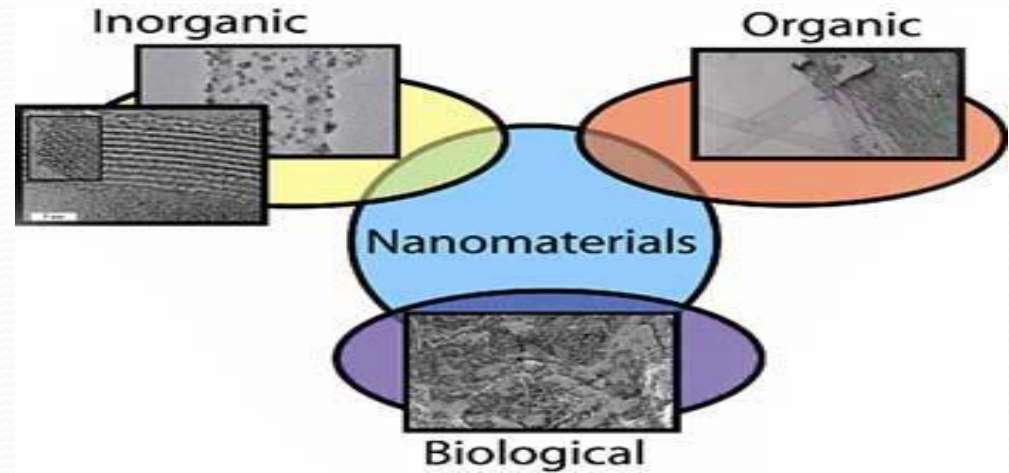


Lecture 4

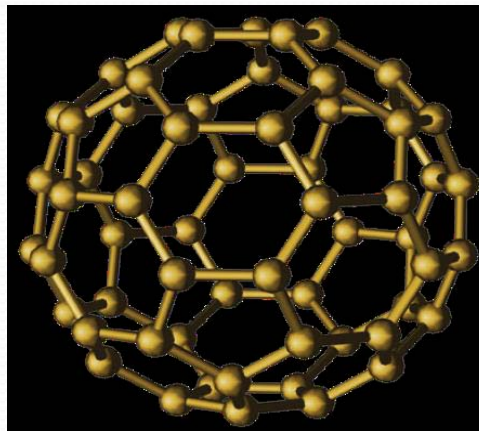
Nanomaterials



Nano scale of nanomaterials can be in following:

- *Zero dimension*

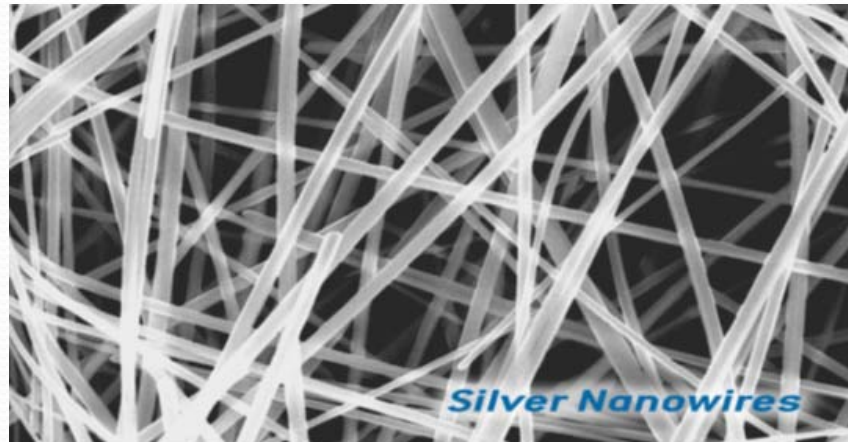
Fullerene: A *fullerene* is a molecule of carbon in the form of a hollow sphere



Optical Properties

- **One dimension**

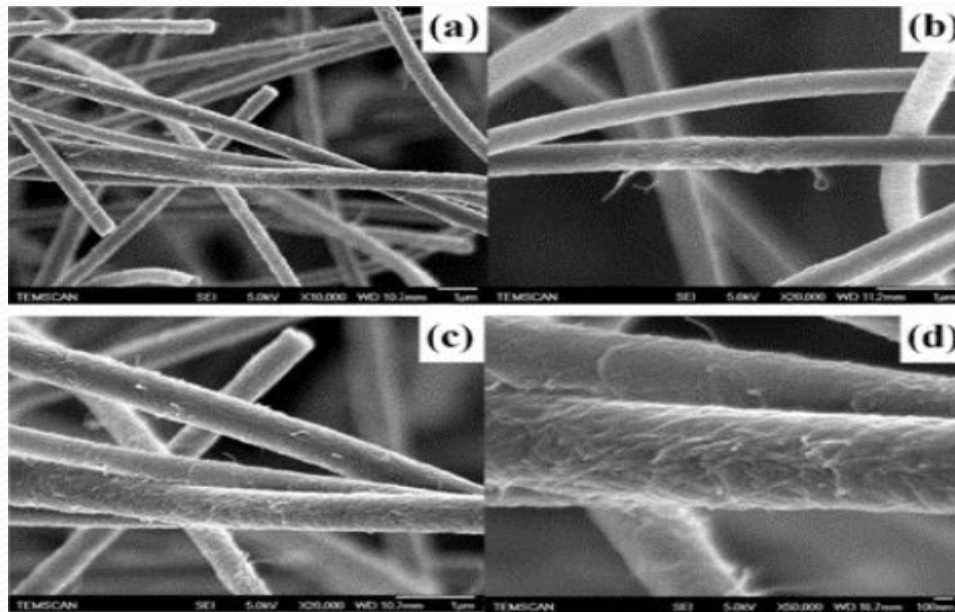
Nano Wires One of the most fascinating and useful aspects of nanomaterials is their optical properties.



Electrical Properties

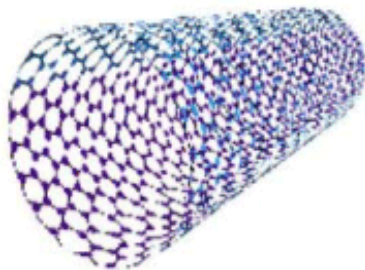
- **Tow dimension**

Fibers

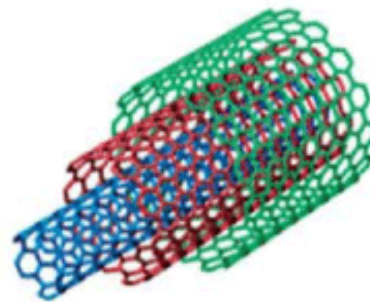


Mechanical properties

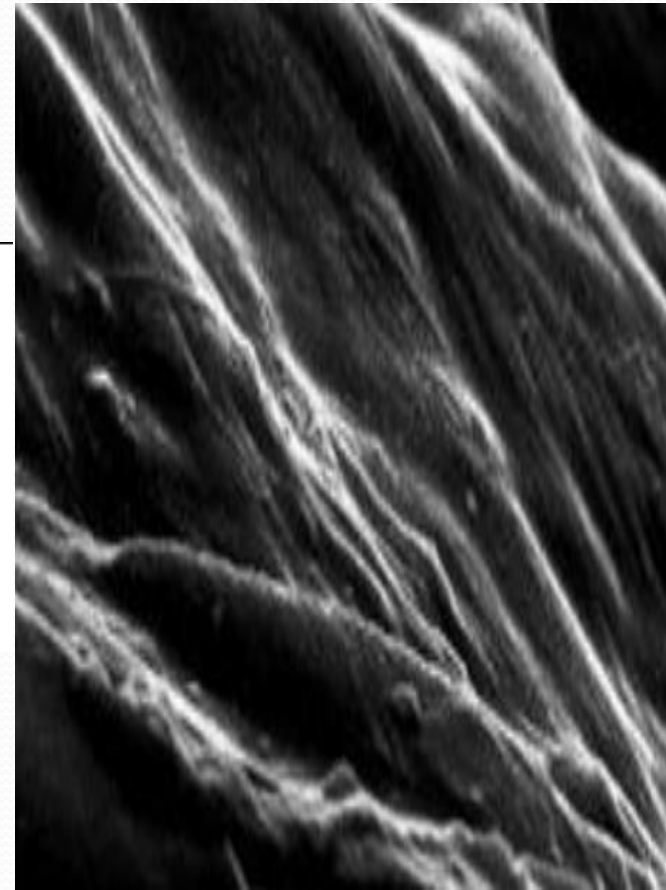
- *Carbon Nanotubes* (CNTs)



Singlewalled CNT



Multiwalled CNT



Mechanical properties

- **Three dimension**

Particles



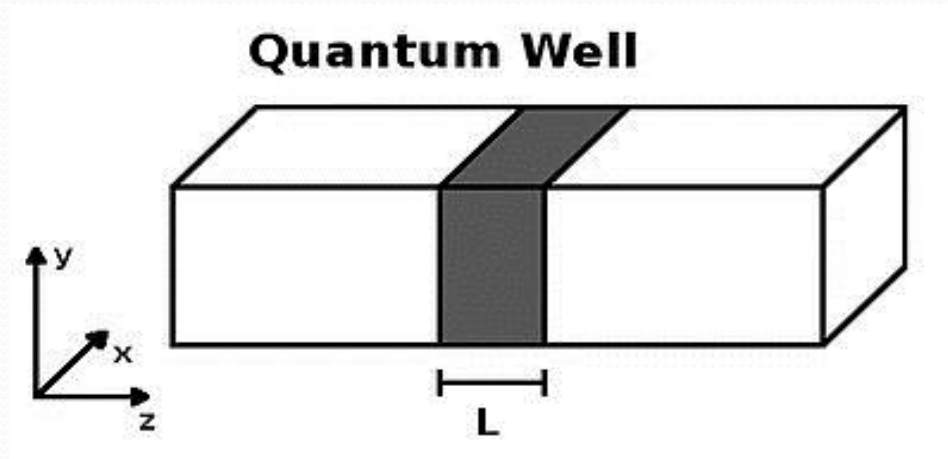
A nanomaterial can exist in single, fused, aggregated or agglomerated form with spherical, tubular and irregular shapes.

Why nanomaterials have different properties?

Why nanomaterials have superior chemical reactivity?

Crystal structure

- Quantum well احدي الابعاد بالنانو

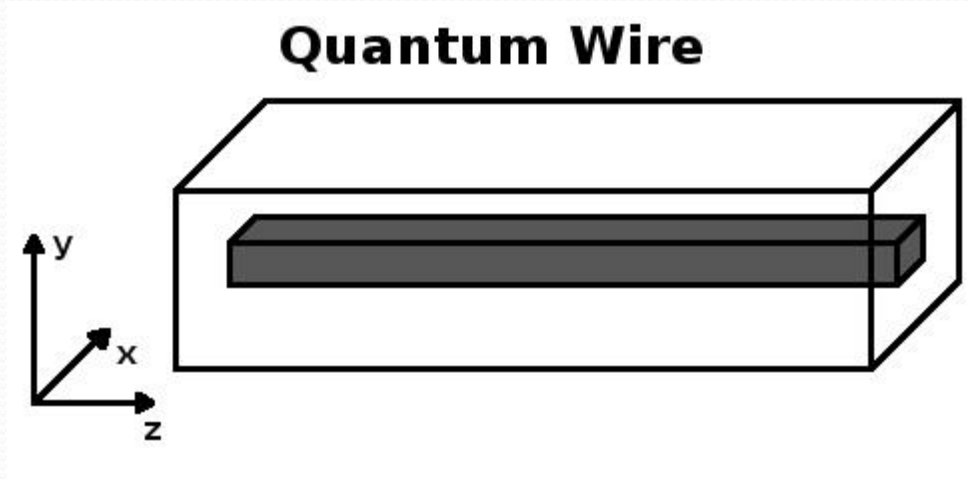


Bravais lattice

- **Quantum wire**

Is an electrically conducting Quantum wire in which quantum effects influence the transport properties.

فقط اعطاء المثال على ذلك ذات البعدين بالنانو

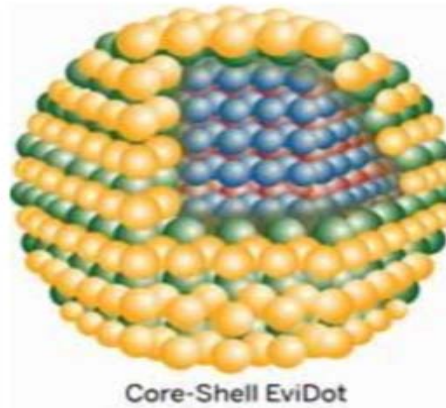


Atomic Packing Factor (APF)

- **Quantum dots (QD)** are very small semiconductor particles, only several nanometers in size, so small that their optical and electronic properties.

ثلاثة ابعاد

What is a quantum dot?



- Nanocrystals
- 2-10 nm diameter
- semiconductors

APF FOR SIMPLE CUBIC

Why the nanomaterials are more interest?

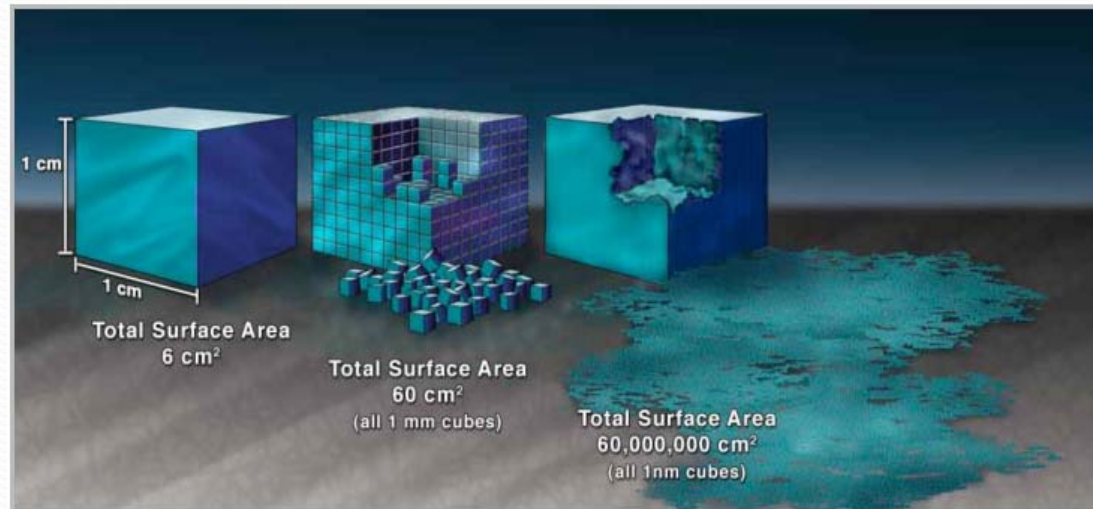


Illustration demonstrating the effect of the increased surface area provided by nanostructured materials

APF FOR BCC / BODY CENTERED CUBIC

Scale at Nanoscale materials have far larger surface areas than similar masses of larger-scale materials. As surface area per mass of a material increases, a greater amount of the material can come into contact with surrounding materials, thus affecting reactivity.

