

(6 hrs)

8- Program Excel:

Open, Taskbar, Toolbar, Format, Sheet, Cell, Construction of sheet, Oper file, Clear (Delete), Save (Save as), Close, cascade, Copy and Paste, Add, Delete, Find, Replace, Auto summation, Insert, equation, Format sheet, Auto format, Print, Print preview, Draw.

(12 hrs)

University of Technology	Chemical Engineering Department	First Year B Sc. Syllabus
		The state of the s

CE 321

Computer Science

Units 6

Theor tical 2 hr/week
Tutor al - hr/week

Pract cal 2 hr/week

9- Power Point:

Definition

FILE: New, Open, Open, Close, Save, Save as, Page set up, Print, Exit)

EDIT: Undo, Repeat, Cut, Copy, Paste, Select All, clear Duplicate, Delete slide,

find, Replace

FORMAT: Font, Alignment, Text Direction slide, Layout, Background Apply Slid Show, View show, Rehearse timing design set up show, present Animation Animation Preview, Slide Transition).

(10 hrs)

10- Auto Cad:

Introduction, define screen draw, select point in the screen Limits, Stitus, Circle, Erase, Line, Color, Point, zoom, Redraw, Arc, Ellipse, Polygon, Donut Solid, Fill Break, Copy, More, Pan, Mirror Trim, Extend, Charge, Line type Ltscal, Dis Area, Trace, Hatch text, Snap, Grid, Rotate, Cops, Ortho, Fillet, Dim Layer.

(12 hrs)

11- Internet:

(6 hrs)

University of Technology	Chemical Engineering Department	First Year B Sc. Syllabus	
Seed and the control of the control		Units 4	

Theoretical - hr/week

Pract cal 6 hr/week

- hr/week

Tutor al

Workshop

Refer to Centralize Curriculum

CE 431

University of Technology	Chemical Engineering Department	Second Year B.Sc.	Syllabus
		Units	4
CE 112	Human Rights	Theoretical Tutor al	2 hr/week - hr/week
		Pract cal	- hr/week

Standard Syllabus

University of Technology	Chemical Engineering Department	Second Year I .Sc. Syllabus	
		Units	4
CE 122	Mathematics	Theo etical Tuto ial	2 hr/week 1 hr/week

1. Partial Differentiation:

Function of two or more variables, limits and continuity, partia derivatives, chain rule, Gradients, Directional derivatives, Higher order derivatives, Maxima, minimum and saddle points, Lagrange multipliers.

(10hrs)

Prac ical - hr/week

2. Multiple integrals:

Double integrals, Area, Triple integrals in rectangular coordinates, Physical application of double and triple integration.

(9hrs)

3. Ordinary Differential Equations:

Solution of first order ordinary differential equations, Solution of second order ordinary differential equations, and higher order differential equation.

(9hrs)

4. Infinite Series:

Power series of functions, Taylor's theorem, Integration, Differentiation, Fouries series, Even and odd functions, Half range expansion, Periodic functions.

(8hrs)

5. Function and Definite Integrals:

The error function, The gamma function, The beta function, factor al function.

(6hrs)

6. Vector Analysis:

Products of three vectors, Equation of lines and planes.

(5hrs)

7. Complex Algebra:

Continuity, Derivation of Complex Variables, Analytic function, Integration of functions of complex variables and cauchy's theorem.

(5hrs)

8. Matrices:

The Matrix, Matrix algebra, The transpose of matrix, The inverse of matrix, Eigen Values and Eigen vectors, solution of linear equation by matrix.

(8hrs)

University of Technology Chemical Engineering Department - Second Year F.Sc. Syllabus

CE 142

Fluid Flow

Units 8

Theo etical 3 hr/week
Tutorial 1 hr/week

Pract cal 2 hr/week

1. Introduction:

Physical properties of fluid, Definition of type fluid: Newtonian's, non-Newtonian, Incompressible, compressible fluid, static, Dynamic fluid, snear stresses in fluid.

(4hrs)

2. Dimensional- Analysis:

Rayleigh's method (Power series), Buckingham's ∏-theorem.

(6hrs)

3. Fluid- Dynamic:

Fluid dynamic system, Boundary layer, Continuity equation, Momentum equation, Bernoulli's equation, Euler's equation, Application of Bernoul i's equation and continuity equation.

(10hrs)

4. Newtonian's Fluid-(Incompressible Fluid):

Laminar and Turbulent flow, Reynolds's number, Friction losses and Pressure drop in pipes and Fitting, Determine of pipe diameter, Flow in Branch pipes, Friction losses in non-circular pipes, velocity Distribution.

(10hrs)

5. Non- Newtonian Fluid:

Definition, Type of fluid depend on time, Calculation of friction and pressure drop for general time independent in laminar and Turbulent flow.

(6hrs)

6. Compressible Fluid in pipes:

Pressure wave, Mach number, General equation, General equation for isothermal and adiabatic condition, work for isothermal and adiabatic of Compressor.

(10hrs)

7. Flow-Measurement:

Pilot-tube, Orifice meter, Venturi meter, Rotameters, Notch air Weirs, Special Flowerets (Hot wire anemometer, Gas meter).

(12hrs)

8. Pumping of Liquids:

Calculation of Total head, NPSH, Performance characteristics curve, Calculation of horsepower, equipment, and cost.

(8hrs)

9. Flow of Fluid Through Granular Beds:

Beds & Packed Column, Fixed beds, Fluidized Bed.

(8hrs)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus	
		Units	8
CE 142	Fluid Flow	Theo etical	
		Tutoi al Pract cal	1 hr/week 2 hr/week

10. Mixing of liquids in Tanks:

Mixing, Mixers type, Small blade area, large blade area, Dimensionless groups, Power consumption.

(6hrs)

11. Two- Phase Flow and Pressure drop in pipes:

(4hrs)

12. Introduction in unsteady Flow:

(6hrs)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus	
		Units 3	
OF 122	D 4' CM - 4 ! - 1 -	779 1 1 1 1 / 1	

CE 132	Properties of Materials	Theore ical	1 hr/week
		Tutorial	1 hr/week
		 Practic il 	2 hr/week

1. Engineering requirement for materials:

Mechanical Properties, stress-strain curve, elasticity, plasticity, duct lity, young modulus, tensile stress, yield stress, bricking stress, true and engineering stress-strain diagram.

(4hrs)

2. Atoms and atomic coordination:

Ionic bond, inter-atomic distance attraction forces between atoms, coordination number, covalent bond, Metallic bond.

(4hrs)

3. Crystal structure:

Crystal structure, unit cell, types of unit cells simple cubic, Face centered cubic, body centered cubic, atomic packing factor, Previous lattice, Miller index, crystalline Planes, X-ray diffraction.

(4hrs)

4. Atomic disorder in solid:

impurities in solids, solid solutions in metals, Imperfection incrystals, point defect, line dislocation, surfaces + grain boundaries.

(4hrs)

5. Atomic movement in solid state:

Atomic diffusion, self-diffusion, diffusivity Fick's first law, diffusion constant, diffusivities versus temperature, Factors effect diffusivity.

(4hrs)

6. Phase diagrams:

Definition of alloys, binary alloys, phase diagrams (equilibrium thermal diagrams), eutectic; solid solution and combination type diagram, Iron-carbon face diagram, allotrope Trans formation (polymorphism).

(6hrs)

7. Corrosion:

Definition, why it happen, Type of corrosion, Form of corrosion, corrosion reactions (anode + cathode) corrosion rate measurement, methods of prevention.

(4hrs)

Note:

• 2nd term

University of Technology Chemical Engineer	ring Department	Second Year I.S	c. Syllabus
CE 222 <u>Computer Program</u>	nming & MatLA	Units Theoret Tuto ial Prac ica	
Part -1- Visual Basic			
1. <u>Introduction to Visual Basic</u> : Open, Save File As Standard EXE	file.		
2. Menus:			(2hr)
Form window, Project Explorer w		es window, Cod	e window,
Project window, Toolbox window, Debug 3. <u>Using Toolbox</u> :	g window.		(4hr)
Simple examples on using toolbox iter	ns.		(4hr)
4. <u>Variables</u> : Data, numerical and string data, va	riables, Assignm	ent statement, ina	athematical
and logic operations.			(4hr)
5. Arithmetic operators:			(2hr)
6. <u>Condition</u> : If Then statement, multiple If		If Then ols	se, multiple
IfThen Else , Select case statement	ė.		(4hr)
7. Messages: Msgbox, InputBox.			(4hr)
8. <u>Loop:</u> For Next Loop, Do while Loop	, Do Loop Wh	ile, Exit For , Ex	it Loop. (4hr)
9. Array: Array definition, Declaring Arrays,	Using for next L	oon with Arrays	
operations on Arrays.	Osing for next by	20p 1	(6hr)
10. <u>Database:</u> Working with database, Adding and o	eleting records, I	Displaying data.	(Shr)

(8hr)

University of Technology	Chemical Engineering Department	Second Year F.Sc. Syllabus
CE 222	Computer Programming & MatLA	Units 6 Theo etical 2 hr/week Tuto ial - hr/week Prac ical 2 hr/week
Part -2- MatLab		
1. Introduction to Mar M-files, getting help		(21)
2. Algebra: 2.1 Variable Names		(2hr)
	art with a letter and the intial letter	can be followed by letters (1hr)
2.2 <u>Basic Arithmetic</u> Addition +, Subtraction	<u>c</u> n -, Multiplication *, Division /, Exp	onentiation ^. (1hr)
 min: smallest element factorial: factorial fun 	in an array max([132,129,66,120]) tin an array min([132,129,66,120]) action factorial(12) (n) to an integer round(3.5) an integer floor(-3.1) (1) (1) (1) (1) (2) (2) (2) (3)	(2hr)
2.4 <u>Complex Numb</u> MatLAB can also hand 2.5 <u>Loops</u>		
3.1 Representing a	you can use the command solve. function n, use the command inline.	(1hr) (1hr)
3.2 <u>Differentiation</u> To differentiate a funct	tion, we use the command diff	(1hr)

We can use MATLAB to solve differential equations. The command for finding the symbolic solution is dsolve. For that command, the derivative of the function y is represented by Dy.

(1hr)

3.5 Multiple integrals

(1hr)

3.6 Limits

You can use limit to compute limits, left and right limits as well as infinite limits.

(1hr)

4. Graphics:

4.1 Two-Dimensional Plots

The simplest way to graph a function is to use the command ezplot

(1hr)

4.2 Curves in Three-Dimensional Space

The command for drawing 3D curves is ezplot3.

(1hr)

5. Vectors and Matrices:

5.1 Vectors

You can enter a vector by typing a list of numbers separated by either commas or spaces inside the square brackets.

- dot (dot product)
- cross (cross product)
- norm (norm or length of a vector)

(1hr)

5.2 Matrices

- A+B (matrix addition)
- A-B (matrix subtraction)
- c*A (scalar multiplication)
- A*B (matrix multiplication)
- A' (conjugate transpose of A)
- det(A) (determinant of square matrix A)
- rank(A) (rank of A)

(2hr)

University of Technology	Chemical Engineering Department	Second Year	B. c. Syllabus

		Units 3	
CE 242	Fuels Technology	Theore cal 1 hr/weel	(
		Tutorial 1 hr/wee	k
120		Practi al 2 hr/wee	k

1. Energy and Fuel:

Energy and his classification in Technology section, Physical and chemical classification of fuel, solid Fuels (coal), the industrial processes on coal to Derived different Product.

(3hr)

2. Properties of oil stocks and prepare fractions oil:

Oil stocks, industrial processes which applied from the well to the different fractions in tower distillation and the processes on fraction.

(4hr)

3. Petroleum Fractions properties:

Benzene, Kerosene, Jet Fuel, Diesel, lubricants oil Asphalt, Fuel oil.

(9hr)

4. Gaseous Fuels and Liquefied gases:

Energy source types gases fuel types and liquefied gases which source of energy in industrial and houses.

(4hr)

5. Combustion Calculation:

Different type of equations about the subject and from it found the component which burner and the combustion equation and guess the flue gases and study behavior of combustion.

(6hr)

6. Burner of gases fuel:

The burner types of gases fuel and liquid fuel properties arrange it in Firnace.

(2hr)

7. Combustion in Furnaces:

Furnace types in oil industrial, Boilers and steam production that used in different industrial.

(2hr)

Note:

• 1st term

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
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		Units 4	
CE 342	Thermodynamics	Theoret cal 2 hr/wee	K
		Tutorial 1 hr/wee	k
		Practical - hr/wee	k

1. Introduction:

Fundamental quantities Time, Length, Mass, Force, Temperature.

(2hrs)

2. First Law and Other Basic Concepts:

Joules experiments, Internal Energy Formulation of the first law, Thermodynamics state and state functions Enthalpy, The steady state flow processes, The reversible processes, Heat Capacity and specific Heat, Equilibrium the phase rule.

(4hrs)

3. Applications of the Equations of Ideal Gases, Generalized Correlation and the Acentric Factor.

(8hrs)

4. Heat Effects:

Heat capacities of gases as a function of temperature, solids and liquids, Heat change accompanying phase change, Heat of industrial reactions.

(6hrs)

5. The Second Law of Thermodynamics:

Heat engine, Entropy, Second law limitation and real process Entropy change and irreversibility and probability (statistical thermodynamics), Third law of thermodynamics.

(6hrs)

6. Thermodynamics Properties of Fluids:

Relationships among the thermodynamics properties (including Helomtz and Gibes free energies and chemical potential), Steam formation and two phase system, Saturated temperature and pressure, Triple point, Wet vapor and dryness fraction, Types of thermodynamic tables and diagrams, Steam power plant cycle ard analysis, Barometric condenser, Metering and throttling processes, Steam and as turbines.

(10hrs)

7. Refrigeration and Liquefaction:

Refrigeration cycles (Carnot, Air, Vapor - Compression) and Comparisons, Choice of refrigerant, Absorption Refrigeration, The heat pump, L quefaction process.

(10hrs)

University of Technology	Chemical Engineering Department	Second Year B. Sc. Syllabus	
		Units	4
CE 342	Thermodynamics	Theore ical	
		Tutorial	1 hr/week

8. Phase Equilibrium:

The nature and criteria of equilibrium, Binary system, Vapor pressure of an ideal solution and non ideal solutions, Henrys law, Activity and activity coeffic ents, Flash separation calculations.

(8hrs)

Practical - hr/week

9. Chemical Reaction Equilibrium:

Thermodynamics of ideal gases and mixtures, Derivation of the general equilibrium expression, Chemical equilibrium of ideal and non-ideal gases, Reaction equilibrium in solution, effect of temperature on chemical equilibrium.

(6hrs)

University of Technology	Chemical Engineering Department	Second Year B. Sc. S	<u>Syllabus</u>
		Units	6
CE 232	Physical Chemistry	Theoretical	2 hr/week 1 hr/week

1. Change of State:

One component systems: Water, carbon dioxide and helium, Vapor pressure and external pressure. Liquid-vapor composition diagram, Liquid-Liquid binary solution mixture, Roulette's law for ideal mixtures, Non-ideal Liquid mixtures, The principle of distillation, Henry's law, solubility of gases in Liquids.

(10hrs)

Practi:al 2 hr/week

2. Surface Chemistry:

Pressure difference across curved surfaces, Surface tension, and capillary rise, Langmuir adsorption isotherm and langmuir theory of adsorption, Cal sulations of surface area of the adsorbent, BET equation for surface area calculation.

(12hrs)

3. Chemical Kinetics:

The rate of chemical reactions, order of reaction and rate constant zero, first, second and third order rate equations, Reversible reactions, Consecutive reactions, parallel reactions, Determination of the order, Reactions in flow systems, Effect of temperature on reaction rate, The transition-state theory.

(16hrs)

4. Catalysis:

Reaction of catalyst in homogeneity system, Enzyme reactions an l Kinetic of reactions.

(6hrs)

5. Electrochemistry:

Conductivity measurements, Diffusion and ionic mobilities, Activity and ionic strength, Determination of activity coefficient from solubility, The Debye-Hackle theory, Acid-base catalysis and their dissociation constant.

(8hrs)

6. Electrochemical Cells:

Electromotive force (EMF) of a cell, Measurements of EMF- the po entiometer, The polarity of electrodes, The cell reactions and reversible cells, Free energy and reversible cells, Types of half cells and classification EMF, Standard electrode potentials, Standard free energy and entropy of aqueous ions, Calculation of EMF of a cell, Oxidation-reduction reactions, Concentrations cells, Electrolysis Corrosion. Kinetic of reactions.

(8hrs)

University of Technology	Chemical Engineering Department Third Year		Syllabus
		Units	4
CE 113	Democracy	Theoretical	2 hr/week
		Tutor al	- hr/week
		Pract cal	- hr/week

Standard Syllabus

University of Technology	Chemical Engineering Department	Third Year 3.Sc.	Syllabus
		Units	6
CE 133	Applied Mathematics	Theo etical	3 hr/week
		Tuto ial	1 hr/week
		Prac ical	 hr/week

1. Revision ordinary differential equations:

First and second order, simultaneous differential equations, Application for chemical engineering.

(18hrs)

2. Solution of differential equations by series:

Simple series, method of Frobeins, Bassel's equation. Application for chemical engineering.

(18hrs)

3. The Laplace transformation:

The Laplace transformer, Properties of the Laplace transformation The inverse transformation. Inversionby partial fraction. Convolution. Solution of differential equations. The transforms of special functions. Step function, staircase function, periodic function, Trigonometric function.

(18hrs)

4. Partial Differential Equations:

Formulating of partial differential equations. Solution of partial differential equations. Method of separating variables and Laplace transformation method.

(18hrs)

5. Numerical Analysis:

The difference operators. Interpolation. Finite difference equations. Solution of first and second order differential equations.

(18hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus		
		Unit:	7	
CE 143 <u>Mass Transfer</u>	The retical Tuterial	3 hr/week 1 hr/week		

1. Introduction:

(4hrs)

Practical 2 hr/week

2. Mass transfer fundamentals:

Steady state mass transfer. Diffusion in binary and multicompouent gases. Diffusion in binary and multicompouent liquids. Connective mass t ansfer. Mass transfer coefficients. Two-film theory. Penetration theory. Unsteady state mass transfer methods

(25hrs)

3. Methods of gas-liquid separation (absorption and stripping):

Determination of interfacial mole fractions. Individual and overall mass transfer coefficients. Effect of gas solubility on equilibrium curve. Single stage absorption, multistage absorption. Finding number of stages graphically and through use of correlations for absorption and stripping. Packed columns limits of operation. Design of packed beds for dilute and concentrated solutions. Finding number of transfer units. Absorption with chemical reaction.

(16hrs)

4. Binary distillation:

Introduction, (T-x,y) and (x,y) equilibrium curves. Flash or equilibrium distillation for binary and multicomponent mixtures. Distillation with reflux using Mckabe-Thiele. Sorel method to find number of stages. Fenske equation to find minimum reflux ratio and minimum theoretical stages. Column efficiency. Thermal method for finding number of ideal stages (Ponchoue-Savarit Method). Multicomponent differential distillation.

(25hrs)

5. Multicompouent distillation:

Dew and bubble points calculation methods. Multicomponen equilibrium relationships. Minimum reflux and minimum number of stages. Short-cut method. Checking validity of reflux ratio used. Operating limits of distillation columns.

(14hrs)

6. Mechanical separation process and size reduction:

Classification of equipment. Particle shape factor. Power requiremen's in crushers. Particle size analysis.

(6hrs)

Note:

• 1st term

University of Technology Chemical Engineering	Department Third Year B Sc. Syllabus
CE 233 Statistics and Instrument	Units 3 Theore ical 1 hr/week Tutori: 1 1 hr/week Practical 1 hr/week
Part A: Statistics	
1. <u>Introduction:</u> Qualitative and conclusive statistics, popularly of data.	ation samples, graphical representation
	(2hrs)
 Table and graphical representation: Primary data, repeatability, normal distribut 	ion. (2hrs)
3. Media determination:	
4. Scatter determination: Standard and relative deviations.	(2hrs)
5. Standard distributions:	(2hrs) (4hrs)
6. <u>Chi distribution:</u> Confidence limits, Chi test, degree of fre	
7. Curve fitting:	(2hrs)
Least square method, linear relations, correlation coefficient.	exponential equations, gradient, and
	(2hrs)
Part B: Measurements	
1. Measurements: Absolute and relative. Environmental properties.	conditions. Application on physical

(2hrs)

2. Errors:

Types of errors, determination of errors.

(4hrs)

3.Instruments:

For temperature, Pressure, level, composition, velocity, flow rate.

(8hrs)



University of Technology	Chemical Engineering Department	Third Year I .Sc. Syllabus		
		Units	6	
CE 243	Reactor Design	Theo etical	3 hr/week	
CB 2.13		Tutor al	1 hr/week	
		Pract cal	- hr/week	

1.Introduction:

Stoichiometry, reaction rate constant, order of reaction, reversible and nonreversible reactions, conversion, yield, selectivity.

(9hrs)

2. Types of reactors:

Batch reactors, continuous flow reactors i.e. CSTR, PFR. Industrial reactors.

(9hrs)

3. Bach reactors:

Design equation, reaction time, maximum production rate, no 1-isothermal operation, adiabatic operation

(18hrs)

4. Tubular reactors:

Design equation, isothermal and non-isothermal operation, res dence time, length of reactor, number of tubes, pressure drop, space velocity.

(18hrs)

5. Continuous stirred tank reactions:

Design equation for isothermal operation, scale-up of liquid-phase batch data to the design of CSTR. CSTR's in parallel and in series.

(15hrs)

6. Unsteady state operation of reactors:

Start up of CSTR, semi-batch reactor.

(9hrs)

7. Multiple reactions:

Operation in parallel and in series for single and multiple reactions to obtain maximum amount of desired product.

(6hrs)

8. Comparison between reactors:

Comparison between batch, tubular, and CSTR for single and multiple reactions

(6hrs)

University of Technology	Chemical Engineering Department	Third Year 3.Sc.	Syllabus
		Units	7
CE 343	Heat Transfer	Thecretical Tuto ial	3 hr/week 1 hr/week

Chemical Engineering Department

1. Modes of heat transfer:

Material properties of importance in heat transfer.

(2hrs)

Third Year B.Sc. Syllabus

Practical 2 hr/week

2. Steady state heat conduction in one dimension:

Plane wall, radial systems, heat source systems. Boundary surrounded by fluids. Overall heat transfer coefficient. Extended surface, conductio 1-convection systems, fins.

(24hrs)

3. Principles of convection:

Transport equations. Fluid mechanism aspect of convection. Lami ar boundary layer. Thermal boundary layer. Empirical and practical relations for pipe and tube flow. Flow normal to single and tube banks.

(18hrs)

4. Heat exchangers:

Various types and their general characteristics. Fouling factor. Heat exchanger mean temperature differences. Co-current and counter-current flow.

(12hrs)

5. Shell and Tube Exchangers:

Types and various specifications. Design calculations by conventional and by effectiveness (NTU) methods. Optimum design calculation.

(6hrs)

6. Condensation and Boiling Heat Transfer:

Condensation of single vapors. Design calculations for condense; condensersubcooler and superheater condenser.

(9hrs)

7. Radiation and furnace design:

Radiation properties, shape factor, heat exchange for non black bedies, parallel planes, shields, gas tradition, boiler.

(9hrs)

8. Unsteady state heat transfer:

Temperature as a function of time, lumped capacity system, quenching of small bodies, heating of tank reactor.

(10hrs)

Note:

• 2nd term

University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus		
		Units	6	
CE 443	Equipment Design by Computer	Theoretical Tutorial	2 hr/week - hr/week	
		Practical	1 hr/week	

Part (A)

1. Process Planning:

Scheduling and flow sheet design. Flow sheet types and design tion. Block diagram. Process flow sheet. Piping and instruments diagram. Utility flow sheet. Equipment layout and plot plan. Project evaluation. Feasibility studies. Site selection. Project cost estimation.

(6hrs)

2. Piping networks and pumps:

Pipe fittings, valves. Steam traps. Piping design standards. Piping material and selection. Pump specifications and data sheets. Optimum diameter. Types of pumps. Hydraulic characteristics for centrifugal pumps.

(6hrs)

3. Vessels and Tanks:

Types of vessels. Flash drums. LPG tanks. Criteria in vessel design. Stress considerations. Design of tall vertical vessel. Design of pressure vessels. Supports and foundations

(6hrs)

4. Heat transfer equipment:

Types of exchangers and applications. Exchanger ratings. Exchanger design. Exchanger specification sheets. Furnaces. Convection and radiation zones. Types of fuels. Burners and arrangements. Steam boilers, types of boilers.

(6hrs)

5. Mass Transfer Equipment:

Types of columns. Plate and packed Types of plates and packing. Design features. Pressure drops in columns.

(6hrs)

University of Technology	Chemical Engineering Department	Third Year 3.Sc.	Syllabus
		Units	6
CE 443	Equipment Design by Computer	Theo etical	
		Tuto ial	- hr/week
		Prac ical	1 hr/week

Part (B): Complete equipment design of each of the following units:

1. Pressure vessels, pumps and compressors:

Flash drum. Gas-liquid separator, LPG storage tanks. Cyclone separators. Gas movers and compressors. Pumps. Piping network. etc.

(8hrs)

2. Heat Equipment:

Shell and tube heat exchangers. Plate heat exchanger. Boilers. Furnaces. Vaporizers. Air coolers. Coil type heaters. Condensers. etc.

(10hrs)

3.Mass transfer equipment and reactors:

Distillation column. Absorber. Extractor. Dryer. Absorption towers. Leaching equipment. Scrubbers. Stripping towers. Reactors.

(12hrs)

University of Technology	Chemical Engineering Department	Third Year E.Sc. Syllabus
CE 543-U	Biochemical Engineering Unit Operations Branch	Units 2 Theor tical 2 hr/week Tutor al 1 hr/week Pract cal - hr/week
1. Definition, introduct	ion:	(2hrs)
2. Industrial micro org	anisms and methods of separation	(2hrs)
3. Micro organisms gen	netics:	(4hrs)
1	thods of growths: patch and continuous cultures.	Breaded cu tures Solid
Fermentation's		(6hrs)
5. Single cells proteins: Microorganisms use	td, treatment of waste, production	of single cell protein. Diet
value		(3hrs)
6. Bakery yeast: Dry and ready eff	fective types.	(3hrs)
7. Amino acids: Clotamic acid.		(1hr)
8. Organic acids: Acetic acid.		
		(1hr)
9. Enzymes:		(1hr)
10. <u>Antibiotics:</u> Microorganisms pro	oducing antibiotics, Penicillin	(1hr)
11. <u>Industrial Alcohol</u> Alcoholic fermentat	<u>:</u> ion, microorganism used, productio	on method.
12. Animal cell cultiva	tion:	(2hrs)
13. Plant cell Cultivat	ion:	(1hr)
14. Environmental cle In the fields of w	eaning using biotechnology: vater treatment. Organic Pollutants.	Petroleum wastes. (2hrs)

(2hrs)

University of Technology	ersity of Technology Chemical Engineering Department	Third Year B.Sc. Sylla		
		Units	2	
CE 643-U	Environmental Engineering	Theo etical Tuto: ial	2 hr/week 1 hr/week	
	Unit Operations Branch	Practical	- hr/week	

1. Introduction:

What is pollution and a historical brief on international concern with pollution. Types of pollutants and measurements. Pollution allowable limits in the environment.

(2hrs)

2. Cleaning of air and industrial gases:

Sources and effects of air pollution. Methods of control. Designing pollution equipment. Settling chamber, Cyclone separator, Venture scrubber, pray tower, cyclone scrubber, packed bed, electrostatic precipitator. Economic evaluation of gas cleaning.

(10hrs)

3. Water and heavy water treatment:

Physical methods of treatment. Occupation methods. River water cleaning, heavy water treatment. Desalination.

(10hrs)

4. Industrial and sewage treatment:

Industrial water treatment. Sewage water treatment.

(4hrs)

5. Solid pollutants:

Sources and effects. Methods of control of solid pollutants.

(4hrs)

University	of Technology	Chemical Engineering Department	Third Y	ear 1 .Sc.	Syllabus	
				Units	2	ı
CE 543-R	Storage and T	ransport of Crude and Petroleum	Products		2 hr/week 1 hr/week	

Refinery Engineering Branch

1. Fluid Flow:

Petroleum liquids flow. Petroleum gases flow. Designing pipe network from optimum velocity and diameter viewpoint. Crude and petroleum products pipelines, Heating and pumping stations, Increasing pumping rates by looping. Increasing pumping rates by polymer additives. Two-phase flow. Unsteady state flow and hammering.

(14hrs)

- hr/week

Practical

2. Pressure Vessels / Tanks:

Stress systems of shells. Bending and membrane shells. Equilibrium e juation for a shell element and LPG tanks, Rotationally symmetrically loaded membrane shell. Crude oil huge tanks.

(8hrs)

3. Selection of Materials for Vessels and Pipelines:

Selection of materials for vessels to limit corrosion. Selection of naterials for pipelines to limit corrosion. Catholic protection.

(8hrs)

University of Technology	Chemical Engineering Department	Third Year 1:.Sc.	Syllabus
		Units	2
CE 643-R	Combustion Engineering	Theo etical Tutorial	2 hr/week 1 hr/week
	Refinery Engineering Branch	Pract cal	- hr/week
1. Combustion reaction	stoichiometry:		
. Combastion reaction	- Store Manual J.		(8hrs)
2. Excess air and flame	temperature:		
3. Combustion product	<u>s:</u>		
4. Gas, solid, and liquid	l fuels:		
5. Flammability limits:	~		(01
			(8hrs
6. Flame and combustic	on speed:		
7. Flame stability:	fuels buymays		
8. Gas, liquid and solid	ideis burners.		(6hrs
9. Combustion zones a	nd temperature profiles:		(omo
	ection rooms in furnaces:		
11. Furnaces and the tu			
12. Chimney height cal	culation:		
			(8hrs
	and heat loss calculations :		
14. Furnace wall layers	and refractories:		

15. Tube layers in furnaces:

University of Technology	Chemical Engineering Department	Fourth Year I .Sc. Syllabus
		Units 4
CE 144	Project	Theor :tical 1 hr/week Tutor al - hr/week

1. Introduction:

Literature, Methods of manufacturing, Capacity of the Process, Physical and Chemical Properties.

(4 hrs)

Pract cal 2 hr/week

2. Material and Energy Balances:

Review of Material and Energy balances, Flow sheet Symbols, PFD Information in Flow Diagram.

(6 hrs)

3. Choice of Process Devices and Equipment:

Pumps, Valves, Pipes, Measuring Devices and Controllers, Material of Construction, PID Plot.

(6 hrs)

4. Choice of Plant Location and Layout Standards:

(2 hrs)

5. Industrial Services:

Cooling Tower, Boilers, Study Problems from Industrial Waste Disposals, and How to Prevent the Environmental pollution.

(6 hrs)

6. Use a Computer Program to Design:

Introducing a Complete Project with Material and Energy Balances Design of Devices, Calculation of Costs, Choice of Location.

(6 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc.	Syllabus
		Unit:	7
CE 244	Transport Phenomena	Thec retical Tutc rial	3 hr/week 1 hr/week

1. Boundary Layer and analogies:

Boundary Layer in Laminar and turbulent Flow, Boundary Layer in Laminar sub Layer, Transition layer, Velocity distribution on Surfaces and Pipes, Monnentum, heat and mass transfer molecular diffusion, Eddy transfer, Reynolds analogy, Modified Reynolds analogy, Chilton and Colburn analogy.

(20 hrs)

Practical 3 hr/week

2. Evaporation:

Introduction, Types of Evaporations, Evaporation Equipment, Heat transfer in Evaporation Process, single, double and Multi effect Evaporators, Design of evaporators, Comparison of Forward, backward and Parallel effect evaporators, boiling Point rise.

(12 hrs)

3. Drying:

Introduction and general Principle in drying, Rate of drying, the mechanism of moisture movement, Calculation of rate of drying, moisture transport in Solids at Constant and falling rate Period, Capillary movement, Material and Energy Balances in Continuous dryers, Types of Dryers

(8 hrs)

4. Humidification, dehumidification and Cooling towers:

Humidification, dehumidification, Temperature humidity Chart for air – water system. Enthalpy – humidity –temperature chart. Addition of Vapor or liquid Stream to a gas Stream. Mechanism of dehumidification Evaluation of heat and Mass transfer Coefficient, Cooling tower, height of Packing in Cooling towers, Ninimum gas Condition, Change of Condition of liquid and gas in Cooling towers.

(14 hrs)

5. Extraction:

Definition, Extraction process, Equilateral Triangular coordinates (Ternary Diagram), system of three liquid one pair partially soluble, system of three liquid two, pairs partially soluble, choice of solvent, Equipment in extraction cross current extraction, multi-stage Cross Current extracting cross current for insoluble Liquid, Continuous Counter current extraction, Continuous Counter Current in Soluble, Liquid, Minimum Solvent, Counter Current extraction with reflux.

(22 hrs)

Note:

• 1st term

University of Technology	Chemical Engineering Department	Fourth Year 3.Sc.	Syllabus
		Units	7
CE 244	Transport Phenomena	Theo etical Tutcrial	3 hr/week
		Practical	3 hr/week

6. Filtration:

Type of Filters, Filtration theory, Plate and frame filter press, leat fi ter, filtration at Constant ΔP , Filtration at Constant rate, washing Time.

(10 hrs)

7. Crystallization:

Introduction Equilibrium Solubility in crystallization, Classification of Crystallizes, Nucleation theory, Rate of Crystal Growth, Material and Er ergy balance in Crystallization.

(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year 3.Sc.	Syllabus
		Units	5
CE 344	Process Control	Theo etical	2 hr/week
		Tuto ial	1 hr/week
		Practical	3 hr/week

1. Revision of Laplace Transformation:

(4 hrs)

2. Process Dynamic and Transient Response of the system:

Dynamic and Control For chemical Process, Dynamic behavior for 1st order system, lineraization, Interacting and non-interacting, Second order system, Transportation.

(16 hrs)

3. Characteristics of the closed loop System:

Over all closed loop transfer functions and block diagram algebra Transient of Simple Closed loops System, Stability of Control, Introduction to the Frequency analysis and design technique.

(20 hrs)

4. Industrial Controller:

Actions selection Criteria for Various Control modes, final Con rol element, Dynamics and Control of Chemical reactor System.

(10 hrs)

5. Dynamic and Control of some Complex Chemical Process:

Heat exchanger, distillation Column, Introduction to Compute Control of Chemical Process.

(10 hrs)

Note:

• 2nd term

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 444-U	Chemical Industries	Unit 6 The retical 2 hr/week Tut rial 1 hr/week
	Unit Operation Branch	Practical 3 hr/week
1. Chemical Processing Introduction, Operati	g ng Conditions, Type of Process.	
, -1	, , ,	(3 hrs)
 Sulpher and Sulpher Raw materials, Fras Production. 	ric acid: ch Process, Iead Chamber Proce	ess, Contact Process, Oleum
Froduction.		(4 hrs)
3. Ammonia and Nitric	acid:	(
	action Steps, Production Process.	
		(6 hrs)
4. Nitrogenous Fertiliza		
Ammonium Suipnate	e, Ammonium nitrate, Urea.	(4 hrs)
. Phosphoric acid:		(4 1113)
	oric and, Electric Furnace	
	•	(4 hrs)
6. Phosphate Fertilizer	<u>s:</u>	
Superphosphate, Trip	ole Superphosphate.	
		(4 hrs)
7. Electrolytic Industri	es: Caustic Soda Production Chlorine	Production
rimeiple and theory,	Causiic Soua Froduction Chlorine	(4 hrs)
3. Industrial Salts:		(· mo)
Sodash Production		
		(3 hrs)
9. <u>Ceramic Industries:</u>		~ # *** ***
Raw materials, Chem	nical Conversions, Porcelain, Fire	
10. Cement Industrie		(4 hrs)
 Cement Industries Raw materials, Portla 		
ran materials, rotti	and Comone	(4 hrs)
11. Glass Industries:		()
Raw materials, Manu		
		(4.1)

(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year 3.Sc.	Syllabus
		Unit:	6
CE 444-U	Chemical Industries		2 hr/week
	Unit Operation Branch	Tutc rial Practical	1 hr/week 3 hr/week
12. Surface Coating I			
,			(4 hrs)
13. Oils and Fats:			` ′
Fatty acids, Manufac	ture of Vegetable Oils and Waxes		
10 ² / ₂			(4 hrs)
14. Soap and Deterge	ents:		
Soap Production, Det	ergents Production		
			(4 hrs)
15. Sugar Industries:			
Cane Sugar, Beet Sug	gar		
			(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc.	Syllabus
		Unit	2
CE 134	Optimization	Thec retical Tutc rial	2 hr/week 1 hr/week
		Donation	ha/waala

1. Introduction to Optimization Methods:

(4 hrs)

2. Organization of Optimization problems:

(4 hrs)

3. Single Variable:

Analytical methods, numerical methods, graphical methods, numerical Search, restriction Function, unrestricted Function, Direct search, Dichotornous Search, golden Search, Fibonacci Search

(12 hrs)

4. Multivariables Optimization Methods:

- Necessary Conditions For Extreme Values in graphical Cases.
- Solution by graphical method.
- Simplex method.
- Linear Programming and application in Chemical engineering (transportation mixing)

(10 hrs)

			Syllabus
	Units		2
Engineering of Catalysts			
Unit Operations Branch			1 hr/week - hr/week
s of Catalysts			(2 hrs)
	ons for each St	n.	
		P	(4 hrs)
d Mechanism: niting Step.			
3 1			(4 hrs)
Gas – Solid Reactions:			5.23
			(8 hrs)
֡֡֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜	Unit Operations Branch s of Catalysts Reaction: Reaction, Decoration, Rate Equation d Mechanism: niting Step.	Engineering of Catalysts Unit Operations Branch Seaction: Reaction: Reaction, Decoration, Rate Equations for each State Equations Step. Gas – Solid Reactions:	Unit Operations Branch Tuto ial Pract cal s of Catalysts Reaction: Reaction, Decoration, Rate Equations for each Step. d Mechanism: niting Step. Gas – Solid Reactions:

Consistent with Experimental Observations, Evaluation of the Rate law Parameters, Reactor Design

(12 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc.	h Year B.Sc. Syllabus	
		Unit;	2	
CE 544-R	Engineering of Catalysts	Thecretical	2 hr/week	
		Tute rial	1 hr/week	
	Refinery Engineering Branch	Practical	- hr/week	

1. Introduction:

Properties of Catalysts, Determination of Surface Area, Void Volume and Solid Density, Pore Volume Distribution, Classification of Catalysts, Promoters and Inhibitors, Catalyst Poisoning.

(4 hrs)

2. Rate Equations of Fluid-Solid Catalytic Reactions:

Rate of Adsorption, Desorption and Surface Reaction, Rate of the limiting Step, Mechanism, Design of Reactor for- Gas- Solid Reactions. Data Analysis for Reactor Design

(10 hrs)

3. External Transport Processes in Heterogeneous Reactions:

Equations of Mass –and Heat Transfer, Calculations of the Concentrations on the Surface, Calculations of Temperature Differences in External Transport Processes, Fluidized Bed Reactors, slurry Reactors

(4 hrs)

4. Internal Transport Processes:

Gaseous Diffusion in Single Cylindrical Pores, Surface Diffusion, Effectiveness Factor, Thiele – Type Modulus

(4 hrs)

5. Design of Heterogeneous Calalytic Reactors:

Fixed - Bed Reactors, Fluidized Bed Reactor, Slurry Reactions

(8 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 644-U	Corrosion Engineering	Units 2 The retical 2 hr/week Tutorial 1 hr/week
	Unit Operations Branch	Pra tical - hr/week
1. <u>Introduction:</u> Definition of Corrosion	on, why it happens, direct and indi	rect disadvantages. (2 hrs)
2. Types of Corrosion:		(2 1113)
Dry Corrosion, wet C	orrosion, Eight form of Corrosion	
3. Mechanism of Crevio	ce Corrosion:	(4 hrs)
4. Electrochemical cell: Electrode Potential, ty	ypes of reference electrodes.	(2 hrs)
5. Corrosion rate Calcu Wet loss method, elec	lation:	(4 hrs)
6. Polarization: Definition, Types.		(2 hrs)
7. Passivity:		(4 hrs)
AND 2017/03/20 10:00/2016	e and velocity on corrosion reac	(2 hrs)
9. Method of Protection		(2 hrs)
1778 0.001	easure Corrosion rate:	(4 hrs)
aments to me	Asare Corrosion rate.	(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.Sc. Syllabus
CE 744-U	Technology of Oil and Gas	Uni s 2 The retical 2 hr/week Tutorial 1 hr/week
	Unit Operations Branch	Practical - hr/week
1. <u>History and Develop</u> Kinds of Refineries.	ment of Refinery Processes:	
STREET, STREET STREET,		(2 hrs)
2. Chemical Composition	on of Petroleum:	
2 DI		(2 hrs)
3. Physical and Thermo		
Physical and Thermo	dynamic Properties of Petroleum of	
4. Introduction to Proc	assing.	(2 hrs)
	ation, Tube still Heaters.	
otaomzation, acriyar	ation, Tuoc still Heaters.	(4 hrs)
5. Fractionation of Petr	oleum:	(4 1113)
Atmospheric and Vac		
		(2 hrs)
6. Fractionation Tower	<u>s:</u>	(/
Material and Energy Towers, Tower Diameter	Balances, Reflux, Temperature D	sistribution in Fractionation
		(6 hrs)
7. Thermal Cracking:		
Catalytic Cracking		0.0001.00
8. Natural Gas:		(6 hrs)
Refinery Gases		
Retificity Gases		(4 hus)
9. Natural Gas Processi	ng Units:	(4 hrs)

9. Natural Gas Processing Units:

(2 hrs)

University	of Technology	Chemical Engineering Department	Fourth Year	3.Sc.	Syllabus
			Unit		2
CE 234	<u>Industrial</u>	Management and Safety Requirem	ents Theo		2 hr/week 1 hr/week

1. Management:

Principle of management, types and classifications, management responsibility, organization responsibility.

(4 hrs)

Prac ical - hr/week

2. Industrial organization

Site, Feasibility study, Development of efficient work method (plant layout, flow of material, material handling), workstations, Inputs and Outputs Production planning (types of productions).

(3 hrs)

3. Maintenance:

Classifications, cost, machine replacements, case studies and example i.

(3 hrs)

4. Network Analysis:

Principles and applications, Critical Path Method (CPM), Gantt Chart, Pert techniques (examples and case studies).

(3 hrs)

5. Work Measurement Techniques

Time and Motion study.

(3 hrs)

6. Costing:

Framework of management, cost of production (row material cost, labor cost, machinery cost).

(3 hrs)

7. Quality Control:

Standardization, specification, sampling techniques, Inspections – analysis of results. Quality costs (preventive cost, appraisal cost and failure cost). A plication of quality control chart – examples, Reliability.

(4 hrs)

8. <u>ISO:</u>

Requirements, applications, ISO series, Quality management system (QMS), Total Quality managements (TQM), Requirements and applications.

(3 hrs)

9. Safety Requirements:

Hazards (types e.g. industrial hazards, pollution (air pollution, water pollution, industrial pollution). Industrial by products and industrial waste, Safety requirements of industrial sites. Requirements of suitable work environment (examples with particular emphasis in chemical industry).

(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B. Sc. Syllabus
		Units 2
CE 844-U	Polymer Technology	Theore ical 2 hr/week Tutori: 1 1 hr/week
	Unit Operations Branch	Practical - hr/week
1. Polymers:		
Nomenclature		
		(2 hrs)
2. Polymers Classificat		ad on Chamica nature of
Polymers, Technical Cla	on Sources, Classification bases	ed on Chemica nature of
		(2 hrs)
3. Polymerization:		
	ymerization (condensation pol	
	cationic, Anionic) Coordination Perization by group transfer	orymerization (Eteroregular
,		(4 hrs)
4. Polymerzation Syste		
Bulk, Solution, Susp	ension, Precipitation	(4 hrs)
5. Molecular Weight of	f Polymers:	(1110)
	ment of molecular weight of Polyn	
< p. 1		(2 hrs)
6. Polymer composites	<u>:</u>	(2 hrs)
7. Polymer blends:		(=)
		(2 hrs)
8. Physical Properties	of Polymers:	(4 hrs)
9. Mechanical Propert	ies of Polymers:	(41113)
	pact strength, Creep, hardness	62000s X
		(4 hrs)
10. Polymer Processi		
Mording, Extrusion,	Thermal Forming Calendaring	(4 hrs

(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year I .Sc. Syllabus		
CE 444 D		Units 6		
CE 444-R	Petroleum Refining	Theor tical 2 hr/week		
	Refinery Engineering Branch	Tutor al 1 hr/week Pract cal 3 hr/week		
4 Y . 1		Truck Sur 9 III WOOK		
Introduction: History and Develops	ment of Refining Processes Kinds	of Refineries.		
2. Chemical Composition of Petroleum: (2 hrs)				
		(2 hrs)		
3. Physical and Thermo	odynamic Properties of Petroleur	m oil:		
		(2 hrs)		
4. Evaluation of oil stoc	eks:			
5 Introduction to Duce		(4 hrs)		
5. <u>Introduction to Proc</u> Stabilization, dehydra				
Stabilization, deliyara	tion, tube still heaters	(10 hus)		
6. Fractionation of Petr	oleum:	(10 hrs)		
Atmospheric and Vac				
		(2 hrs)		
7. Fractionation Tower		20 20		
Material and Energy Balances, Reflux Temperate Distribution in Fractionation				
Tower, Tower Diameter		9)		
& Treating Processes		(8 hrs)		
8. <u>Treating Processes:</u> Removal of Acid Gases, Sweetening Processes, improvement in Perfo mance and				
storage stability.	ses, 5 weetening 1 tocesses, improv	ement in Perio mance and		
0		(8 hrs)		
9. Upgrading processes:				
Thermal Crackling, (Cooking, Visbreaking, Catalytic,	Cracking, Hyd ocracking,		
Catalytic Reforming, Alk	cylation, Isomerisation.	_		
10 P 1 (D) 1		(10 hrs)		
10. Product Blending: Blending of vapor pressure, octane number, viscosity, flash point ariline point,				
and pour point.	ressure, octane number, viscosity	y, mash point ar iline point,		
pour pont.		(6 hrs)		
11. Manufacturing of	lubricating oils:	(0 1113)		
		(4 hrs)		
12. Treating process u	sing Hydrogen:	,/		
		(0.1.)		

(2 hrs)

University of Technology	Chemical Engineering Department	Fourth Year I .Sc. Syllabus
CE 744-R	Gas Technology	Units 2 Theoretical 2 hr/week Tutor al 1 hr/week
	Refinery Engineering Branch	Practi:al - hr/week
1. Classification of Gas	eous fuels thermodynamic:	
		(3 hrs)
2. Physical and Chemic	cal Properties for natural Gas:	
1 D 1 10 -	8	(3 hrs)
3. Drying of Gaseous F		
Drying by Adsorption	n, Drying by Absorption	
4. Sweetening of Gaseo	us Fuels:	(4 hrs)
Sweetening by Adsor	ption, Sweetening by Absorption	
		(6 hrs)
5. Gas Processing units	<u>:</u>	` ′
Flash Separation Proc	cesses, Distillation of multicompor	nemt gas mixture
	•	(6 hrs)
6. <u>Liquefied Petroleum</u>	Gases:	60.0 (30.333310)
		(2 hrs)
7. Production of Hydro	gen:	
8. Sulfur recovery processes from Acid gases streams:		

Tutorial 1 hr/wee **Refinery Engineering Branch 1. **Introduction:** Chemical Process Industries** Chemical Process Industries** 2. **Basic Petrochemical Materials:** Low Olefins, Steam Cracking Processes Cooling, Purification Processes Cooling Processes, Purification Processes Compression, Separation Operations, Energy System, Diolefins, Sources of diolefins, Butadiene Separation, Isobutylen Production, Conversion Process For Production of diolefins, Hig ier Olefing Production methods of higher olefins, linear, Alkyl benzene Complex Aromatic Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation** 3. **Synthetic Zeolites:** 4. **Intermediate Petrochemicals:** Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Et nylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid.** 5. **Petrochemical Detergents:** (2 hr	University of Technology	Chemical Engineering Department	Fourth Year B Sc. Syllabus
Tutorial 1 hr/wee **Refinery Engineering Branch 1. **Introduction:** Chemical Process Industries** Chemical Process Industries** 2. **Basic Petrochemical Materials:** Low Olefins, Steam Cracking Processes Cooling, Purification Processes Cooling Processes, Purification Processes Compression, Separation Operations, Energy System, Diolefins, Sources of diolefins, Butadiene Separation, Isobutylen Production, Conversion Process For Production of diolefins, Hig ier Olefing Production methods of higher olefins, linear, Alkyl benzene Complex Aromatics Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation** 3. **Synthetic Zeolites:** 4. **Intermediate Petrochemicals:** Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Et nylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid.** 5. **Petrochemical Detergents:** C2 hr.* 6. **Polymers:** LDPE, HDPE, PP, PVC** (4 hr.* 7. **Synthetic Fibers:**			
1. Introduction: Chemical Process Industries Chemical Process Industries 1. Basic Petrochemical Materials: Low Olefins, Steam Cracking Processes Cooling, Purification Processes Cooling Processes, Purification Processes Compression, Separation Operations, Energ System, Diolefins, Sources of diolefins, Butadiene Separation, Isobutylen Production, Conversion Process For Production of diolefins, Higher Olefin Production methods of higher olefins, linear, Alkyl benzene Complex Aromatic Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation 3. Synthetic Zeolites: 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, Vinylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Ethylben Zen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid. (7 hr. 5. Petrochemical Detergents: (2 hr. 7. Synthetic Fibers:	CE 644-R	Petrochemical Industries	Theoretical 2 hr/week Tutorial 1 hr/week
Chemical Process Industries 2. Basic Petrochemical Materials: Low Olefins, Steam Cracking Processes Cooling, Purification Processes Coolin Processes, Purification Processes Compression, Separation Operations, Energ System, Diolefins, Sources of diolefins, Butadiene Separation, Isobutylen Production, Conversion Process For Production of diolefins, Higher Olefins Production methods of higher olefins, linear, Alkyl benzene Complex Aromatics Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation 3. Synthetic Zeolites: (2 hrs. 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Ethylenzen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid. (7 hrs. 5. Petrochemical Detergents: LDPE, HDPE, PP, PVC (4 hrs.) 7. Synthetic Fibers:		Refinery Engineering Branch	Practi al - hr/week
2. Basic Petrochemical Materials: Low Olefins, Steam Cracking Processes Cooling, Purification Processes Cooling Processes, Purification Processes Compression, Separation Operations, Energ System, Diolefins, Sources of diolefins, Butadiene Separation, Isobutylen Production, Conversion Process For Production of diolefins, Higher Olefins Production methods of higher olefins, linear, Alkyl benzene Complex Aromatics Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation 3. Synthetic Zeolites: 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, Vinylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Ethylene Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid. (7 hr. 5. Petrochemical Detergents: LDPE, HDPE, PP, PVC (4 hr. 7. Synthetic Fibers:	1. Introduction:		
2. Basic Petrochemical Materials: Low Olefins, Steam Cracking Processes Cooling, Purification Processes Coolin Processes, Purification Processes Compression, Separation Operations, Energ System, Diolefins, Sources of diolefins, Butadiene Separation, Isobutylen Production, Conversion Process For Production of diolefins, Higher Olefin Production methods of higher olefins, linear, Alkyl benzene Complex Aromatics Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation 3. Synthetic Zeolites: 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, Vinylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Ethylene Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid. 5. Petrochemical Detergents: (2 hr 6. Polymers: LDPE, HDPE, PP, PVC (4 hr 7. Synthetic Fibers:	Chemical Process Inc	dustries	(1 hrs)
Low Olefins, Steam Cracking Processes Cooling, Purification Processes Cooling Processes, Purification Processes Compression, Separation Operations, Energ System, Diolefins, Sources of diolefins, Butadiene Separation, Isobutylen Production, Conversion Process For Production of diolefins, Higher Olefing Production methods of higher olefins, linear, Alkyl benzene Complex Aromatics Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation 3. Synthetic Zeolites: 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, EthylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid. 5. Petrochemical Detergents: Ca hr. Ca hr.	2 Pacia Patrachamical	Materials:	(1 1113)
Processes, Purification Processes Compression, Separation Operations, Energ System, Diolefins, Sources of diolefins, Butadiene Separation, Isobutylen Production, Conversion Process For Production of diolefins, Higher Olefins Production methods of higher olefins, linear, Alkyl benzene Complex Aromatics Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation 3. Synthetic Zeolites: (2 hrs. 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, EthylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid. (7 hrs. 5. Petrochemical Detergents: LDPE, HDPE, PP, PVC (4 hrs.) 7. Synthetic Fibers:	Low Olefins, Steam	n Cracking Processes Cooling, Pur	rification Processes Cooling
System, Diolefins, Sources of diolefins, Butadiene Separation, Isobutylen Production, Conversion Process For Production of diolefins, Higher Olefin Production methods of higher olefins, linear, Alkyl benzene Complex Aromatics Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation (10 hrs.) 3. Synthetic Zeolites: (2 hrs.) 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Ethylene Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Γerephthal acid. (7 hrs.) 5. Petrochemical Detergents: (2 hrs.) 6. Polymers: LDPE, HDPE, PP, PVC (4 hrs.)	Processes. Purification	Processes Compression, Separ	ration Operations, Energy
Production, Conversion Process For Production of diolefins, Higher Olefins Production methods of higher olefins, linear, Alkyl benzene Complex Aromatics Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation 3. Synthetic Zeolites: (2 hrs. 4. Intermediate Petrochemicals:	System, Diolefins, S	Sources of diolefins, Butadien	e Separation, Isobutylene
Production methods of higher olefins, linear, Alkyl benzene Complex Aromatic Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation (10 hrs.) 3. Synthetic Zeolites: (2 hrs.) 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Et nylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid. (7 hrs.) 5. Petrochemical Detergents: (2 hrs.) 6. Polymers: LDPE, HDPE, PP, PVC (4 hrs.)	Production, Conversio	n Process For Production of	diolefins, Higher Olefins
Basic materials Aromatic Sources, Separation of benzene – Toluene – Xylene Syntheses Gas, Steam reforming, Partial Oxidation 3. Synthetic Zeolites: (2 hrs. 4. Intermediate Petrochemicals:	Production methods of	higher olefins, linear, Alkyl be	nzene Complex Aromatics,
3. Synthetic Zeolites: (2 hrs. 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Et nylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Γerephthal acid. (7 hrs.) 5. Petrochemical Detergents: LDPE, HDPE, PP, PVC (4 hrs.) 7. Synthetic Fibers:	Basic materials Aron	natic Sources, Separation of ben	zene – Toluene – Xylene,
3. Synthetic Zeolites: 4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Et nylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Ferephthal acid. (7 hr. 5. Petrochemical Detergents: (2 hr. 6. Polymers: LDPE, HDPE, PP, PVC (4 hr. 7. Synthetic Fibers:	Syntheses Gas, Steam re	eforming, Partial Oxidation	. (101)
4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Et nylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Γerephthal acid. (7 hr. 5. Petrochemical Detergents: (2 hr. 6. Polymers: LDPE, HDPE, PP, PVC (4 hr. 7. Synthetic Fibers:			(10 nrs)
4. Intermediate Petrochemicals: Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, V nylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Et nylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Γerephthal acid. (7 hr. 5. Petrochemical Detergents: (2 hr. 6. Polymers: LDPE, HDPE, PP, PVC (4 hr. 7. Synthetic Fibers:	3. Synthetic Zeolites:		(2 hrs)
Methanol, Ethylene Oxide Acetic Acid, Ethanolamine, Vinylchlorid Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Et iylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Γerephthal acid. (7 hr. 5. Petrochemical Detergents: (2 hr. 6. Polymers: LDPE, HDPE, PP, PVC (4 hr. 7. Synthetic Fibers:	4 Introducto Datum	shamiaalar	(2 1113)
Ethyleneglycol Acrylonitrile, Adipic Acid, Methyl Tetrabutyl ether, Et nylbenZen Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, lerephthal acid. (7 hr. 5. Petrochemical Detergents: (2 hr. 6. Polymers: LDPE, HDPE, PP, PVC (4 hr. 7. Synthetic Fibers:	Methanol Ethylet	ne Ovide Acetic Acid Eth	anolamine, V nylchloride,
Styrene, Phenol, Nitrobenzene, Ethylene, Cyclohexane, Benzoic acid, Terephthal acid. (7 hr. 5. Petrochemical Detergents: (2 hr. 6. Polymers: LDPE, HDPE, PP, PVC (4 hr. 7. Synthetic Fibers:	Ethyleneglycol Acrylor	nitrile. Adipic Acid. Methyl Teta	rabutyl ether, Et nylbenZene,
acid. 5. Petrochemical Detergents: (2 hr 6. Polymers: LDPE, HDPE, PP, PVC (4 hr 7. Synthetic Fibers:	Styrene, Phenol, Nitro	benzene, Ethylene, Cyclohexane	, Benzoic acid, Terephthalic
5. Petrochemical Detergents: (2 hr 6. Polymers: LDPE, HDPE, PP, PVC (4 hr 7. Synthetic Fibers:			
6. Polymers: LDPE, HDPE, PP, PVC (4 hr 7. Synthetic Fibers:			(7 hrs)
6. Polymers: LDPE, HDPE, PP, PVC 7. Synthetic Fibers:	5. Petrochemical Dete	rgents:	(0.1)
LDPE, HDPE, PP, PVC 7. Synthetic Fibers:			(2 hrs)
7. Synthetic Fibers:		210	
7. Synthetic Fibers:	LDPE, HDPE, PP, F	VC	(4 hrs)
/. Synthetic Fibers:	7 Counthatia Eibana		(4 1113)
	7. Synthetic Fibers:		(2 hrs)

(2 hrs)

8. Petrochemical Complexes:

		Units 2
CE 844-R	Environmental Pollution	Theoretical 2 hr/week Tutorial 1 hr/week

Chemical Engineering Department

Refinery Engineering Branch

1. Introduction:

University of Technology

Definition of Pollution, literature surveys on international interest of Environmental Pollution Problems. Pollutants, Type of Pollutants, Measuring Devices, Allowable level of Pollutants in Environment.

(2 hrs)

Fourth Year B.Sc. Syllabus

Practical - hr/week

2. Air Pollution:

Air Pollutants, Particulate matter, SO₂, CO, NO, ...etc Sources and effect, Control Treatment Methods, Design of some Equipment: Centrifugal, Cyclone Scrubber, Packed Bed Ventura, Spray – tower, Electrostatic Separator, and Chimney Design.

(10 hrs)

3. Water - Pollution:

Water Pollutants, Sources and effect, Organic and non – organic, Radiation activity, Suspended Solids and Sculling Salts, Control Treatment of Water Pollution for: Drinking, Industry Processes, Desalination methods, Reverse Osmosis, Electrodalysis, Solar Distillation, Primary, secondary and advanced treatment of Wastewater, Treatment of oil Pollution.

(12 hrs)

4. Pollution of Solid - Waste:

Sources, Effects, Control Treatment method of Removal Solid – Waste, Treatment by economy methods.

(6 hrs)

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