MINISTRY OF HIGHER EDUECTION

University Of Technology

Chemical Engineering Department



B.Sc PROGRAMME IN CHEMICAL ENGINEERING AND M.Sc. PROGRAM

OUTLINE OF
SYLLABUSES ALLOC3ATION OF SUBJEC'TS
AND
WEEKLY LOAD

2007-2008

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INTRODUCTION

Chemical Engineering is distinguished from other branches of er gineering by its strong dependence on chemistry. This enables the chemical engineer o understand properties involving changes in physical state, chemical composition, or energy content for systems ranging in scale from molecules to full sized manufacturing plants.

The chemical engineer may be employed in an established indus ry producing chemicals, petroleum products, petrochemicals, pharmaceuticals, syrthetic fibers, foods, plastics or metals. These products are steadily needed in an increasing amounts due to the expanding growth of population. His function may be involved in making innovations in operations, doing research, development and analysis of existing plants, technical services, or scale up.

Due to his board knowledge, the chemical engineer often occupies a dominant position in the above mentioned industries.

He also apply his knowledge in such diverse areas as air and water pollution or biochemical research.

The formal course work for B.Sc. programme involves mathematics, including both analysis and computer application, chemistry, mechanics, electricity chemical engineering principles and practice subjects including fluid dynamics, heat, unit operation, chemical reactor design, thermodynamics, and chemical processes. In addition the programme involves English Language which is essential for any engineering student to enable him/her to benefit from thousands of textbooks and periodicals.

A considerable emphasis is laid, in this department, on practical training and laboratory work, which helps students in getting better understanding of the theoretical part of their curriculum.

Moreover students are required to spent twelve weeks, training in industry during summer vacations.

This booklet contains the academic programme and the syllabus for subjects which are taken by the chemical engineering students during their four years study.

Prof. M. A. Zablouk Chairman of Chemical Eng. Dept.

Contents

B.Sc. Programme

		Page
	First Year Syllabus	
_	English	10
_	Mathematics	12
_	Principle of Chemical Engineering	13
-	Chemistry	14
_	Engineering Drawing and AutoCAD	16
_	Mechanics and Strength of Materials	18
_	Electrical Technology	20
_	Computer Science	21
_	Workshop	23
-	Workshop	
	Second Year Syllabus	
35 25 <u>22</u> 5	Human Rights	24
	Mathematics	25
	Fluid Flow	26
-	Properties of Materials	28
- 54 127	Computer Programming & MATLAB	29
20	Fuels Technology	32
-	Thermodynamics	33
-	Physical Chemistry	35
-	1 mysicar Chemistry	
	Third Year Syllabus	
	Democracy	36
_	Applied Mathematics	37
-	Mass Transfer	38
-	Statistics and Instruments Measurement	39
-	Reactor Design	40
-	Heat Transfer	41
-	Equipment Design by Computer	42
-	Biochemical Engineering	44
_	Environmental Engineering	45
-	Storage and Transport of Crude and Petroleum Products	40
-	Combustion Engineering	47
_	A THILDHAUDH PARAIRCELINE	

		Page
	• Fourth Year Syllabus	40
-	Project	48
_	Transport Phenomena	49
_	Process Control	51
_	Chemical Industries	52
-	Optimization	54
_	Engineering of Catalysts	55
_		57
_		58
_		59
	Table 19 19 19 19 19 19 19 19 19 19 19 19 19	60
	•	61
2) 2)		62
7		63
-		64
	Corrosion Engineering Technology of Oil and Gas Industrial Management and Safety Requirements Polymer Technology Petroleum Refining Gas Technology Petrochemical Industries Environmental Pollution	58 59 60 61 62 63

				Hours	/ Week			Units
Subject Symbol	Subject	F	irst Term		Second Term			
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 111	English Language	2	-	-	2	-	-	4
CE. 121	Mathematics	3	-	1	3	-	1	6
CE. 141	Principle of Chemical Engineering	3	-	1	3	-	1	6
CE. 221	Chemistry	3	2	1	3	2	1	8
CE. 131	Engineering Drawing and AutoCAD	1	2	-	1	2	-	4
CE. 231	Mechanics and Strength of Materials	2	-	1	2	-	1	4
CE. 331	Electrical Technology	1	-	1	1	-	1	2
CE. 321	Computer Science	2	2	-	2	2	-	6
CE. 431	Workshop	-	6	-	-	6	-	4
	Total	17	12	5	17	12	5	44

Note: The student is required to complete twelve weeks summer training in industry

Number of Hours / Week

First Term	Second Term
34	34

Key of subject symbol (from lift):

CE: Chemical Engineering

First No.: Order of subject within specialization

Second No.: specialization

1: Human subject

2 : Basic subject

3: General subject

4 : Specific subject

Third No.: Year of Study

-U: Unit Operation of Chemical Industry Branch

-R: Refinery Engineering of Oil and Gas Branch

		Hours / Week							
Subject Symbol		F	irst Term		Se	Units			
		Theoretical	Practical	Tutorial	Theoretical	Practical	1 utorial		
CE. 112	Human Rights	2	-	-	2	-	-	4	
CE. 122	Mathematics	2	-	1	2	-	1	4	
CE. 142	Fluid Flow	3	2	1	3	2	1	8	
CE. 132	Properties of Materials	1	-	1	1	2	1	3	
CE. 222	Computer Programing & MATLAB	2	2	-	2	2	-	6	
CE. 242	Fuels Technology	1	2	1	1	-	1	3	
CE. 342	Thermo dynamics	2	-	1	2	-	1	4	
CE. 232	Physical Chemistry	2	2	1	2	2	1	6	
	Total	15	8	6	15	8	6	38	

First Term	Second Term
29	29

A-Unit Operation of Chemical Industry Branch

				Hours	/ Week			
Subject Symbol	Subject	First Term			Second Ter n			Units
	Therwody	Theoretical	Practical	Tutorial	Theoretical	Practica	Tutorial	
CE. 113	Democracy	32	-	21	32	(-)	21	X
CE. 133	Applied // Mathematics	3	-	1	3	-	1	6
CE. 143	Mass Transfer	3	2	1	3	-	1	7~
CE. 233	Statistics and Instruments Measurement,	1	2	1	1	+	1	3 4
CE. 243	Reactor Design	3	-	. 1	3	(-)	1	6
CE. 343	Heat Transfer	3	-	1	3	2	1	7
CE. 443	Equipment Design by Computer	2	2	1	2	2	1	6
CE. 543-U	Biochemical Engineering	2	-	1	-	-	-	2
CE. 643-U	Environmental Engineering	-	-	-	2	-	1	2
	Total	19	5	7	19	5	7	43

First Term	Second Term
31	31

B-Refinery Engineering of Oil and Gas Branch

6-11				Hours	/ Week			
Subject Symbol	Subject	First Term			Second Tel m			Units
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 113	Democracy	2	-	(-)	2	-	-	4
CE. 133	Applied Mathematics	3	-	1	3	-	1	6
CE. 143	Mass Transfer	3	2	1	3	-	1	7
CE. 233	Statistics and Instruments	1	1	1	1	1	1	3
CE. 243	Reactor Design	2	-	1	2	-	1	4
CE. 343	Heat Transfer	3	-	1	3	2	1	7
CE. 443	Equipment Design by Computer	2	2	1	2	2	1	5
CE. 543-R	Storage and Transport of Crude and Petroleum Products	-	-	-	2	1-	1	2
CE. 643-R	Combustion Engineering	2	-	1	-	-	-	2
	Total	19	5	7	19	5	7	43

First Term	Second Term
31	31

A-Unit Operation of Chemical Industry Branch

University OF Technology

				Hours	/ Week			
Subject Symbol	Subject	First Term			Second Tel m			Units
		Theoretical	Practical	Tutorial	Theoretical	Practic: 1	Tutorial	
CE. 144	Project	1	2	-	1	2	-	4
CE. 244	Transport Phenomena	3	3	1	3	-	1	7
CE. 344	Process Control	2	-	1	2	3	1	5
CE. 444-U	Chemical Industries	2	3	/1	2	3	1	6
CE. 134	Optimization	-	-	-	2	-	1	2
CE. 544-U	Engineering of Catalyst	2	-	1	-	-	-	2
CE. 644-U	Corrosion Engineering	-	-	-	2	-	1	2
CE. 744-U	Technology of Oil and Gas	-	-	-	2	-	1	2
CE. 234	Industrial Management and Safety Requirements	2	-	1	-	-	-	2
CE. 844-U	Polymers Technology	2	-	1	-	-	-	2
	Total	14	8	6	14	8	6	34

First Term	Second Term
28	28

B- Refinery Engineering of Oil and Gas Branch

		Hours / Week						
Subject Symbol	Subject	First Term		Second Term			Units	
		Theoretical	Practical	Tutorial	Theoretical	Practical	Tutorial	
CE. 144	Project	1	2	-	1	2	-	4
CE. 244	Transport Phenomena	3	3	1	3	-	1	7
CE. 344	Process Control	2	-	1	2	3	1	5
CE. 444-R	Petroleum Refining	2	3	1	2	3	1	6
CE. 134	Optimization	-	-	-	2	-	1	2
CE. 544-R	Engineering of Catalyst	2	-	1	-	-	-	2
CE. 644-R	Petrochemical Industries	2	-	1	-	-	-	2
CE. 744-R	Gas Technology	-	-	-	2	-	1	2
CE. 234	Industrial Management and Safety Requirements	2	-	-2	2	-	1	2
CE. 844-R	Environmental Pollution	2	-	1	-	-	-	2
	Total	14	8	6	14	8	6	34

First Term	Second Term
28	28

University of Technology	Chemical Engineering Department	First Year B.S : Syllabus
CE 111	<u>English</u>	Units 4 Theoreti al 2 hr/week Tutorial - hr/week Practic: 1 - hr/week
6-Prepositions: A general idea		(1hrs)
7-Conjunctions A-Main Conjunction	ons	
B-Subsidiary 8-Interjections:		(1hrs)
Common ones		(1hrs)
Second Semester 1- Comprehension Reading Scientific Pa	issages	
1-Electricity in Ea 2-Comets 3-Exploring the m	rly Days	(16hrs)
2- Grammatical Points	<u>s</u>	
1-Articles A-definite B-indefinite		(2hra)
2-The Sentence A-Structure		(2hrs)
B-Types C-Main Parts:	Subject	
2-P	redicate	
D-Expansion		(8hrs)
3-Written and Oral Tr	ranslation	(4hrs)

University of Technology	Chemical Engineering Department	First Year B. Sc.	Syllabus
		Units	6
CE 121	Mathematics	Theoretical	3 hr/week

1- Revision:

Equation of the straight line, Trigonometric functions and their sketches, Domain, Range, Inverse of functions, Absolute value, limits, Limits, applications, Polar coordinates (general definition) Conic sections (general definition).

(12 hrs)

Pract cal

2- Differentiation and Integration:

Algebraic functions and trigonometric functions, Application (limits, Hop tals rule)
(6 hrs)

3- Transcendental Functions:

Inverse of Trigonometric functions, Natural logarithmic Exponential and Power functions (Definitions, Properties, Sketches, Derivatives and Integrals.

(12 hrs)

4- Hyperbolic Functions:

Definitions, Properties, Sketches, inverse and Derivatives and Integrals.

(8 hrs)

5- Determinants:

Definitions, Properties, Solution of systems of equations (Cramers Rule)

(6 hrs)

6- Application of Definite Integrals:

Area between two curves, Volumes, Length of curves, Surface area.

(12 hrs)

7- Moods of integration:

Substitution, Integration by parts, The Substitution $U = \tan \frac{X}{2}$, Power of trigonometric functions, Integrals involve $\sqrt{a^2 - \chi^2}$, $\sqrt{a^2 + \chi^2}$, $\sqrt{\chi^2 - a^2}$ integrals of $ax^2 + bx + c$, Partial functions, Improper Integrals (conv and dity)

(16 hrs)

8- Complex Numbers:

Definitions, Argands diagram, Addition and subtraction and multiplication and dividing numbers, Trigonometric and exponential formula of complex number, Demoivors theorem, Roots of equation, Power of complex numbers.

(9 hrs)

9- Vector Analysis:

Definition, Addition and subtraction of vectors, Multiplication by scalars, Direction, products of vector (scalar product and vector product).

(9 hrs)

University of Technology	Chemical Engineering Department	First Year B. Sc.	Syllabus
		Units	6
CE 141	Principle of Chemical Engineering	Theore ical	3 hr/week
		Tutori d	1 hr/week

1- Concept of Chemical Engineering:

Units, Dimensions, Conversion factors, Conversion equation, Temperature, Pressure, Composition, Chemical Analysis, Chemical equations.

(24 hrs)

2- Material Balance Calculation:

Without and with chemical reaction, Recycle, By pass and purge calculations for steady state, Combustion Calculation.

(30 hrs)

3- Ideal Gas laws:

Definitions of Ideal gas, Real gas, Ideal gases mixture, Real gas mixture.

(9 hrs)

4- Energy Balance:

General energy balance, Enthalpy, Heat capacities their predications and variation with temperature, Heat effects.

(18 hrs)

5- Simultaneous, Mass and Energy Balance:

(9 hrs)

University of Technology	Chemical Engineering Department	First Year B. Sc. Syllabus	
		Units 8	
CE 221	Chemistry	Theore ical 3 hr/week	

Part -1- Analytical Chemistry

1- Introduction:

Atomic weight, Molecular formula, Chemical equations, Mole concept, and Chemical equilibrium.

(4 hrs)

1 hr/week

Practi al 2 hr/week

Tutori: 1

2- Solution:

Definition, Preparation and properties, Molarity, Normality, Formality, PH, POH, Solubility

(8 hrs)

3- Analytical Methods of Analysis:

a-Qualitative Analysis

b-Quantitative Analysis.

- Volumetric (titrimetric) & analysis,
 Acid- base, Redox, Precipitation, Complex titration, Methods of calculation,
 Titration curves
- Gravimetric Analysis
 Precipitation reactions, Direct and indirect methods of analysis, Ksr.
- iii. instrumental Methods of Analysis
 Photometric,, Colourimetic, Potentiometric titration, , PH-measurements
 Chromatography, Atomic absorption.

(33 hrs)

University of Technology	Chemical Engineering Department	First Year B.5 c. Syllabus
CE 221	Chemistry	Units 8 Theoret cal 3 hr/week Tutoria 1 hr/week Practic al 2 hr/week
Part -2- Organic Chem	istry	
P-Aliphatic Compounds: Nomenclature, Preparation a) Alkanes b) Alkenes	ion of organic compounds. on, Properties, Reactions.	(3 hrs)
	res: Alkyl halides Alcohol's, Eth cids, Amines, Mercaptans, Ca nes, Disulphide.	
- Aromatic Compounds:		(20 1113)
Structural formula of ben reactions; Nitration, Sulp	zene ring, Nomenclature, Preparate Phonation, Halogenation, Alkylation Chemical reactions of – Toluene es, Naphthalene.	on, Benzene- homologues;

(3 hrs)

5- Introduction to Polymer and Carbohydrate:

(5 hrs)

6- Organo-Metallic Compounds:

(4 hrs)

University of Technology	Chemical Engineering Department	First Year B. Sc. Syllabus
CE 131	Engineering Drawing and AutoCAD	Units 4 Theore ical 1 hr/week Tutorial - hr/week Practi al 2 hr/week
Part -1- Engineering D	Prawing	
	bout engineering drawing and its used in drawings, Hidden lines.	s importance. Drawing
2 Coometrical construct	ione	(2 hrs)
2- Geometrical construct	1011:	(2 hrs)
3- <u>Pictorial Drawings</u> : (Isometric and Oblique) 4- <u>Projection Drawings:</u>		(2 hrs)
First and Third Angle Pr	rojection.	(4.1)
5- The Finding of 3 rd vie	ew:	(4 hrs)
		(6 hrs)
5- <u>Sectioning:</u>		(6 hrs)
- <u>Fasteners:</u>		0. * ±3,000,000 * 1
- Assembly and detail di	rawing:	(2 hrs)
2 22000 morr und detail di	**************************************	(A brs)

(4 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc.	<u>Syllabus</u>
		Units	4
CE 131	Engineering Drawing and AutoCAD	Theoretical	1 hr/week
		Tutoria	- hr/week
		Practical	2 hr/week

Part -2- AutoCAD

- 1- Definition of AutoCAD programs, its applications, and explain the connection between its program and engineering drawing.
- 2- Explain in details the command of AutoCAD drawing (line, circle, ...at 2.), and giving the applications examinations.
- 3- Explain the drawing of projections and sections (2D) in engineering shape, using the command of Layer and Dimensions.
- 4- Giving a simple idea for three dimensions drawing (3D).
- 5- Forming a simple shape for three dimensions drawing (3D).
- 6- Explain the method of forming for three dimensions shape, and giving the applications examinations.

University of Technology	Chemical Engineering Department	First Year B.5 c. Syllabus
CE 231 <u>M</u>	echanics and Strength of Materials	Units 4 Theoret cal 2 hr/week Tutoria 1 hr/week Practical - hr/week
A-Mechanics		
1- Principles of static's:		(2 hrs)
2- Resultants of Force sys	stems:	
3- Equilibrium of force sy	vstems:	(5 hrs)
		(5 hrs)
4- <u>Friction:</u>		(6 hrs)
5- Centurions and Center	rs of Gravity:	(5 hrs)
6- Moment of Inertia:		
7- Principle of Movement	1	(4 hrs)
		(3 hrs)
B-Strength of Materials	<u>s</u>	
1- Internal Forces in Rigi	d Bodies:	(1 hms)
2- Definition of stress and	l strain:	(1 hrs)
Types of stresses and stra		(1 hrs)
3- Stress- Strain Diagram		(1 mo)
Stress- strain diagrams for	or ductile and brittle materials	(2 hrs)
4- Proportional limits:	ecticity placticity Toughness Pacil	ience Hardness
Working stress.	asticity, plasticity, Toughness, Resil	100 20 20
		(2 hrs)

University of Tec	hnology	Chemical Engineering Department	First Year B.S c. S	Syllabus
CE 231	Me	echanics and Strength of Materials	Units Theoret cal Tutoria Practical	4 2 hr/week 1 hr/week - hr/week
5- Hook law:				
6- Poisson's ratio	Compo	eite etresses		(2 hrs)
0-1033011-3-12110	, compo	isite stresses.		(4 hrs)
7- Volumetric str	ess, Bull	k modules:		(4 hrs)
8- Thin walled C	ylinders:			(4 1115)
0. Th				(4 hrs)
9- Thermal stress	8:			(4 hrs)
10- shear and Mor	ments in	Beam:		
Deflection				(4 hrs)
11- Wind Effect of	n the Tov	wer high Tower:		(41113)
				(2 hrs)

	University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
	CE 331	Electrical Technology	Units 2 Theoret cal 1 hr/week Tutoria 1 hr/week Practic 1 - hr/week
	1- Semiconductors Equipm	ment:	(3 hrs)
- -	2- Rectifiers and Detectors	<u>s:</u>	(2 hrs)
	3- Electronic Amplifiers: 4- D.C. Circuits:		(4 hrs)
	5- <u>D.C. Generators and M</u>	otors:	(2 hrs) (3 hrs)
, ,	6- A.C Circuits:		(3 hrs)
	7- <u>Polyphases Circuits:</u> 8- <u>Transformers and Indu</u>	ation Motors:	(2 hrs)
ı)- <u>Starters</u> :	ction Motors.	(3 hrs)
10	- Integrated Circuits, Me	asuring Instruments, Tranducer	s, Transmitter: (2 hrs) (4 hrs)
11	- Electrical Heating Appl	iance:	(2 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus
CE 321	Computer Science	Units 6 Theoretical 2 hr/week Tutoria - hr/week Practic 1 2 hr/week
I- <u>Introduction:</u> Component (hard ware), co	omputer work, computer, compositi	on, Application (2 hrs)
2- <u>Files:</u> Definition, Types, Nomeno	clature, MSDos, External and intern	
3- Intoduction to window Desktop, Mouse, my comp	v operating system: outer-icons, close window, stand by.	(2 hrs)
I- <u>Folders:</u> Size and cascade, windo ile or folder copy file or fo	ws folder construction, construction	n choose file or 'older find
5- Start up:	n windows, format & Test Floppy d	(2 hrs)
Ielp.	The state of the s	(2 hrs)
	Open, Close, Save, Save as, Page Paste, Clear, Select All, Find, Repl	
7- Menu Table:	Tubic, Crour, Bereet Frin, Fring, Feep	(2 hrs)
View (zoom, Header an umbers, Textbox, Object) Format (Paragraph, For Table (shading, Colum	d footnote, Tool bar Insert (Picture nt, Borders, Bullets & numbering Sp ns, Insert, Table Insert Rows, Lang elect column) (Table Autoformat Sc	pelling and grar ımar guage, Delete, Cells, Split
3- Program Excel:		(6 hrs)
Open, Taskbar, Toolbar, (Delete), Save (Save as). Auto summation, Insert,	Format, Sheet, Cell, Construction of Close, cascade, Copy and Paste, A equation, Format sheet, Auto format	dd, Delete, Finc, Replace,
Draw.		(12 hrs)

(12 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. Syllabus

CE 321

Computer Science

Units 6
Theoretical 2 hr/week
Tutorial - hr/week
Practical 2 hr/week

9- Power Point:

Definition

FILE: New, Open, Open, Close, Save, Save as, Page set up, Print, Exit)

EDIT: Undo, Repeat, Cut, Copy, Paste, Select All, clear Duplicate, Delete slide,

find, Replace

FORMAT: Font, Alignment, Text Direction slide, Layout, Background, Apply Slid Show, View show, Rehearse timing design set up show, present Animation, Animation Preview, Slide Transition).

(11 hrs)

10- Auto Cad:

Introduction, define screen draw, select point in the screen Limits, Stat 1s, Circle, Erase, Line, Color, Point, zoom, Redraw, Arc, Ellipse, Polygon, Donut, Solid, Fill Break, Copy, More, Pan, Mirror Trim, Extend, Charge, Line type Ltscale, Dis Area, Trace, Hatch text, Snap, Grid, Rotate, Cops, Ortho, Fillet, Dim Layer.

(14 hrs)

11- Internet:

(6 hrs)

University of Technology	Chemical Engineering Department	First Year B.Sc. S	Syllabus
		Units	4
CE 431	Workshop	Theoret cal Tutoria	hr/weekhr/week
		Practic al	6 hr/week

Refer to Centralize Curriculum

University of Technology	Chemical Engineering Department	Second Year B.S. Syllabus
		Units 4
CE 112	Human Rights	Theoretical 2 hr/week Tutorial - hr/week
		Practice 1 - hr/week

Standard Syllabus

Chemical Engineering Department	Second Year B.S : Syllabus
Mathematics	Units 4 Theoreti al 2 hr/week Tutorial 1 hr/weel Practic: 1 - hr/weel

Function of two or more variables, limits and continuity, partial derivatives, chain rule, Gradients, Directional derivatives, Higher order derivatives, Maxima, minimum and saddle points, Lagrange multipliers.

(10hrs)

Multiple integrals:

Double integrals, Area, Triple integrals in rectangular coordinates, Physical application of double and triple integration.

(9hrs)

3. Ordinary Differential Equations:

Solution of first order ordinary differential equations, Solution of second order ordinary differential equations, and higher order differential equation.

(9hrs)

4. Infinite Series:

Power series of functions, Taylor's theorem, Integration, Differentiation, Fouries series, Even and odd functions, Half range expansion, Periodic functions.

(8hrs)

5. Function and Definite Integrals:

The error function, The gamma function, The beta function, factorial function. (6hrs)

6. Vector Analysis:

Products of three vectors, Equation of lines and planes.

(5hrs)

7. Complex Algebra:

Continuity, Derivation of Complex Variables, Analytic function, Integration of functions of complex variables and cauchy's theorem.

(5hrs)

8. Matrices:

The Matrix, Matrix algebra, The transpose of matrix, The inverse of matrix, Eigen Values and Eigen vectors, solution of linear equation by matrix.

(8hrs)

University of Technology	Chemical Engineering Department	Second Year B.S : Syllab	us
		Units 8	
CE 142	Fluid Flow	Theoreti al 3 hr/we Tutorial 1 hr/we	

1. Introduction:

Physical properties of fluid, Definition of type fluid: Newtonian's, non-Newtonian, Incompressible, compressible fluid, static, Dynamic fluid, shear stresses in fluid.

(4hrs)

Practic: 1 2 hr/week

2. Dimensional- Analysis:

Rayleigh's method (Power series), Buckingham's Π-theorem.

(6hrs)

3. Fluid- Dynamic:

Fluid dynamic system, Boundary layer, Continuity equation, Momentum equation, Bernoulli's equation, Euler's equation, Application of Bernoulli's equation and continuity equation.

(10hrs)

4. Newtonian's Fluid-(Incompressible Fluid):

Laminar and Turbulent flow, Reynolds's number, Friction losses and Pressure drop in pipes and Fitting, Determine of pipe diameter, Flow in Branch pipes, Friction losses in non-circular pipes, velocity Distribution.

(10hrs)

5. Non- Newtonian Fluid:

Definition, Type of fluid depend on time, Calculation of friction and pressure drop for general time independent in laminar and Turbulent flow.

(6hrs)

6. Compressible Fluid in pipes:

Pressure wave, Mach number, General equation, General equation for isothermal and adiabatic condition, work for isothermal and adiabatic of Compressor.

(10hrs)

7. Flow-Measurement:

Pilot-tube, Orifice meter, Venturi meter, Rotameters, Notch air Weis, Special Flowerets (Hot wire anemometer, Gas meter).

(12hrs)

8. Pumping of Liquids:

Calculation of Total head, NPSH, Performance characteristics curve, Calculation of horsepower, equipment, and cost.

(8hrs)

9. Flow of Fluid Through Granular Beds:

Beds & Packed Column, Fixed beds, Fluidized Bed.

(8hrs)

University of Technolog	y Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 142	Fluid Flow	Unit; 8 Theoretical 3 hr/week Tutorial 1 hr/week Practical 2 hr/week
10. Mixing of liquids Mixing, Mixen groups, Power consur	rs type, Small blade area, large	
11. Two- Phase Flow	v and Pressure drop in pipes :	(6hrs)
12. Introduction in 1		(4hrs)

(6hrs)

University of Technolog	gy Chemical Engineering Department	Second Year B Sc.	Synabus
		Units	3
CE 132	Properties of Materials	Theor tical	1 hr/week
CL 132	1100011100	Tutor al	1 hr/week
		• Pract cal	2 hr/week

1. Engineering requirement for materials:

Mechanical Properties, stress-strain curve, elasticity, plasticity, duct lity, young modulus, tensile stress, yield stress, bricking stress, true and engineering stress-strain diagram.

(4hrs)

2. Atoms and atomic coordination:

Ionic bond, inter-atomic distance attraction forces between atoms, coordination number, covalent bond, Metallic bond.

(4hrs)

3. Crystal structure:

Crystal structure, unit cell, types of unit cells simple cubic, Face centered cubic, body centered cubic, atomic packing factor, Previous lattice, Miller index, crystalline Planes, X-ray diffraction.

(4hrs)

4. Atomic disorder in solid:

impurities in solids, solid solutions in metals, Imperfection increstals, point defect, line dislocation, surfaces + grain boundaries.

(4hrs)

5. Atomic movement in solid state:

Atomic diffusion, self-diffusion, diffusivity Fick's first law, diffusion constant, diffusivities versus temperature, Factors effect diffusivity.

(4hrs)

6. Phase diagrams:

Definition of alloys, binary alloys, phase diagrams (equilibrium thermal diagrams), eutectic; solid solution and combination type diagram, Iron-carbon face diagram, allotrope Trans formation (polymorphism).

(4hrs)

7. Corrosion:

Definition, why it happen, Type of corrosion, Form of corrosion, corrosion reactions (anode + cathode) corrosion rate measurement, methods of prevention.

(4hrs)

Note:

7 2nd term

University of Technolo	gy Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 222	Computer Programming & MatLA	Unit; 6 Theo etical 2 hr/week Tuto ial - hr/week Pra tical 2 hr/week
Part -1- Visual Bas	sic	
1. Introduction to Open, Save File	Visual Basic : As Standard EXE file.	(21)
2. Menus:		(2hr)
Form window, I	Project Explorer window, Properties	s window, Ccde window,
Project window, Too	olbox window, Debug window.	(4hr)
3. <u>Using Toolbox</u> :		
Simple examples	on using toolbox items.	(4hr)
	and string data, variables, Assignme	nt statement, mathematical
and logic operations	•	(4hr)
5. Arithmetic oper	ators:	(2hr)
	nent, multiple If Then statement, I Select case statement.	
		(4hr)
7. Messages: Msgbox, InputBo	ox.	(4hr)
8. Loop: For Next Loop	p, Do while Loop, Do Loop Wh	ile, Exit For, Exit Loop. (4hr)
9. Array:		· page
Array definition, operations on Arrays	Declaring Arrays, Using for next Lo	op with Array:, Arrays and
operations on Arrays	J.	(6hr)
10. <u>Database:</u> Working with da	tabase, Adding and deleting records, I	Displaying data.

(8hr)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 222	Computer Programming & MatL	Units 6 Theoretical 2 hr/week Tutorial - hr/week Pra_tical 2 hr/week
Part -2- MatLab		
1. <u>Introduction to Ma</u> M-files, getting help		(2hr)
2. Algebra:		(2111)
	art with a letter and the intial letter	can be followed by letters (1hr)
2.2 <u>Basic Arithmetic</u> Addition +, Subtraction	en -, Multiplication *, Division /, Ex	ponentiation ^. (1hr)
 min: smallest element factorial: factorial fun round: round (up/dow floor: round down to a ceil: round up to an in exp: exponential exp(log: natural logarithm 	in an array max([132,129,66,120]) in an array min([132,129,66,120]) in an array min([132,129,66,120]) action factorial(12) in to an integer round(3.5) an integer floor(-3.1) atteger ceil(-3.1) in log(exp(2))	
- log10: logarithm to be 2.4 Complex Number		(2hr)
MATLAB can also han 2.5 <u>Loops</u>		(1hr)
3.1 Representing a f	you can use the command solve. <u>Function</u>	
	, use the command inline.	(1hr)
3.2 <u>Differentiation</u> To differentiate a funct	ion, we use the command diff	(1hr)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 222	Computer Programming	Unit; 6 Thec etical 2 hr/week Tutc ial - hr/week Practical 2 hr/week
3.3 <u>Integration</u> We can use MATLAB to command int.	for computing both definite and in	
	ations to solve differential equations. The olve. For that command, the deri	
3.5 <u>Multiple integral</u>	<u>s</u>	(1hr) (1hr)
	mpute limits, left and right limits a	s well as infinite limits. (1hr)
	ph a function is to use the commar	nd ezplot (2hr)
4.2 <u>Curves in Three-</u> The command for drawi	<u>Dimensional Space</u> ng 3D curves is ezplot3.	(2hr)

(2hr)

5. Vectors and Matrices:

5.1 Vectors

You can enter a vector by typing a list of numbers separated by either commas or spaces inside the square brackets.

- dot (dot product)
- cross (cross product)
- norm (norm or length of a vector)

(2hr)

5.2 Matrices

- A+B (matrix addition)
- A-B (matrix subtraction)
- c*A (scalar multiplication)
- A*B (matrix multiplication)
- A' (conjugate transpose of A)
- det(A) (determinant of square matrix A)
- rank(A) (rank of A)

(2hr)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus
CE 342	Fuels Technology	Units 3 Theoretical 1 hr/week Tutcrial 1 hr/weel • Practical 2 hr/weel
1. Energy and Fuel: Energy and his classification of fuel, so	assification in Technology section olid Fuels (coal), the industrial pr	on, Physical and chemica ocesses on coal to Derive
different Product.		(3hr)
Oil stocks, industri	al processes which applied from	the well to the differen
fractions in tower distil	lation and the processes on fraction	(4hr)
3. Petroleum Fraction Benzene, Kerosene,	<u>is properties</u> : Jet Fuel, Diesel, lubricants oil Asp	ohalt, Fuel oil. (9hr)
4. Gaseous Fuels and Description Energy source types in industrial and houses	gases fuel types and liquefied ga	ses which source of energ
III industrial and nouses		(4hr
which burner and the	quations about the subject and fro combustion equation and guess	om it found the componer the flue gases and stud
behavior of combustion	1.	(6hr
6. Burner of gases fue The burner types of	!: gases fuel and liquid fuel properti	es arrange it in Furnace. (2hr
7. Combustion in Fur Furnace types in oil	naces: industrial, Boilers and steam prod	luction that used in differen
industrial.		(2hr

 $\frac{\textbf{Note:}}{\square_1^{st} \text{ term}}$

University of Technology	Chemical Engineering Department	Second Year 15.Sc. Syllabus	
		Unit	4
CE 242	Thermodynamics	Theo etical Tuto ial Practical	2 hr/week 1 hr/week - hr/week

1. Introduction:

Fundamental quantities Time, Length, Mass, Force, Temperature.

(2hrs)

2. First Law and Other Basic Concepts:

Joules experiments, Internal Energy Formulation of the Thermodynamics state and state functions Enthalpy, The steady state flow processes, The reversible processes, Heat Capacity and specific Heat, Equilibrium, the phase rule.

(4hrs)

3. Applications of the Equations of Ideal Gases, Generalized Correlation and the Acentric Factor.

(8hrs)

4. Heat Effects:

Heat capacities of gases as a function of temperature, solids and iquids, Heat change accompanying phase change, Heat of industrial reactions.

(6hrs)

5. The Second Law of Thermodynamics:

Heat engine, Entropy, Second law limitation and real process Entropy change and Third law of irreversibility and probability (statistical thermodynamics), thermodynamics.

(6hrs)

6. Thermodynamics Properties of Fluids:

Relationships among the thermodynamics properties (including Helomtz and Gibes free energies and chemical potential), Steam formation and two phase system, Saturated temperature and pressure, Triple point, Wet vapor and dryness fraction, Types of thermodynamic tables and diagrams, Steam power plant cycle and analysis, Barometric condenser, Metering and throttling processes, Steam and as urbines.

(10hrs)

7. Refrigeration and Liquefaction:

Refrigeration cycles (Carnot, Air, Vapor - Compression) and Comparisons, Choice of refrigerant, Absorption Refrigeration, The heat pump, Liquefaction process.

(10hrs)

University of Technology	Chemical Engineering Department	Second Year B.Sc. Syllabus	
		Units 4	7
CE 242	Thermodynamics	Theoretical 2 hr/week	2
		Tutorial 1 hr/week	C
* <u></u>		Practic 1 - hr/weel	5

8. Phase Equilibrium:

The nature and criteria of equilibrium, Binary system, Vapor pressure of an ideal solution and non ideal solutions, Henrys law, Activity and activity coefficients, Flash separation calculations.

(8hrs)

9. Chemical Reaction Equilibrium:

Thermodynamics of ideal gases and mixtures, Derivation of the general equilibrium expression, Chemical equilibrium of ideal and non-ideal gases Reaction equilibrium in solution, effect of temperature on chemical equilibrium.

(6hrs)

	University of Technology	Chemical Engineering Department	Second Year I.Sc. Syllabus	
			Unit -	6
CE 232	Physical Chemistry	Theoretical Tutorial	2 hr/week 1 hr/week	
			Practical	2 hr/week

1. Change of State:

One component systems: Water, carbon dioxide and helium, Vapor pressure and external pressure, Liquid-vapor composition diagram, Liquid-Liquid bit ary solution mixture, Roulette's law for ideal mixtures, Non-ideal Liquid mixtures, The principle of distillation, Henry's law, solubility of gases in Liquids.

(10hrs)

2. Surface Chemistry:

Pressure difference across curved surfaces, Surface tension, and capillary rise, Langmuir adsorption isotherm and langmuir theory of adsorption, Ca culations of surface area of the adsorbent, BET equation for surface area calculation.

(12hrs)

3. Chemical Kinetics:

The rate of chemical reactions, order of reaction and rate constan, zero, first, second and third order rate equations, Reversible reactions, Consecutive reactions, parallel reactions, Determination of the order, Reactions in flow systems, Effect of temperature on reaction rate, The transition-state theory.

(16hrs)

4. Catalysis:

Reaction of catalyst in homogeneity system, Enzyme reactions and Kinetic of reactions

(6hrs)

5. Electrochemistry:

Conductivity measurements, Diffusion and ionic mobilities, Activity and ionic strength, Determination of activity coefficient from solubility, The Lebye-Hackle theory, Acid-base catalysis and their dissociation constant.

(6hrs)

6. Electrochemical Cells:

Electromotive force (EMF) of a cell, Measurements of EMF- the potentiometer, The polarity of electrodes, The cell reactions and reversible cells, Free energy and reversible cells, Types of half cells and classification EMF, Standard electrode potentials, Standard free energy and entropy of aqueous ions, Calculation of EMF of a cell, Oxidation-reduction reactions, Concentrations cells, Electrolysis, Corrosion. Kinetic of reactions.

(8hrs)

University of Technology	Chemical Engineering Department	Third Year 3.Sc.	Syllabus
		Units	4
CE 113	Democracy	Thec retical	
		Tutcrial Pra:tical	 hr/week hr/week

Standard Syllabus

University of Technology	Chemical Engineering Department	Third Year B.Sc.	Syllabus
		Unit	6
CE 133	Applied Mathematics	The retical	3 hr/week 1 hr/week
		Practical	- hr/week

1. Revision ordinary differential equations:

First and second order, simultaneous differential equations, Application for chemical engineering.

(18hrs)

2. Solution of differential equations by series:

Simple series, method of Frobeins, Bassel's equation. Application for chemical engineering.

(18hrs)

3. The Laplace transformation:

The Laplace transformer, Properties of the Laplace transformation. The inverse transformation. Inversionby partial fraction. Convolution. Solution of differential equations. The transforms of special functions. Step function, staircase function, periodic function, Trigonometric function.

(18hrs)

4. Partial Differential Equations:

Formulating of partial differential equations. Solution of partial differential equations. Method of separating variables and Laplace transformation method.

(18hrs)

5. Numerical Analysis:

The difference operators. Interpolation. Finite difference equations. Solution of first and second order differential equations.

(18hrs)

	University of Technology	Chemical Engineering Department	Third Year B. Sc. S	yllabus
			Units	7
CE 143	Mass Transfer	Theoret cal	3 hr/week	
		Tutoria	1 hr/week	
			• Practi al	2 hr/week

1. Introduction:

(4hrs)

2. Mass transfer fundamentals:

Steady state mass transfer. Diffusion in binary and multicompouent gases. Diffusion in binary and multicompouent liquids. Connective mass transfer. Mass transfer coefficients. Two-film theory. Penetration theory. Unsteady state mass transfer methods

(25hrs)

3. Methods of gas-liquid separation (absorption and stripping):

Determination of interfacial mole fractions. Individual and overall mass transfer coefficients. Effect of gas solubility on equilibrium curve. Single stage absorption, multistage absorption. Finding number of stages graphically and through use of correlations for absorption and stripping. Packed columns limits of operation. Design of packed beds for dilute and concentrated solutions. Finding number of transfer units. Absorption with chemical reaction.

(16hrs)

4. Binary distillation:

Introduction, (T-x,y) and (x,y) equilibrium curves. Flash or equilibrium distillation for binary and multicomponent mixtures. Distillation with reflux using Mckabe-Thiele. Sorel method to find number of stages. Fenske equation to find minimum reflux ratio and minimum theoretical stages. Column efficiency. Thermal method for finding number of ideal stages (Ponchoue-Savarit Method). Multicomponent differential distillation.

(25hrs)

5. Multicompouent distillation:

Dew and bubble points calculation methods. Multicomponent equilibrium relationships. Minimum reflux and minimum number of stages. Short-cut method. Checking validity of reflux ratio used. Operating limits of distillation columns.

(14hrs)

6. Mechanical separation process and size reduction:

Classification of equipment. Particle shape factor. Power requirements in crushers. Particle size analysis.

(6hrs)

Note:

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University of Technology	Chemical Engineering Department	Third Year B.Sc. Syllabus
CE 233 §	Statistics and Instrument Measureme	Unit: 4 Theoretical 2 hr/week Tuto ial - hr/week Pragtical - hr/week
Part A: Statistics		
1.Introduction: Qualitative and con of data.	clusive statistics, population sample	es, graphical representation
50 Sect 400 AND 400	N P P21	(4hrs)
2. Table and graphica	al representation: tability, normal distribution.	
Tilliary data, repeat	taomity, normal distribution.	(4hrs)
3. Media determination	on:	(4hrs)
4. Scatter determinati	ion:	(11115)
Standard and relative		(4hrs)
5. Standard distributi	ions:	(41115)
o. Daniella distributi		(8hrs)
6. Chi distribution: Confidence limit	ts, Chi test, degree of freedom.	(4hrs)
7. Curve fitting:		1
Least square m correlation coefficient.	ethod, linear relations, exponential	l equations, gradient, and
correlation coefficient.		(4hrs)
Part B: Measurem	<u>ients</u>	
	ative. Environmental conditions.	Application on physical
properties.		(4hrs)
2. Errors: Types of errors, det	termination of errors.	(8hrs)
3. Instruments:	\$000 Feet 1500	
For temperature, Pr	essure, level, composition, velocity,	flow rate. (16hrs)

(16hrs)

University of Technology	Chemical Engineering Department	Third Year 3.Sc. Syllabus
CE 243	Reactor Design	Units 6 The retical 3 hr/week Tut rial 1 hr/week Pra:tical - hr/week
reversible reactions, con	action rate constant, order of read version, yield, selectivity.	ction, reversible and non- (9hrs)
2. Types of reactors: Batch reactors, contin	uous flow reactors i.e. CSTR, PFR	. Industrial reactors. (9hrs)
3. <u>Bach reactors:</u> Design equation, reoperation, adiabatic operation	eaction time, maximum product	ion rate, non-isothermal
4. <u>Tubular reactors:</u> Design equation, length of reactor, number	isothermal and non-isothermal or of tubes, pressure drop, space ve	operation, residence time,
Design equation to the design of CSTR.	ank reactions: for isothermal operation, scale-up CSTR's in parallel and in series.	of liquid-phase batch data (15hrs)
6. <u>Unsteady state opera</u> Start up of CSTR	ation of reactors: , semi-batch reactor.	(9hrs)
7. Multiple reactions: Operation in para maximum amount of de	allel and in series for single and maired product.	
8. Comparison between Comparison between		(6hrs) for single and multiple
reactions		(6hrs)

University of Technology	Chemical Engineering Department	Third Year I .Sc.	Syllabus
CE 343		Units	7
	Heat Transfer	Theo etical	3 hr/week
	Tions I was a	Tuto ial	1 hr/week
		Practical	2 hr/week

1. Modes of heat transfer:

Material properties of importance in heat transfer.

(2hrs)

2. Steady state heat conduction in one dimension:

Plane wall, radial systems, heat source systems. Boundary surrounded by fluids. Overall heat transfer coefficient. Extended surface, conduction-convection systems, fins.

(24hrs)

3. Principles of convection:

Transport equations. Fluid mechanism aspect of convection. Laminar boundary layer. Thermal boundary layer. Empirical and practical relations for p pe and tube flow. Flow normal to single and tube banks.

(18hrs)

4. Heat exchangers:

Various types and their general characteristics. Fouling factor. Heat exchanger mean temperature differences. Co-current and counter-current flow.

(12hrs)

5. Shell and Tube Exchangers:

Types and various specifications. Design calculations by conventional and by effectiveness (NTU) methods. Optimum design calculation.

(6hrs)

6. Condensation and Boiling Heat Transfer:

Condensation of single vapors. Design calculations for condenser, condensersubcooler and superheater condenser.

(9hrs)

7. Radiation and furnace design:

Radiation properties, shape factor, heat exchange for non black bc dies, parallel planes, shields, gas tradition, boiler.

(9hrs)

8. Unsteady state heat transfer:

Temperature as a function of time, lumped capacity system, quenching of small bodies, heating of tank reactor.

(10hrs)

Note:

2 nd term

University of Technology	Chemical Engineering Department	Third Year B.Sc. S	yllabus
CE 443	Equipment Design using Computers	4 444 7 7 444	6 2 hr/week - hr/week 1 hr/week

Part (A)

1. Process Planning:

Scheduling and flow sheet design. Flow sheet types and designation. Block diagram. Process flow sheet. Piping and instruments diagram. Utility flow sheet. Equipment layout and plot plan. Project evaluation. Feasibility studies. Site selection. Project cost estimation.

(6hrs)

2. Piping networks and pumps:

Pipe fittings, valves. Steam traps. Piping design standards. Piping material and selection. Pump specifications and data sheets. Optimum diameter. Types of pumps. Hydraulic characteristics for centrifugal pumps.

(6hrs)

3. Vessels and Tanks:

Types of vessels. Flash drums. LPG tanks. Criteria in vessel design. Stress considerations. Design of tall vertical vessel. Design of pressure vessels. Supports and foundations

(6hrs)

4. Heat transfer equipment:

Types of exchangers and applications. Exchanger ratings. Exchanger design. Exchanger specification sheets. Furnaces. Convection and radiation zones. Types of fuels. Burners and arrangements. Steam boilers, types of boilers.

(6hrs)

5. Mass Transfer Equipment:

Types of columns. Plate and packed Types of plates and packing. Design features. Pressure drops in columns.

(6hrs)

University of Technology	Chemical Engineering Department	Third Year B.Sc.	Syllabus
		Unit:	6
CE 443	Equipment Design using Computers	Theoretical Tuto ial Practical	2 hr/week - hr/week 1 hr/week

Part (B): Complete equipment design of each of the following units:

1. Pressure vessels, pumps and compressors:

Flash drum. Gas-liquid separator, LPG storage tanks. Cyclone separators. Gas movers and compressors. Pumps. Piping network. etc.

(8hrs)

2. Heat Equipment:

Shell and tube heat exchangers. Plate heat exchanger. Boilers. Furnaces. Vaporizers. Air coolers. Coil type heaters. Condensers. etc.

(10hrs)

3. Mass transfer equipment and reactors:

Distillation column. Absorber. Extractor. Dryer. Absorption towers. Leaching equipment. Scrubbers. Stripping towers. Reactors.

(12hrs)

	Chemical Engineering Department	Third Year 3.Sc. S	Syllabus
University of Technology	Chemical Engineering Department	Units	2
CE 543-U	Biochemical Engineering	Theo etical :	2 hr/week 1 hr/week
	Unit Operations Branch	Pra :tical	- hr/week
1. Definition, introduct	ion:		(2hrs)
2. <u>Industrial micro org</u>	anisms and methods of separatio	<u>n:</u>	(2hrs)
3. Micro organisms gei	netics:		(4hrs)
4. Biotechnological me	thods of growths: atch and continuous cultures.	Breaded cultures	s Solic
Fermentation's			(6hrs)
5. Single cells proteins Microorganisms use	: d, treatment of waste, production	of single cell prote	ein. Die
value			(3hrs
6. <u>Bakery yeast:</u> Dry and ready ef	fective types.		(3hrs
7. Amino acids: Clotamic acid.			(1hr
8. Organic acids: Acetic acid.			(1hr
9. Enzymes:			(1hi
10. <u>Antibiotics:</u> Microorganisms pro	oducing antibiotics, Penicillin		(1h
11. <u>Industrial Alcoho</u> Alcoholic fermenta	l: tion, microorganism used, product	ion method.	(1h
12. Animal cell cultiva	ation:		(2hr
13. Plant cell Cultiva	tion:		(1h
14. Environmental cl In the fields of v	eaning using biotechnology: water treatment. Organic Pollutants	s. Petroleum v/astes	(2hı

University of Technology	ity of Technology Chemical Engineering Department	Third Year 3.Sc. Syllabus	
		Unit:	2
CE 643-U	Environmental Engineering	Theo etical Tutc rial	2 hr/week 1 hr/week
	Unit Operations Branch	Prac tical	- hr/week

1. Introduction:

What is pollution and a historical brief on international concern w th pollution. Types of pollutants and measurements. Pollution allowable limits in the environment. (2hrs)

2. Cleaning of air and industrial gases:

Sources and effects of air pollution. Methods of control. Design ng pollution equipment. Settling chamber, Cyclone separator, Venture scrubber, spray tower, cyclone scrubber, packed bed, electrostatic precipitator. Economic evaluation of gas cleaning.

(10hrs)

3. Water and heavy water treatment:

Physical methods of treatment. Occupation methods. River water cleaning, heavy water treatment. Desalination.

(10hrs)

4. Industrial and sewage treatment:

Industrial water treatment. Sewage water treatment.

(4hrs)

5. Solid pollutants:

Sources and effects. Methods of control of solid pollutants.

University	of Technology	Chemical Engineering Department	Third Y	ear B.Sc.	Syllabus
				Uni s	2
CE 543-R	Storage and T	ransport of Crude and Petroleum P	roducts	Theoretical Tutorial	2 hr/week 1 hr/week

Refinery Engineering Branch

Pra:tical - hr/week

neering Branch Pra :tical - hr/w

1. Fluid Flow:

Petroleum liquids flow. Petroleum gases flow. Designing pipe r etwork from optimum velocity and diameter viewpoint. Crude and petroleum products pipelines, Heating and pumping stations, Increasing pumping rates by looping. Increasing pumping rates by polymer additives. Two-phase flow. Unsteady state flow and hammering.

(14hrs)

2. Pressure Vessels / Tanks:

Stress systems of shells. Bending and membrane shells. Equilibrium equation for a shell element and LPG tanks, Rotationally symmetrically loaded membrane shell. Crude oil huge tanks.

(8hrs)

3. Selection of Materials for Vessels and Pipelines:

Selection of materials for vessels to limit corrosion. Selection of materials for pipelines to limit corrosion. Catholic protection.

(8hrs)

University of Technology	Chemical Engineering Department	Third Year J.Sc. Syllabus
CE 643-R	Combustion Engineering	Units 2 Theoretical 2 hr/week Tutorial 1 hr/week Pract cal - hr/week
	Refinery Engineering Branch	Pract cai - III/week
1. Combustion reaction	stoichiometry:	(8hrs)
2. Excess air and flame		
3. Combustion product		
 Gas, solid, and liquid Flammability limits: 	Tueis:	
		(8hrs)
6. Flame and combustic	on speed:	
7. Flame stability:		
8. Gas, liquid and solid	fuels burners:	(6hrs)
9. Combustion zones an	nd temperature profiles:	(01113)
	ection rooms in furnaces:	
11. Furnaces and the tu		
12. Chimney height cale	culation:	7724 V
		(8hrs)
	and heat loss calculations:	
14. Furnace wall layers		
15. Tube layers in furna	aces:	

University of Technology

Chemical Engineering Department

Fourth Year B.Sc. Syllabus

Units

3

CE 244

Project

1. Introduction:

Literature, Methods of manufacturing, Capacity of the Process Physical and Chemical Properties.

2. Material and Energy Balances:

Review of Material and Energy balances, Flow sheet Symbols, PFD In formation in Flow Diagram.

3. Choice of Process Devices and Equipments:

Pumps, Valves, Pipes, Measuring Devices and Controllers, Material of Construction, PID Plot.

4. Choice of Plant Location and Layout Standards:

5. Industrial Services:

Cooling Tower, Boilers, Study Problems from Industrial Waste Disposals, and How to Prevent the Environment, pollution.

6. Use a Computer Program to Design:

Introducing a Complete Project with Material and Energy Balances Desi 3n of Devices, Calculation of Costs, Choice of Location.

University of Technology Chemical Engineering Department Fourth Year B.S.c. Syllabus

CE 344

Transport Phenomena

Units 7

Theoretical 3 hr/week
Tutorial 1 hr/week
• Practic: 1 3 hr/week

1. Boundary Layer and analogies:

Boundary Layer in Laminar and turbulent Flow, Boundary Layer in Laminar sub Layer, Transition layer, Velocity distribution on Surfaces and Pipes, Momentum, heat and mass transfer molecular diffusion, Eddy transfer, Reynolds analogy, Modified Reynolds analogy, Chilton and Colburn analogy.

(20 hrs)

2. Evaporation:

Introduction, Types of Evaporations, Evaporation Equipment, Heat transfer in Evaporation Process, single, double and Multi effect Evaporators, Design of evaporators, Comparison of Forward, backward and Parallel effect evaporators, boiling Point rise.

(12 hrs)

3. Drying:

Introduction and general Principle in drying, Rate of drying, the mechanism of moisture movement, Calculation of rate of drying, moisture transport in Solids at Constant and falling rate Period, Capillary movement, Material and Energy Balances in Continuous dryers, Types of Dryers

(8 hrs)

4. Humidification, dehumidification and Cooling towers:

Humidification, dehumidification, Temperature humidity Chart for a r - water system. Enthalpy - humidity -temperature chart. Addition of Vapor or liquid Stream to a gas Stream. Mechanism of dehumidification Evaluation of heat and Mass transfer Coefficient, Cooling tower, height of Packing in Cooling towers, Minimum gas Condition, Change of Condition of liquid and gas in Cooling towers.

(14 hrs)

5. Extraction:

Definition, Extraction process, Equilateral Triangular coordinates (Ternary Diagram), system of three liquid _ one pair partially soluble, system of three liquid _ two, pairs partially soluble, choice of solvent, Equipment in extraction cross _ current extraction, multi stage Cross Current extracting cross current for insoluble Liquid, Continuous Counter current extraction, Continuous Counter Current in Soluble, Liquid, Minimum Solvent, Counter Current extraction with reflux.

(22 hrs)

Note:

□ ist term

University of Technology	Chemical Engineering Department	Fourth Year B. Sc. Syllabus
	•	

CE 344 <u>Transport Phenomena</u>

Units

Theoretical 3 hr/week
Tutoria 1 hr/week

Practical 3 hr/week

6. Filtration:

Type of Filters, Filtration theory, Plate and frame filter press, leat filter, filtration at Constant ΔP , Filtration at Constant rate, washing Time.

(10 hrs)

7. Crystallization:

Introduction Equilibrium Solubility in crystallization, Classification of Crystallizes, Nucleation theory, Rate of Crystal Growth, Material and Energy balance in Crystallization.

CE 444

Process Control

Units

Theoretical 2 hr/week
Tutorial 1 hr/week

5

• Practice 1 3 hr/week

1. Revision of Laplace Transformation:

(4 hrs)

2. Process Dynamic and Transient Response of the system:

Dynamic and Control For chemical Process, Dynamic behavior for 1st order system, lineraization, Interacting and non-interacting, Second order system, Transportation.

(16 hrs)

3. Characteristics of the closed loop System:

Over all closed loop transfer functions and block diagram algebra Trunsient of Simple Closed loops System, Stability of Control, Introduction to the l'requency analysis and design technique.

(20 hrs)

4. Industrial Controller:

Actions selection Criteria for Various Control modes, final Control element, Dynamics and Control of Chemical reactor System.

(10 hrs)

5. Dynamic and Control of some Complex Chemical Process:

Heat exchanger, distillation Column, Introduction to Computer Control of Chemical Process.

(10 hrs)

Note:

2nd term

Terramental Sections	University of Technology	Chemical Engineering Department	Fourth Year B.	c. Syllabus
	CE 544-U	Chemical Industries Unit Operation Branch	Tutoria	6 cal 2 hr/week 1 hr/week al 3 hr/week
	Chemical Processing Introduction, Operation			(2 hrs)
	2. Sulpher and Sulphe Raw materials, Frasc Production.	ric acid: ch Process, Iead Chamber Process,	, Contact Proces	(3 hrs)
	3. Ammonia and Nitrie	<u>c acid:</u> action Steps, Production Process.		(4 hrs)
	4. Nitrogenous Fertiliz	* 8		(6 hrs)
	5. Phosphoric acid:	oric and, Electric Furnace		(4 hrs)
	6. Phosphate Fertilizer Superphosphate, Trip	·s:		(4 hrs)
	7. Electrolytic Industri		Production	(4 hrs)
	8. <u>Industrial Salts:</u> Sodash Production			(4 hrs)
	9. <u>Ceramic Industries:</u> Raw materials, Chem	ical Conversions, Porcelain, Fire Cl	ay	(3 hrs)
	10. <u>Cement Industries:</u> Raw materials, Portla		n	(4 hrs)
	,			(41)

Raw materials, Manufacture of Glass

11. Glass Industries:

University of Technology	Chemical Engineering Department	Fourth Year B.5 c.	Syllabus
		Units	6
CE 544-U	Chemical Industries	Theoreti al	2 hr/week
		Tutorial	1 hr/week
	Unit Operation Branch	Practica 1	3 hr/week
Paint, varnishes, lacq 13.Oils and Fats: Fatty saids, Manufact	86		(4 hrs)
St. Section 2	ture of Vegetable Oils and Waxes		(4 hrs)
14. Soap and Detergents Soap Production, Det			241
15. Sugar Industries: Cane Sugar, Beet Sug	gar		(4 hrs)

University of Technology Chemical Engineering Department Fourth Year B.Sc. Syllab	<u>ous</u>
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CE 134

Optimization

Units 2

Theoretical 2 hr/week
Tutorial 1 hr/week
Practical - hr/week

1. Introduction to Optimization Methods:

(4 hrs)

2. Organization of Optimization problems:

(4 hrs)

3. Single Variable:

Analytical methods, numerical methods, graphical methods, numerical Search, restriction Function, unrestricted Function, Direct search, Dichotomous Search, golden Search, Fibonacci Search

(12 hrs)

4. Multivariables Optimization Methods:

- Necessary Conditions For Extreme Values in graphical Cases.
- Solution by graphical method.
- Simplex method.
- Linear Programming and application in Chemical er gineering (transportation mixing)

(10 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.S :	Syllabus
		Units	2
CE 644-U	Engineering of Catalysts	Theoretic al Tutorial	2 hr/week 1 hr/week
	Unit Operations Branch	Practical	- hr/week
1. Catalysts:			
Definitions, Propertie	es of Catalysts		
, 1			(2 hrs)
2. Steps in a Catalytic			
Adsorption, Surface	Reaction, Decoration, Rate Equation	ons for each Step	(4 1)
	136 1		(4 hrs)
3. Finding Rate Law a			
Equations of Rate Li	mitting Step.		(4 hrs)
4. Design of Reactor fo	or Gas - Solid Reactions:		
Basic Guidelines. De			
			(8 hrs)
	Analysis for Reactor Design:		
Deducing a Rate l	aw from the Experimental Data	a, Finding a Me	chanism
Consistent with Experir	nental Observations, Evaluation o	f the Rate law Para	ameters,

(12 hrs)

Reactor Design

55 Fourth Year B.Sc. Syllabus

University of Technology	Chemical Engineering Department	Fourth Year 3.Sc.	Syllabus
	The serious of Catalysts	Uni s Thec retical	2 2 hr/week
CE 644-R	Engineering of Catalysts Refinery Engineering Branch	Tuterial Praetical	1 hr/week - hr/week

1. Introduction:

Properties of Catalysts, Determination of Surface Area, Void Volume and Solid Density, Pore Volume Distribution, Classification of Catalysts, Promoters and Inhibitors, Catalyst Poisoning.

(4 hrs)

2. Rate Equations of Fluid-Solid Catalytic Reactions:

Rate of Adsorption, Desorption and Surface Reaction, Rate of the limiting Step, Mechanism, Design of Reactor for- Gas- Solid Reactions. Data Analysis for Reactor Design

(10 hrs)

3. External Transport Processes in Heterogeneous Reactions:

Equations of Mass -and Heat Transfer, Calculations of the Concen rations on the Surface, Calculations of Temperature Differences in External Transport Processes, Fluidized Bed Reactors, slurry Reactors (4 hrs)

4. Internal Transport Processes:

Gaseous Diffusion in Single Cylindrical Pores, Surface Diffusion Effectiveness Factor, Thiele - Type Modulus (4 hrs)

5. Design of Heterogeneous Calalytic Reactors:

Fixed - Bed Reactors, Fluidized Bed Reactor, Slurry Reactions

(8 hrs)

University of Technology	Chemical Engineering Department	Fourth Year 3.Sc. Syllabus
CE 744-U	Corrosion Engineering	Unit 2 Theo etical 2 hr/week Tuto ial 1 hr/week
	Unit Operations Branch	Practical - hr/week
	ion, why it happens, direct and ind	irect disadvantages. (2 hrs)
2. Types of Corrosion:	: Corrosion, Eight form of Corrosion	i
Dry Corrosion, wet C	Lorrosion, Light form of Corrosion	(4 hrs)
3. Mechanism of Crev		(2 hrs)
4. Electrochemical cel Electrode Potential, t	<u>l:</u> types of reference electrodes.	(4 hrs)
5. Corrosion rate Cald Wet loss method, ele	ectrochemical method	(2 hrs)
6. Polarization:		
Definition, Types.		(4 hrs)
7. Passivity:		(2 hrs)
8. Effect of Temperatu	ure and velocity on corrosion rea	
9. Method of Protection	on:	
		(4 hrs)
10. Instruments to mea	sure Corrosion rate:	(A hrs)

University of Technology	Chemical Engineering Department	Fourth Year 3.Sc. Syllabus
		Unit 2
CE 844-U	Technology of Oil and Gas	Theo etical 2 hr/week
		Tuto ial 1 hr/week
	Unit Operations Branch	Prac tical - hr/week
1 History and Davelo	pment of Refinery Processes:	
Kinds of Refineries.	pment of Refinery 110cesses.	
Killes of Refilleries.		(2 hrs)
2 Chamical Commonit	ion of Detucloum.	(2 1113)
2. Chemical Composit	ion of Petroleum:	(2 hrs)
		(2 hrs)
3. Physical and Therm		
Physical and Thermo	dynamic Properties of Petroleum of	
		(2 hrs)
4. Introduction to Pro	cessing:	
Stabilization, dehydr	ation, Tube still Heaters.	
		(4 hrs)
5. Fractionation of Per	troleum:	
Atmospheric and Va		
1		(2 hrs)
6. Fractionation Towe	rs:	
	Balances, Reflux, Temperature D	istribution in Fractionation
Towers, Tower Diamete		
Towers, Tower Diamete	1	(6 hrs)
7 71		(0 1113)
7. Thermal Cracking:		
Catalytic Cracking		((1)
192 - Paris Command L orrosia		(6 hrs)
8. Natural Gas:		
Refinery Gases		
		(4 hrs)
	27 E 27 Q 27 T	

(2 hrs)

9. Natural Gas Processing Units:

CE 234 Industrial Management and Safety Requirements

Units 2

Theoret cal 2 hr/week
Tutori: 1 1 hr/week
Practi: al - hr/week

1. Management:

Principle of management, types and classifications, management responsibility, organization responsibility.

(4 hrs)

2. Industrial organization

Site, Feasibility study, Development of efficient work method (plant layout, flow of material, material handling), workstations, Inputs and Outputs, Production planning (types of productions).

(3 hrs)

3. Maintenance:

Classifications, cost, machine replacements, case studies and examples

(3 hrs)

4. Network Analysis:

Principles and applications, Critical Path Method (CPM), Gantt Chart, Pert techniques (examples and case studies).

(3 hrs)

5. Work Measurement Techniques

Time and Motion study.

(3 hrs)

6. Costing:

Framework of management, cost of production (row material cost, labor cost, machinery cost).

(3 hrs)

7. Quality Control:

Standardization, specification, sampling techniques, Inspections – analysis of results. Quality costs (preventive cost, appraisal cost and failure cost). Application of quality control chart – examples, Reliability.

(4 hrs)

8. <u>ISO:</u>

Requirements, applications, ISO series, Quality management system (OMS), Total Quality managements (TQM), Requirements and applications.

(3 hrs)

9. Safety Requirements:

Hazards (types e.g. industrial hazards, pollution (air pollution, water pollution, industrial pollution). Industrial by products and industrial waste, Safety requirements of industrial sites. Requirements of suitable work environment (examples with particular emphasis in chemical industry).

University of Technology	Chemical Engineering Department	Fourth Year E.Sc. Syllabus
CE 944-U	Polymer Technology	(4 hrs) Units 2 Theoretical 2 hr/week Tutor al 1 hr/week
	Unit Operations Branch	Tutor al 1 hr/week Pract cal - hr/week
1. Polymers: Nomenclature		
	1	(2 hrs)
2. Polymers Classificate Classification based Polymers, Technical Cla	on Sources, Classification base	d on Chemica nature of
	9-39-30-30-30-30-30-30-30-30-30-30-30-30-30-	(2 hrs)
Polymerization, Ionic (c	merization (condensation polyr ationic, Anionic) Coordination Po rization by group transfer	
•	Story transfer	(6 hrs)
4. Polymerzation Syste Bulk, Solution, Suspe		(4 hrs)
5. Molecular Weight of	f Polymers:	(4 1115)
Methods of measuren	nent of molecular weight of Polym	
6. Polymer composites:		(2 hrs)
o. rolymer composites:	<u>.</u>	(2 hrs)
7. Polymer blends:		
O Dhysical Duonoutics	of Dalum	(2 hrs)
8. Physical Properties	of Polymers:	(4 hrs)
9. Mechanical Properti	ies of Polymers:	()
	act strength, Creep, hardness	(4 hrs)
10. Polymer Processing:	And the state of t	
iviolaing, Extrusion, 1	Thermal Forming Calendaring	(4 lans)

University of Technology	Chemical Engineering Department	Fourth Year	3.Sc. Syllabus

6 Unit: Theo etical 2 hr/week Petroleum Refining CE 544-R 1 hr/week Tute ial Practical 3 hr/week Refinery Engineering Branch 1. Introduction: History and Development of Refining Processes Kinds of Refineries. (2 hrs) 2. Chemical Composition of Petroleum: (2 hrs) 3. Physical and Thermodynamic Properties of Petroleum oil: (2 hrs) 4. Evaluation of oil stocks: (4 hrs) 5. Introduction to Processing: Stabilization, dehydration, tube still heaters (10 hrs) 6. Fractionation of Petroleum: Atmospheric and Vacuum Fractionation (2 hrs) 7. Fractionation Towers: Material and Energy Balances, Reflux Temperate Distribution in l'ractionation Tower, Tower Diameter (8 hrs) 8. Treating Processes: Removal of Acid Gases, Sweetening Processes, improvement in Performance and storage stability. (8 hrs) 9. Upgrading processes: Thermal Crackling, Cooking, Visbreaking, Catalytic, Cracking, Hydrocracking, Catalytic Reforming, Alkylation, Isomerisation. (10 hrs) 10.Product Blending: Blending of vapor pressure, octane number, viscosity, flash point aniline point, and pour point. (6 hrs)

11.Manufacturing of lubricating oils:

(4 hrs)

12. Treating process using Hydrogen:

(2 hrs)

University of Technology	Chemical Engineering Department	Fourth Year B.5 c. Syllabus
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CE 844-R	Tuto	ts 2 pretical 2 hr/week prial 1 hr/week prical - hr/week
S	Refinery Engineering Branch Flat	tica - ili/week
1. Classification o	f Gaseous fuels thermodynamic:	(2.1)
2 Physical and Cl	hamical Proporties for natural Case	(3 hrs)
2. Physical and Ci	hemical Properties for natural Gas:	(3 hrs)
3. Drying of Gase	ous Fuels:	(5 1113)
	rption, Drying by Absorption	
, , ,	1 7 7 0 7 1	(4 hrs)
4. Sweetening of C	Gaseous Fuels:	
Sweetening by A	Adsorption, Sweetening by Absorption	(61
5 C B	•	(6 hrs)
5. Gas Processing		1170
riash Separation	n Processes, Distillation of multicomponemt gas mixt	(6 hrs)
6. Liquefied Petro	oleum Gases:	(0 1113)
o. Elquelled Tello	Jean Gusesi	(2 hrs)
7. Production of H	Hydrogen:	
A 14 COLD 1000		(2 hrs)
8. Sulfur recovery	processes from Acid gases streams:	
		(4 hrs)

University of Technology	Chemical Engineering Department	Fourth Year E.Sc. Syllabus
	D. J.	Units 2 Theor tical 2 hr/week
CE 744-R	Petrochemical Industries	Tutor al 1 hr/week
	Refinery Engineering Branch	Pract cal - hr/week
Processes, Purification System, Diolefins, So Production, Conversion Production methods of	Materials: Cracking Processes Cooling, Puri Processes Compression, Separ ources of diolefins, Butadiene Process For Production of higher olefins, linear, Alkyl ben	Separation, Isobutylene diolefins, Higher Olefins zene Complex Aromatics,
Basic materials Aroma	tic Sources, Separation of benz	ene – Toluene – Xylene,
Syntheses Gas, Steam re	forming, Partial Oxidation	(10 hrs)
3. Synthetic Zeolites:		(2 hrs)
Ethyleneglycol Acrylon Styrene, Phenol, Nitrob	chemicals: e Oxide Acetic Acid, Etha itrile, Adipic Acid, Methyl Tetra benzene, Ethylene, Cyclohexane,	butyl ether, EthylbenZene,
acid.		(7 hrs)

5. Petrochemical Detergents:

(2 hrs)

6. Polymers: LDPE, HDPE, PP, PVC

(4 hrs)

7. Synthetic Fibers:

(2 hrs)

8. Petrochemical Complexes: (2 hrs)

University of Technology	Chemical Engineering Department	Fourth Year I .Sc. Syllabus
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CE 944-R

Environmental - Pollution

Units 2

Theor:tical 2 hr/week
Tutorial 1 hr/week

Refinery Engineering Branch

Prac ical - hr/week

1. Introduction:

Definition of Pollution, literature surveys on international interest of Environmental Pollution Problems. Pollutants, Type of Pollutants, Measuring Devices, Allowable level of Pollutants in Environment.

(2 hrs)

2. Air Pollution:

Air Pollutants, Particulate matter, SO₂, CO, NO, ...etc Sources and effect, Control Treatment Methods, Design of some Equipment: Centrifugal, Cyclon, Scrubber, Packed Bed Ventura, Spray – tower, Electrostatic Separator, and Chimne / Design.

(12 hrs)

3. Water - Pollution:

Water Pollutants, Sources and effect, Organic and non – organic, Radiation activity, Suspended Solids and Sculling Salts, Control Treatment of Warer Pollution for: Drinking, Industry Processes, Desalination methods, Reverse Osmosis, Electrodalysis, Solar Distillation, Primary, secondary and advanced treatment of Wastewater, Treatment of oily Pollution.

(12 hrs)

4. Pollution of Solid - Waste:

Sources, Effects, Control Treatment method of Removal Solid – Waste, Treatment by economy methods.

(6 hrs)